

Appendix 1

Welcome and introduction

F. Dupouiron (Total)



REFINING & CHEMICALS

October 2015

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CONTENTS

- 1 OVERVIEW
- 2 MARKETS
- 3 STRATEGY
- 4 REFINING & PETROCHEMICALS OPERATIONS
- 5 SPECIALTY CHEMICALS
- 6 ADDITIONAL INFORMATION

1

OVERVIEW



M. Guillaume Perrin

Refining & Chemicals – October 2015

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POSITION WITHIN TOTAL



OUR ACTIVITIES

- Refining and petrochemicals, key activities between upstream and marketing.
- Specialty chemicals affiliates that hold leadership positions in their segments.



UPSTREAM

- 1 OIL AND/OR GAS PRODUCTION
- 2 LNG CARRIER
- 3 OIL TANKER
- 4 LIQUEFACTION/REGASIFICATION

REFINING & CHEMICALS

- 5 REFINERY
- 6 BLENDING PLANT
- 7 SPECIALTY CHEMICALS
- 8 TRADING OFFICE

MARKETING & SERVICES

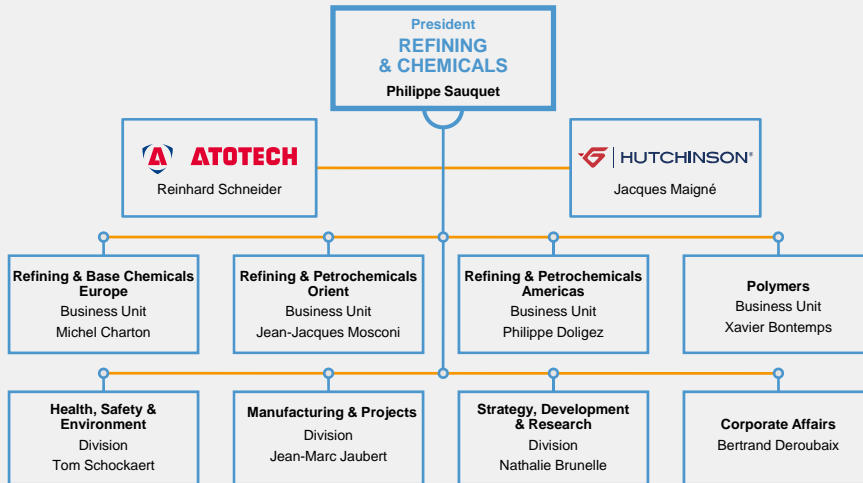
- 9 SERVICE STATION
- 10 LUBRICANTS PLANT
- 11 OIL DEPOT
- 12 SOLAR PANELS

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REFINING & CHEMICALS ORGANIZATION CHART



WORKFORCE



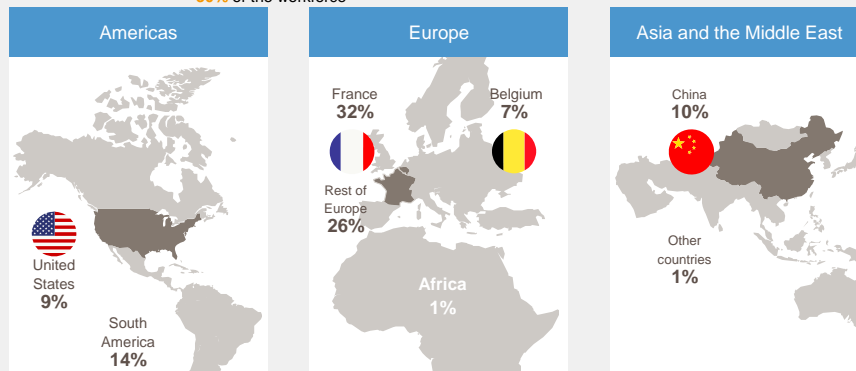
45,000
employees
at January 1, 2015



Refining & Petrochemicals:
30% of the workforce



Specialty Chemicals:
70% of the workforce



KEY FIGURES



Refining

14 refineries worldwide* including **5** in France

No. 1 refiner in Europe
12% of Europe's refining capacity,
or **1.7 million b/d**

No. 10 refiner worldwide
2% of global refining capacity,
or **2.2 million b/d**

* Marketing & Services (M&S) operates seven other refineries.



Petrochemicals

26 petrochemical facilities

No. 2 in petrochemicals in Europe
No. 10 in petrochemicals worldwide
> 20 million tons per year



Specialty Chemicals

Revenue
\$5.9 bn

HUTCHINSON®
World leader in rubber products
96 sites, **28,600** employees

ATOTECH

No. 1 worldwide in metal and surface finishing
17 sites, **4,000** employees

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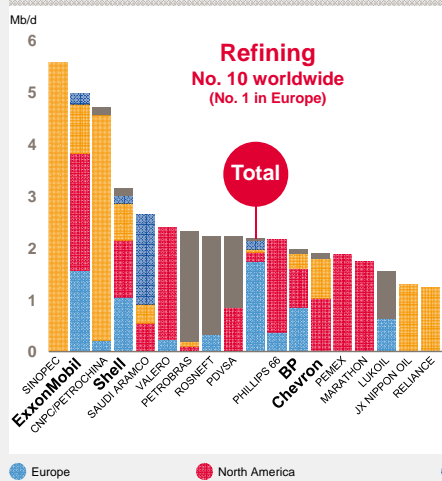
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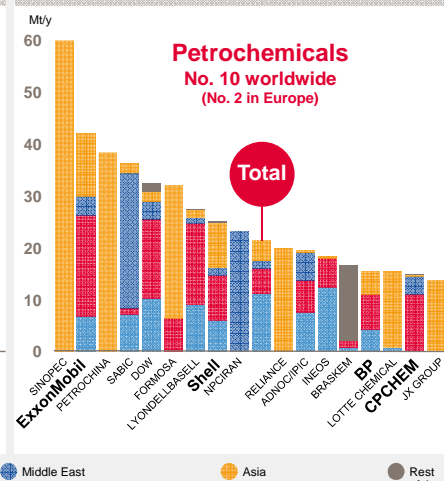
A WORLD-CLASS REFINING & PETROCHEMICALS PLAYER



Refining capacity



Petrochemical capacity



Capacity as at 12/31/2014

Portfolio: Ethylene, Propylene, Benzene, Butadiene, Toluene, Xylene, Paraxylene, Styrene, Polyethylene, Polypropylene, Polystyrene, Ethylene Glycol, PVC, PET, Ethylene Oxide, Propylene Oxide, Linear Olefins, ABS, SAN, PTA

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STRATEGY



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OBJECTIVE 1 PUTTING SAFETY FIRST



**"SAFETY IS OUR NO. 1 PRIORITY
AND A KEY FACTOR
IN OUR INDUSTRIAL PERFORMANCE"**

Philippe Sauquet
President, Refining & Chemicals



© Total

Safety Targets
**ZERO
FATALITIES**
TRIR:
<1.6 in 2015
<1 in 2020



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SIX KEY STRATEGIC OBJECTIVES



BECOME AN INTEGRATED, MORE COMPETITIVE PRODUCER

- 1 Make safety and preventing major environmental risks an absolute priority.**
- 2 Adapt and optimize the production base in Europe:** Cut costs and lower break-even points, develop operational excellence, position our most efficient sites as industry leaders and focus investments on large integrated complexes.
- 3 Develop petrochemicals in the Middle East and our supply to fast-growing Asian markets:** Secure access to cost-advantaged resources and establish a selective local presence in Asia.
- 4 Americas:** Leverage the U.S.'s competitive advantage in energy to consolidate and grow our production base.
- 5 Differentiate through innovative processes, products and solutions** for our customers.
- 6 Support Hutchinson's and Atotech's profitable growth**, in line with Total's strategy.

A SAFE AND COMPETITIVE REFINING & CHEMICALS FOCUSED ON EMERGING MARKETS



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REFINING & PETROCHEMICALS OPERATIONS



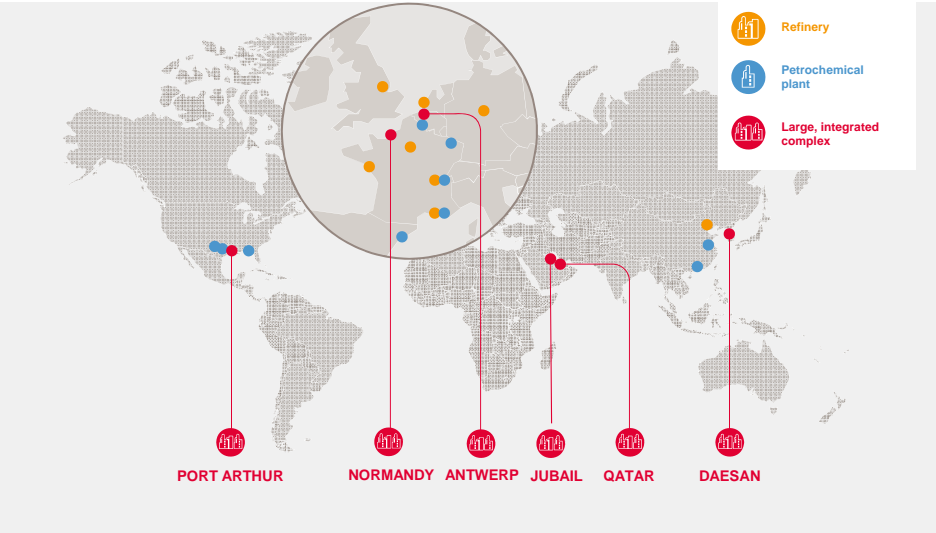
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FACILITIES





Contents





-  Refinery
-  Petrochemical plant
-  Large, integrated complex



PORT ARTHUR


NORMANDY


ANTWERP



JUBAIL


QATAR



DAESAN

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NORMANDY COMPLEX












Contents

- Upgraded facilities
- New configuration operational in early 2014
- Optimized product mix:
 - Less gasoline
 - More diesel
- Enhanced energy efficiency and lower carbon emissions
- Integrated refining and petrochemicals operations


NORMANDY, FRANCE

Refining capacity:
247,000 b/d

Special products: Lubricants and special fluids


Ethylene	Butadiene	Benzene	SM*	Paraxylene
 515 kt/y	 75 kt/y	 370 kt/y	 650 kt/y	 130 kt/y
HDPE	EVA**	PP	PS	
 280 kt/y	 45 kt/y	 240 kt/y	 85 kt/y	

*Styrene Monomer – ** Ethylene vinyl acetate copolymer



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O

ANTWERP COMPLEX

[Contents](#)

- Three integrated facilities:
Third-largest refinery in Europe, an olefins plant and a polymer plant
- Located in the center of the Amsterdam/Rotterdam/Antwerp (ARA) hub
- €1 billion invested to align output with demand by increasing production of distillates (diesel)
- New conversion unit

ANTWERP, BELGIUM

Refining capacity:
338,000 b/d

Special products: **MTBE & ETBE**

Ethylene	Benzene	Toluene
1,075 <small>kt/y</small>	240 <small>kt/y</small>	215 <small>kt/y</small>

HDPE*
480 kt/y

(4 lines)
Line A + B closed at the end of 2014
(70 kt/y)

* High-density polyethylene

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O

SATORP COMPLEX

[Contents](#)

- Complex refinery
- Access to advantaged feedstock
- Units operational since mid-2014
- Integrated with petrochemical units producing paraxylene, propylene and benzene
- Joint venture between Saudi Aramco (62.5%) and Total (37.5%)

JUBAIL, SAUDI ARABIA

Refining capacity:
386,000 b/d

Special products: **coke**

Propylene	Benzene	Paraxylene
200 <small>kt/y</small>	140 <small>kt/y</small>	700 <small>kt/y</small>

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PORT ARTHUR COMPLEX

- Deep conversion refinery
- BTP*, a world-class steam cracker in partnership with BASF
- Improved cracker flexibility:
 - 40% ethane
 - 40% butane and propane
 - 20% naphtha

* BASF Total Petrochemicals LLC

PORT ARTHUR, TEXAS, UNITED STATES

Refining capacity:
169,000 b/d

Special products: special fluids and coke

C3 splitter

1,200

at Mont Belvieu
(Total: 33%)

BTP steam cracker:

Ethylene	Benzene	Toluene	C3 metathesis process	Butadiene
1,090	150	80	300	300

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HTC – DAESAN COMPLEX

HTC: HANWHA TOTAL PETROCHEMICALS CO.
(50/50 JOINT VENTURE)

- Large complex producing a wide range of monomers and polyolefins
- Vast expansion project completed in July 2014: new aromatics unit, new condensate splitter and a 240-kt/y EVA line
- Perfectly positioned to access the Chinese market

DAESAN, SOUTH KOREA

Refining capacity (condensate refinery):
225,000 b/d

Ethylene	C3 metathesis process	Butadiene	Benzene	SM	Paraxylene
1,000	230	120	1,170	1,040	1,800
LLDPE*	HDPE	LDPE**/EVA	EVA	PP	
125	170	150	240	700	

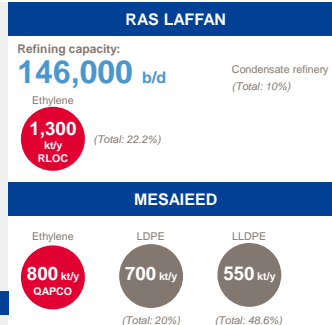
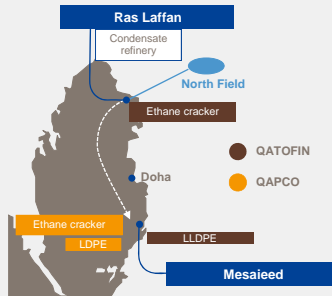
* Linear low-density polyethylene ** Low-density polyethylene

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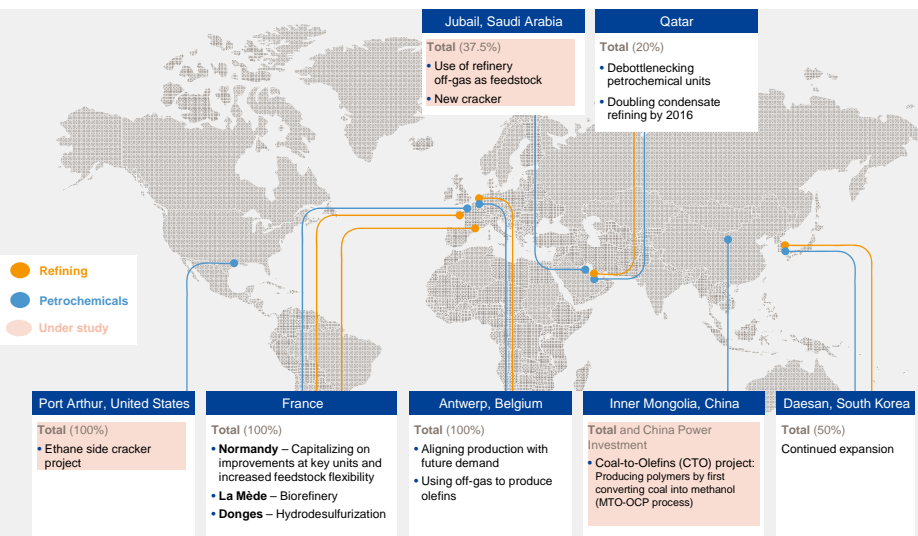
INTEGRATING OUR ASSETS IN QATAR



- Two ethane crackers and large polyethylene units
- Rated capacity of 1.15 million tons a year of polymers
- Condensate refinery: capacity to double in 2016 to 292 kb/d
- Partner: Qatar Petroleum



PROJECTS UNDER WAY



TECH CENTERS AND HEADQUARTERS

Contents

Locations in Europe

Tech Centers

- Total Research & Technology Feluy (TRTF)
- Total Research & Technology Gonfreville (TRTG)
- Dunkirk Tech Center (ATCO)
- Total Research & Technology La Porte – Texas, USA

Headquarters

- Paris, France
- Brussels, Belgium
- + some entities in Dunkirk, TRTF and TRTG

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5 SPECIALTY CHEMICALS

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TWO SPECIALTY CHEMICALS AFFILIATES THAT HOLD LEADERSHIP POSITIONS IN THEIR SEGMENTS



→ Technological expertise in chemicals and materials science

Elastomers



2014 revenue:
\$4.6 billion

Metal and Surface Finishing



2014 revenue:
\$1.3 billion



HUTCHINSON® A WORLD LEADER IN RUBBER AND THERMOPLASTIC ELASTOMER PRODUCTS



2014 revenue:
\$4.6 billion

Automotive



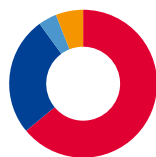
2 major markets
70/30

- + Sealing systems
- + Fluid transfer systems
- + Vibration, acoustic and thermal insulation
- + Transmission and mobility

Aerospace & Other Industries



Breakdown of 2014 sales by region



- Europe 63%
- North America 26%
- South America 6%
- Asia and Rest of the World 5%

Strategy

- Growth in target markets:
 - Aerospace in the United States and Asia
 - Automotive in growth regions
- Innovation and product differentiation
- Excellence



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ENJOY THIS FRUITFUL MEETING



THANK YOU FOR YOUR ATTENTION

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ADDITIONAL INFORMATION



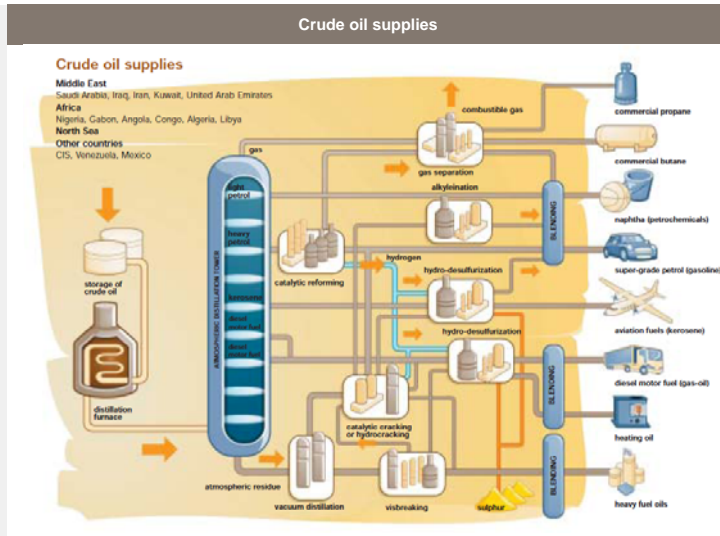
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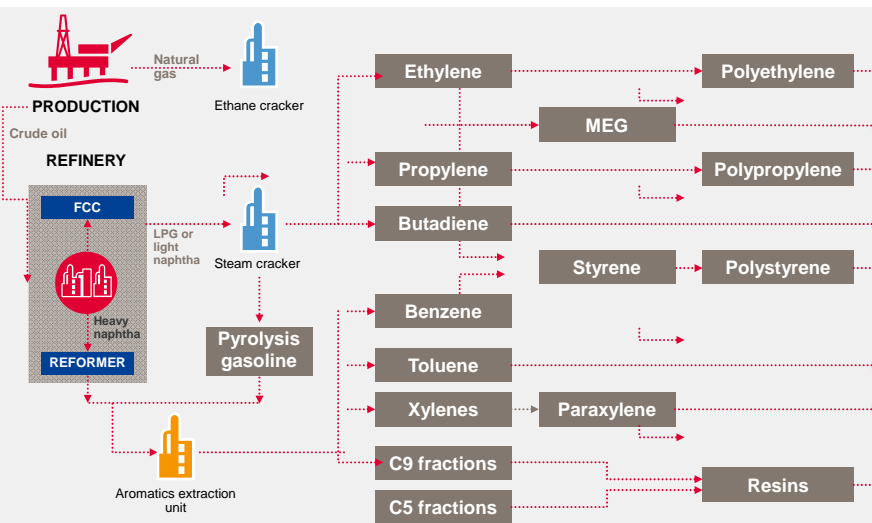
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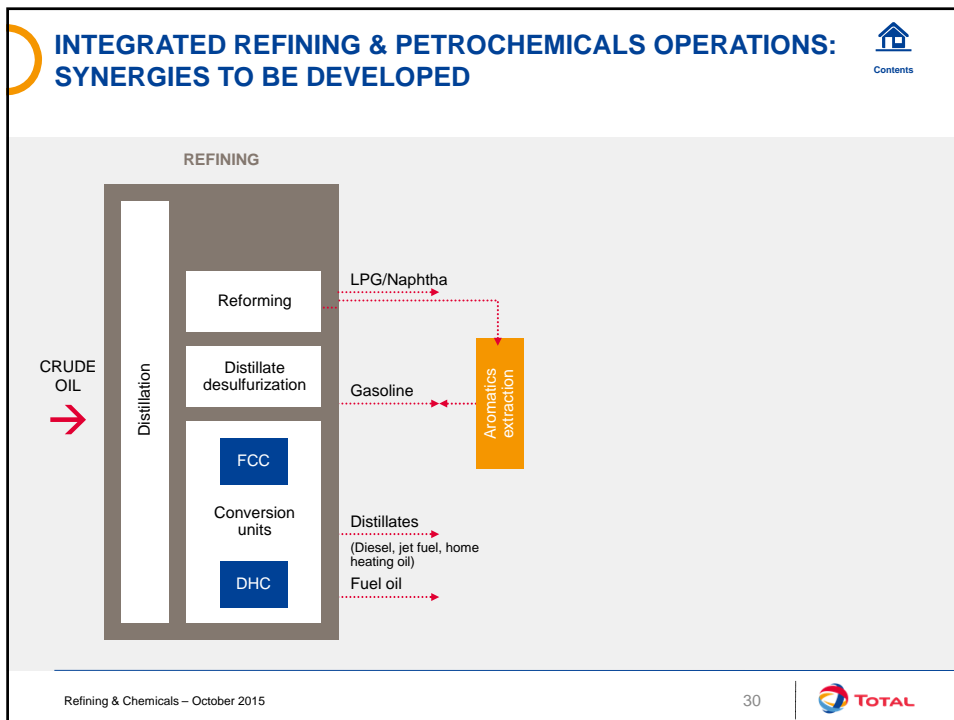
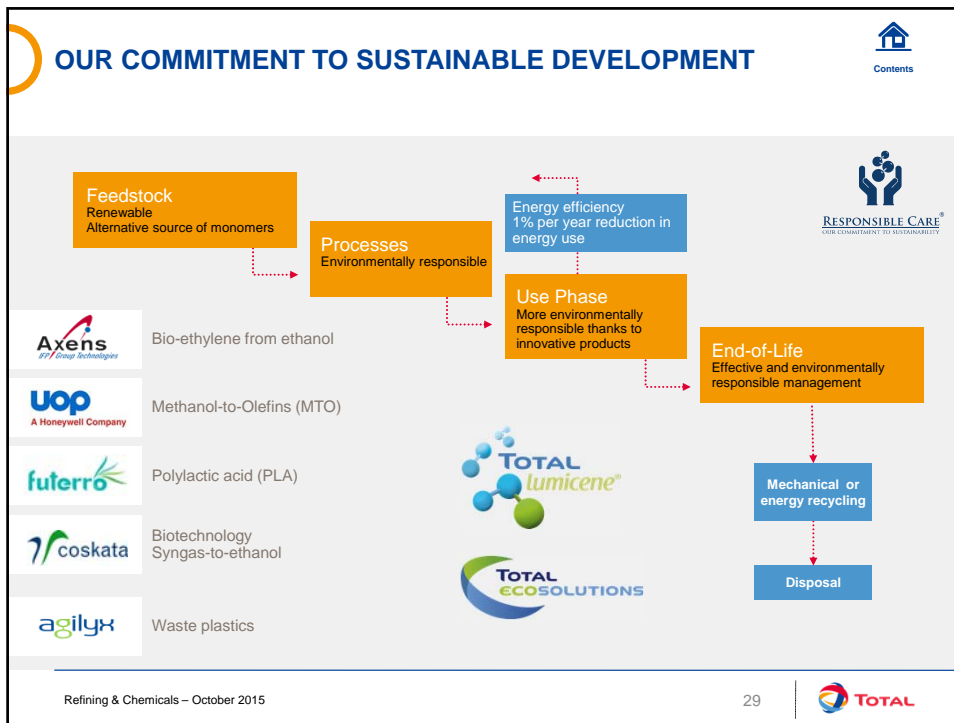


GENERIC REFINERY PROCESS SCHEME

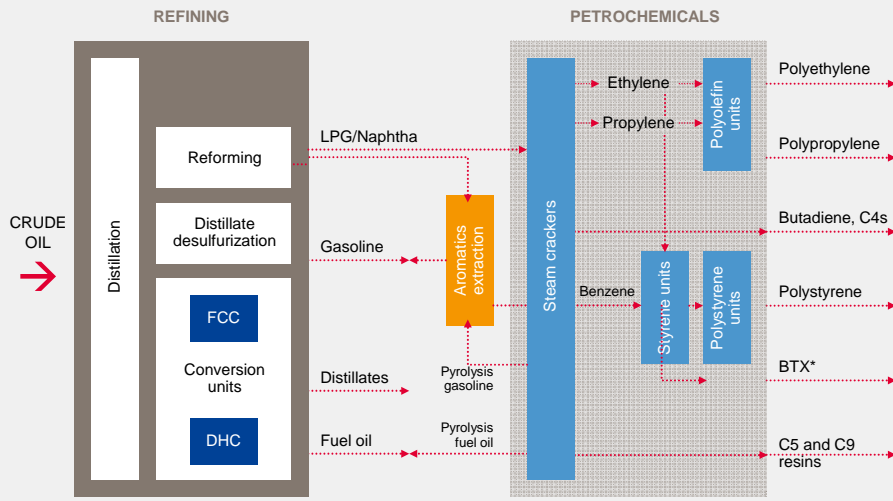


FROM OIL & GAS TO POLYMERS, AN INTEGRATED PROCESS

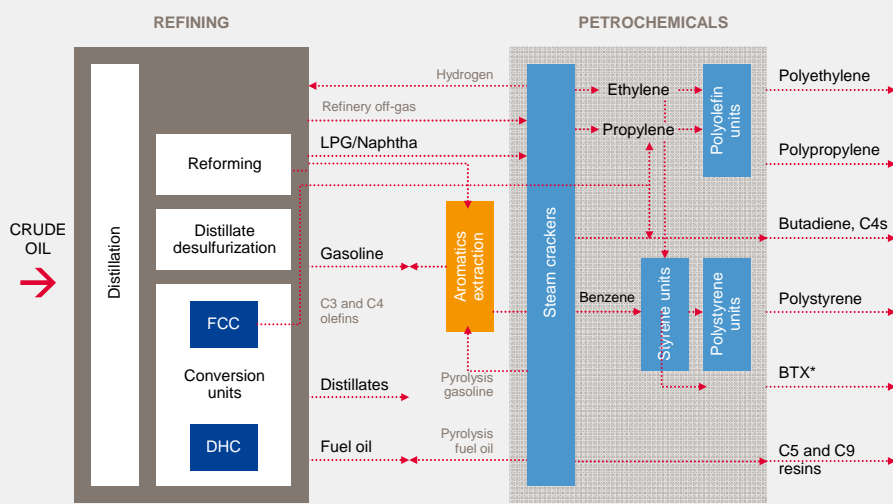




INTEGRATED REFINING & PETROCHEMICALS OPERATIONS: SYNERGIES TO BE DEVELOPED



INTEGRATED REFINING & PETROCHEMICALS OPERATIONS: SYNERGIES TO BE DEVELOPED



Appendix 2

List of participants

Participants EFC WP15 meeting 26th April 2016 Paris (France)

NAME		COMPANY	COUNTRY
Augustin	Christel	Total Refining & Chemicals	FRANCE
Bour Beucler	Valerie	Nalco Champion	FRANCE
Cicomascolo	Francesco	Böhler Welding Holding GmbH	GERMANY
Claesen	Chris J	Nalco Champion	BELGIUM
de Heus	Rob	Sitech services	NETHERLANDS
Delage	Jean Michel	AXENS - IFP Technology Group	FRANCE
Dubois	Francois	AXENS - IFP Technology Group	FRANCE
Dupoiron	François	Total Refining & Chemicals	FRANCE
El Kamel	Meriem	Technip	FRANCE
Escorza	Erick	Tenaris Dalmine	ITALY
Fenton	Stephen	Steve Fenton Consultants	UK
Hofmeister	Martin	Bayernoil Raffineriegesellschaft mbH	GERMANY
Houille	Patrice	Patrice Houille Corrosion Service - MTI	FRANCE
Kamionka	Marc	Technip	FRANCE
Koller	Swen	Holborn Europa Raffinerie GMBH	GERMANY
Kus	Slawomir	Honeywell Control Systems Ltd	UK
Lorkin	David	Permasense Limited	UK
Loyan	Sophie	Total	FRANCE
Marcolin	Giacomo	Tenaris Dalmine	ITALY
MeLampy	Michael	PPG Protective & Marine Coatings	USA
Pothuaud	Alain	GE Water & Distributed Power	FRANCE
Rajek	Joachim	TÜV AUSTRIA HOLDING AG	AUSTRIA
Ropital	François	IFP Energies nouvelles	FRANCE
Roquencourt	Mathieu	Petrolneos	FRANCE
Rothwell	John	TWI	UK
Sager	Patrick	Total	FRANCE
Shapcott	Stefen	Johnson Matthey Process Technologies	UK
Suleiman	Mabruk	Takreer	UNITED ARAB EMIRATES
Surbled	Antoine	A.S – CORR CONSULT	FRANCE
Tabaud	Frederic	BP R< (Refining and Logistic Technology)	NETHERLANDS
van Rodijnen	Fred	Oerlikon metco	GERMANY
van Roij	Johan	Shell Global Solutions International B.V.	NETHERLANDS
Weisang-Hoinard	Francois	Outokumpu	FRANCE

Appendix 3

Agenda of the 27th April 2016

EFC-MTI Round table on

High Temperature Hydrogen Attack

Wednesday, 27 April 2016 MTI/EFC HTHA Roundtable EFC event N°408

Chimie ParisTech
Ecole Nationale Supérieure de Chimie
11, rue Pierre et Marie Curie
75231 Paris Cedex 05

Welcome and Introduction Safety rules	10:00 - 10:15 am	François Ropital ,(EFC , WP15) Patrice Houille , MTI
Key Notes : Some considerations on the thermodynamics and kinetics of carbide attack in HTHA	10:15 - 11:00 am	Michael Schutze, Dechema
Hydrogen In Refinery : HTHA and Creep	11:00 - 11:45 am	Sylvain Pillot, Arcelor
Update on API 941	11:45 - 12:30 am	John Houben, Exxon Mobil
<i>Lunch</i>	<i>12:30 - 1:30 pm</i>	
Company Experiences: Total	1:30 - 2:15 pm	Charles Le Neve, Total
HTHA Damages	2:15 - 2:45 pm	Karl-Friedrich Schneider, Basf
Air Liquide presentation	2:45 - 3:15 pm	Sophie Wastiaux, Air Liquide
Overview of the Pv and Pw methods and other parametric methods for HTHA rate prediction	3:15 - 3:45 pm	Brian Olsen , Stress Engineering
<i>Break</i>	<i>3:45 - 4:00 pm</i>	
"The Use & Misuse of NDE for the identification of HTHA."	4:00 - 4:40 pm	Peter Conlin, Sonomatic
Simulation of NDT	4:40 - 5:20pm	Philippe Dubois, Extende Sébastien Lonné, Extende
R&D project to develop a NDT to measure hydrogen charging in metal and to understand the problem before cracks appear.	5:20 -5:40pm	Joachim Radjek, Tuv Austria
Panel Discussion/ Conclusion	5:40- 6:15 pm	

Appendix 4

EFC WP15 Activities

(F. Ropital)



Presentation of the activities of WP15

European Federation of Corrosion (EFC)

- Federation of 29 National Associations
- 21 Working Parties (WP) and 1 Task Force
- Annual Corrosion congress « Eurocorr »
- Thematic workshops and symposiums
- Working Party meetings (for WP15 twice a year)
- Publications
- EFC - NACE agreement (20% discount on books price)
- for more information <http://www.efcweb.org>

EFC WP15 Spring meeting 26 April 2016 Paris - France

1



EFC Working Party 15 « Corrosion in Refinery » Activities

Who is an EFC member

To be an EFC member you (individually or your company, university) has to be member of one of 29 national EFC "member societies". Your company or university can now also an affiliate member.

For example:

in Norway: Norsk Korrojonstekniske Forening

in France: Cefracor

in Germany: Dechema or GfKORR

in UK: Institute of Corrosion or IOM or NACE Europe

in The Netherlands: Bond voor Materialenkennis

in Poland: Polish Corrosion Society

.....

You will find all these information on www.efcweb.org or in the EFC Newsletter

Benefits to be an EFC member:

- 20% discount on EFC Publications and NACE Publications

- reduction at the Eurocorr conference

- Access the [new EFC web restricted pages](#) (papers of the previous Eurocorr Conference) via your national corrosion society web pages

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EFC Working Parties

<http://www.efcweb.org>

- WP 1: Corrosion Inhibition
- WP 3: High Temperature
- WP 4: Nuclear Corrosion
- WP 5: Environmental Sensitive Fracture
- WP 6: Surface Science and Mechanisms of corrosion and protection
- WP 7: Education
- WP 8: Testing
- WP 9: Marine Corrosion
- WP 10: Microbial Corrosion
- WP 11: Corrosion of reinforcement in concrete
- WP 12: Computer based information systems
- WP 13: Corrosion in oil and gas production
- WP 14: Coatings
- WP 15: Corrosion in the refinery industry
(created in sept. 96 with John Harston as first chairman)
- WP 16: Cathodic protection
- WP 17: Automotive
- WP 18: Tribocorrosion
- WP 19: Corrosion of polymer materials
- WP 20: Corrosion by drinking waters
- WP 21: Corrosion of archaeological and historical artefacts
- WP 22: Corrosion control in aerospace
- Task Force on Corrosion in CO₂ Capture Storage (CCS) applications
- Task Force on Corrosion reliability of Electronics

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EFC Working Party 15 « Corrosion in Refinery » Activities

<http://www.efcweb.org/Working+Parties-p-104085/WP%2B15-p-104111.html>

Chairman: Francois Ropital

Deputy Chairman: Johan Van Roij

The following are the main areas being pursued by the Working Party:

Information Exchange

Sharing of refinery materials /corrosion experiences by operating company representatives.

Forum for Technology

Sharing materials/ corrosion/ protection/ monitoring information by providers

Eurocorr Conferences

WP Meetings

One WP 15 working party meeting in Spring,
One meeting at Eurocorr in September in conjunction with the conference,

Publications - Guidelines

EFC WP15 Spring meeting 26 April 2016 Paris - France

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List of the WP15 spring meetings :

10 April 2003	Pernis - NL (Shell)
8-9 March 2004	Milan -Italy (ENI)
17-18 March 2005	Trondheim- Norway (Statoil)
31 March 2006	Porto Maghera - Italy (ENI)
26 April 2007	Paris - France (Total)
15 April 2008	Leiden -NL (Nalco)
23 April 2009	Vienna - Austria (Borealis)
22 June 2010	Budapest - Hungary (MOL)
14 April 2011	Paris - France (EFC Head offices)
26 April 2012	Amsterdam - NL (Shell)
9 April 2013	Paris - France (Total)
8 April 2014	Mechelen - Belgium (Borealis)
14 April 2015	Leiden -NL (Nalco)
26 April 2016	Paris - France (Total)

Publications from WP15

• **EFC Guideline n°40 « Prevention of corrosion by cooling waters »** available from <http://www.oxbowbooks.com/oxbow/working-party-report-on-control-of-corrosion-in-cooling-waters.html>

• Update in relation with Nace document 11106 "Monitoring and adjustment of cooling water treatment operating parameters" Task Group 152 on cooling water systems

• **EFC Guideline n° 46 on corrosion in amine units**
http://store.elsevier.com/product.jsp?locale=en_US&isbn=9781845692377

• **EFC Guideline n° 42 Collection of selected papers**
http://store.elsevier.com/product.jsp?locale=en_US&isbn=9781845692339

• **EFC Guideline n° 55 Corrosion Under Insulation *New edition nov. 2015***
<http://store.elsevier.com/product.jsp?isbn=9780081007143&pagename=search>

• Future publications : suggestions ?

- best practice guideline to avoid and characterize stress relaxation cracking ?



EUROPÄISCHE FÖDERATION KORROSION
EUROPEAN FEDERATION OF CORROSION
FÉDÉRATION EUROPÉENNE DE LA CORROSION

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Who we are

EFC Membership

Working Parties

WP Corrosion and Scale Inhibition
WP Corrosion by Hot Gases and
Combustion Products
WP Nuclear Corrosion
WP Environment Sensitive
Fracture
WP Surface Science and
Mechanisms of Corrosion and
Protection
WP Corrosion Education
WP Physico-chemical Methods of
Corrosion Testing
WP Marine Corrosion
WP Microbial Corrosion
WP Corrosion of Steel in Concrete
WP Corrosion in Oil and Gas
Production
WP Coatings
WP Corrosion in the Refinery
Industry
WP 15 Refinery Corrosion Atlas
CJA Restricted Web-Page
WP Cathodic Protection
WP Automotive Corrosion

Welcome > Working Parties > WP Corrosion in the Refinery Industry > WP 15 Refinery Corrosion Atlas

EFC Working Party 15: Corrosion in the Refinery Industry

WP 15 REFINERY CORROSION ATLAS

On this page you will find some corrosion failure cases from the refinery and process industries.

These documents are only given for information and do not engage EFC.

Failure case n°1: High temperature corrosion of a first stage reactor of a hydrocracking unit

Failure case n°2: Chloride stress corrosion cracking of a H2S stripping tower in a hydrosulfuration unit

Failure case n°3: Creep and cracks in a hydrosulfuration unit

Failure case n°4: Chloride stress corrosion cracking of mounting hardware in a FCC

Failure case n°5: Metal ducting corrosion of a furnace tube in reforming unit

Failure case n°6: Sulfidation in an atmospheric distillation unit

Failure case n°7: HF stress corrosion cracking in an alkylation unit

Failure case n°8: Carbonate stress corrosion cracking in an FCC unit

If you would like to add other failure cases, you can complete the enclosed file and send it to
François Ropital email: francois.ropital@epn.fr

EFC WP15 Spring meeting 26 April 2016 Paris - France

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Eurocorr 2016

Montpellier France 11-15 September 2016

Authors will be informed by the end of April

Refinery corrosion session on Tuesday 13th September

Annual WP15 working party meeting during Eurocorr on Tuesday 13th September

On Monday 12th September afternoon the second part of the "Polymer session" will be dedicated to corrosion of polymers in the refinery, petrochemical and chemical industries



EUROCORR
2016

<http://eurocorr2016.org/>

8

. Joint sessions with other EFC WP at Eurocorr (2017 Prague-Czech Republic, 2018 Krakow-Poland) on which topics?

Eurocorr 2017 (3-7 September) will be coupled with the 20th International Corrosion Council (ICC) congress (that takes place every 3 years in different parts of the world)

- Update - New EFC guidelines
 - Corrosion in amine treatment units
 - best practice guideline to avoid and characterize stress relaxation cracking ?
- Education - qualification - certification
 - List of "corrosion refinery" related courses on EFC website ?
 - Proposal of courses within Eurocorr ?

•15-20 May 2016

High Temperature Corrosion and Protection of Materials - Les Embiez France
<http://www.htcpm2016.com/>



•11-15 September 2016

EUROCORR 2016 Montpellier France



•26-30 March 2017

CORROSION 2017 NACE Conf New Orleans

3-7 September 2017

EUROCORR 2017 Prague Czech Republic



Look at the Website: www.efcweb.org/Events

Appendix 5

Advancement of the task force on amine corrosion – Minutes of the 25th April 2016 meeting

(J. Van Roij)

Minutes of the “Amine unit corrosion” Taskforce; 25 April 2016,

With thanks to Cefracor (and Francois Ropital & Jean Kittel) the first Taskforce “Amine plant corrosion” was held at Cefracor, Paris, 25 April 2016.

Present:

Jean Kittel (IFPEN)

Slawomir Kus (Honeywell)

Sophie Loyan (Total)

Mabruk Suleiman (Takreer)

Johan van Roij (Shell)

Michel Bonis informed that he could not be present.

1. Fix the objectives of the work.
 - Extend the scope to gas plants (as proposed during the joined EFC WP13+WP15 meeting Sep 2015); as the current title of the publication 46 indicates the Refinery as the scope of the publication it was decided to change the title into: “Corrosion in Amine Units”. The publication needs to become a more practical guideline on corrosion and corrosion mitigation in amine units.
2. Conduct industry survey to obtain corrosion experience (observed corrosion / corrosion rates, solvent type, operating conditions, etc). Target to have the questionnaire available at Eurocorr and distribute it to EFC13 and EFC15 participants.
Action for everybody: Think of questions for the questionnaire and format; to be discussed during next meeting.
3. Literature survey: SK, MS & JK already performed an extensive literature survey. Later it needs to be discussed whether this is sufficient and how to include this in the publication.
4. Extend the general section of the Publication 46 to in more detail describe the gastreating process and the corrosion mechanisms in amine units. Incorporate new aspects: e.g. materials selection, practical aspects, integrity operating windows (in combination with parameter limits), corrosion monitoring, inspection, corrosion modeling.
 - Mabruk Suleiman already has written a process description for a publication. With a little rework this looks very suitable for our Publication 46. Target to have draft ready in Sep 2016?
 - Johan van Roij already has written corrosion descriptions for a publication (and Shell internal corrosion courses). With some rework this can be made suitable for our Publication 46 (obviously on basis of the obtained experience mitigation measures, IOWs, etc. need to be added later). Target to have a draft basic description ready in Sep 2016?
 - It was decided to have the corrosion and corrosion mitigation descriptions (including inspection and corrosion monitoring) described and structured per corrosion loop. Jean Kittel and Johan van Roij showed examples of corrosion loops. JK already forwarded his corrosion loops to JvR. Action: Johan van Roij will look at the available corrosion loop

descriptions and extract a proposal for corrosion loop descriptions. Target to have draft ready in Sep 2016?

5. Next meeting (teleconference): May 19th, 9:00 AM

6. New members task force.

During the EFC WP15 spring meeting two new people volunteered for the Taskforce: Steve Fenton (Protective Polymers Ltd) and Chris Claesen (NALCO Champion). They are welcomed and receive an invitation for the teleconference May 19.

Appendix 6

Continuous corrosion monitoring in amine acid gas units – a European refinery case study

(D. Lorkin)



permasense[®]

Experts in remote monitoring solutions

**Continuous Corrosion Monitoring in Amine Units
– A European Refinery Case Study**
David Lorkin, European Sales Manager

EFC Working Party 15 Corrosion Refinery Industry Meeting – Paris, 26th April 2016

Industry challenges: corrosion and erosion issues

Industry challenges

Company >

Solutions >

System outputs >

System benefits

Business case

Technology

Deployment >

Applications >

Summary

- **Well understood:**
 - Process conditions
 - Crude constituents
 - Abrasive solids
 - Inhibitors
 - Metallurgy
- **Not well understood:**
 - Rate of damage
 - Variability of rate of damage from above factors
- **Leading to:**
 - Conservative operations, and/or feedstocks
 - Unplanned outages, and/or loss of containment



Deployed worldwide, extensive references

Industry challenges

Company >

Solutions >

System outputs >

System benefits

Business case

Technology

Deployment >

Applications >

Summary

>12,000 sensors in >100 facilities in >20 countries

- Refineries
- LNG plants
- Offshore platforms
- Onshore oil gathering systems
- Pipelines
- FPSOs
- Power plants (gas and geothermal)

Permasense System installed on Process Units

Industry challenges

Company

Solutions

Short-range

Long-range

Modular

System outputs >

System benefits

Business case

Technology

Deployment >

Applications >

Summary



Proven Applications in Refining

Industry challenges

Company

Solutions >

System outputs >

System benefits

Business case

Technology

Deployment >

Applications

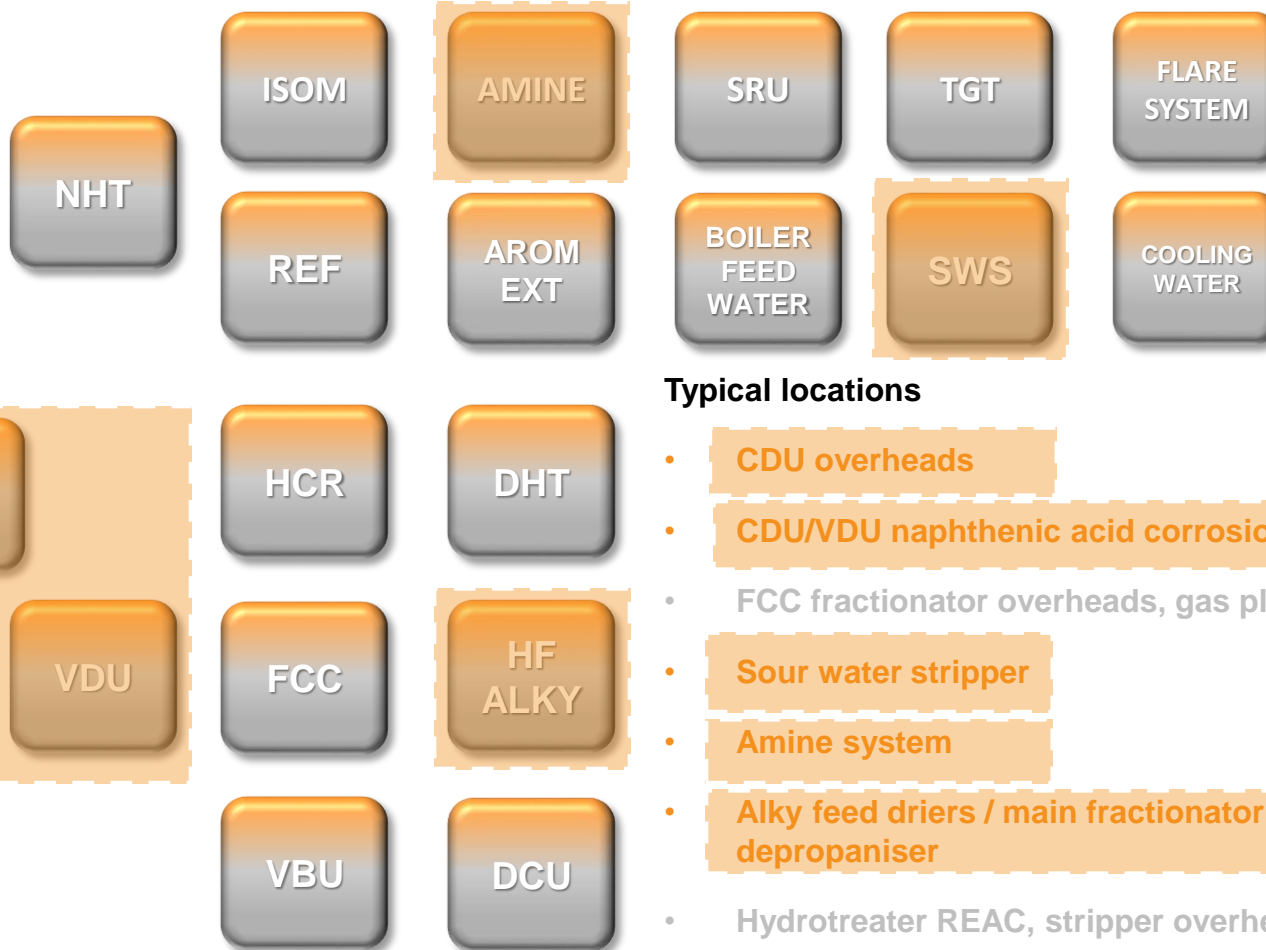
Refining

Upstream

Power generation

Geothermal

Summary



Typical locations

- CDU overheads
- CDU/VDU naphthenic acid corrosion
- FCC fractionator overheads, gas plant
- Sour water stripper
- Amine system
- Alky feed driers / main fractionator / depropaniser
- Hydrotreater REAC, stripper overheads
- Isomerisation, reformer stabilisers

Sensors installed on all metals found in oil & gas: from carbon and chrome steels & chrome steels to super-alloys

Challenges of Corrosion in Amine Units

Industry challenges

Company

Solutions >

System outputs >

System benefits

Business case

Technology

Deployment >

Applications

Refining

Upstream

Power generation

Geothermal

Summary

- Corrosion by CO₂ and H₂S (and possibly accumulation of NH₄, HCN, organic acids)
- Corrosion in vapour phase, amine solution and regenerator reflux
 - Wet acid gas corrosion of carbon steel from reaction of CO₂ and H₂S with iron through a thin liquid film
 - Amine solution corrosion of carbon steel in presence of aqueous amine
- Many variables:
 - Temperature
 - Choice of Amine-type
 - Acid Gas Loading
 - Velocity and Wall Shear Stress
 - Impurities (heat-stable amine salts)
 - CO₂ to H₂S ratio

Modelled Corrosion Rates for Carbon Steel as a Function of Velocity and H₂S Concentration

Averaged corrosion rates for carbon steel				
Velocity [ft/s]	H ₂ S loading as molar ratio to MEA			
	0.2	0.4	0.6	0.8
0	0	0	1	1
20	8	12	12	12
40	12	14	16	20
60	13	16	20	43
80	16	18	25	66

	< 5 mpy
	5 - 10 mpy
	10 - 15 mpy
	15 - 20 mpy
	20 - 50 mpy
	> 50 mpy

Variation of corrosion rates for carbon steel with amine acid gas loading and velocity. As expected, high rich amine H₂S loading, combined with high velocity, results in higher corrosion rates.

Amine Unit – Simplified Process Flow Diagram showing High Risk Corrosion Areas

Industry challenges

Company

Solutions >

System outputs >

System benefits

Business case

Technology

Deployment >

Applications

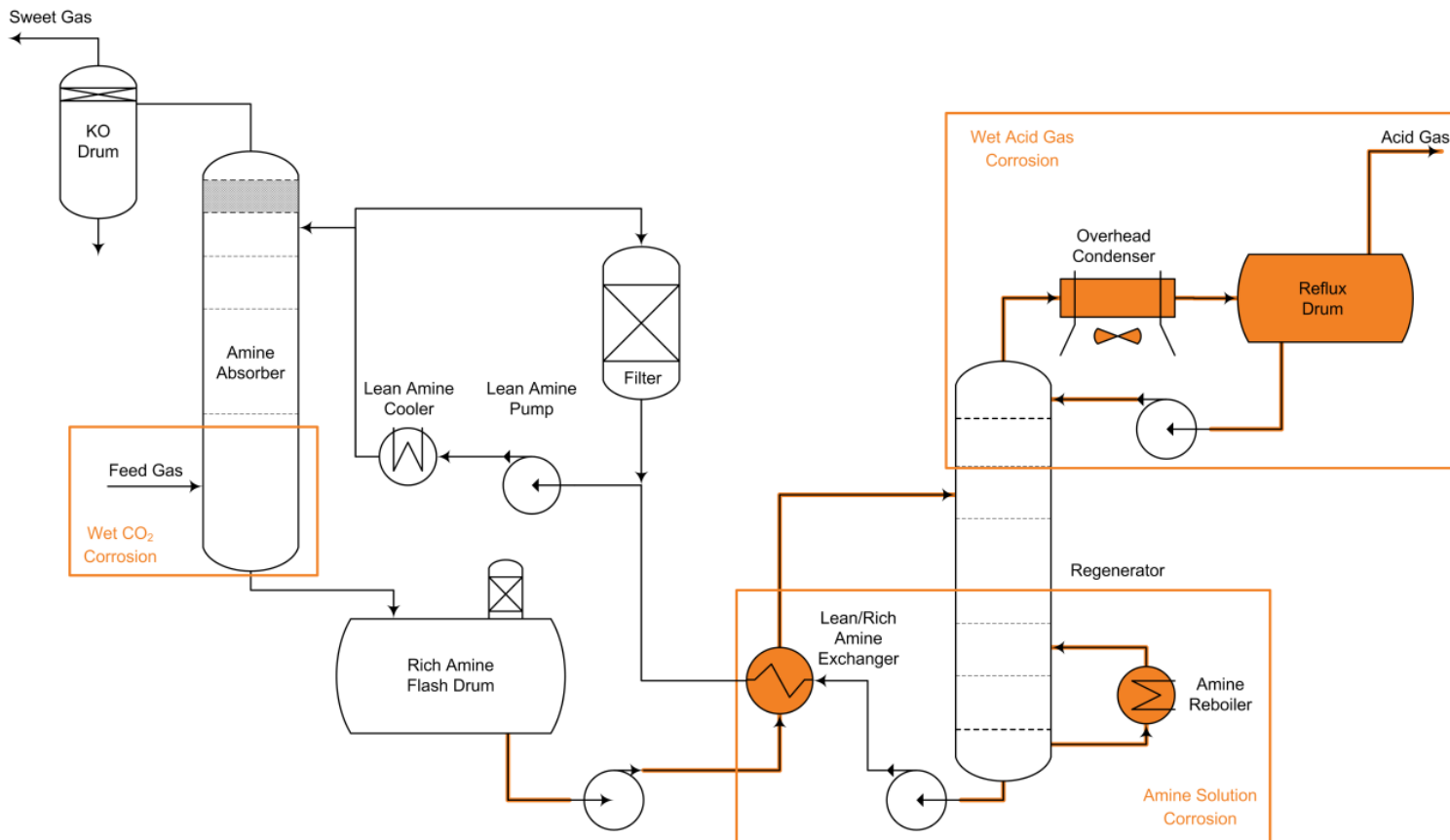
Refining

Upstream

Power generation

Geothermal

Summary



Proposed Location of Sensors in Amine Absorption and Regeneration Sections

Industry challenges

Company

Solutions >

System outputs >

System benefits

Business case

Technology

Deployment >

Applications

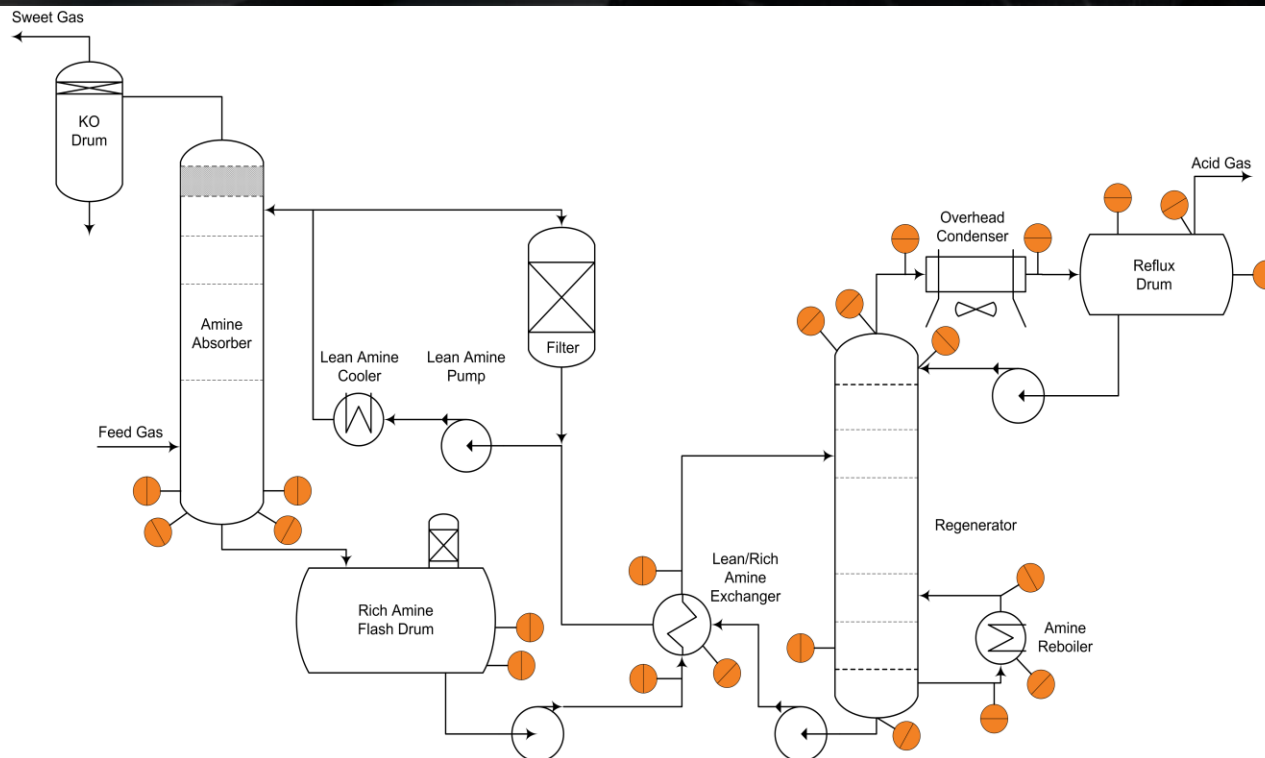
Refining

Upstream

Power generation

Geothermal

Summary



- Corrosion issues

- High gas loading
- Heat stable salts
- Amine degradation
- Oxygen contamination

- Corrosion is uniform and not so localised

- Fewer measurements needed to represent entire system
- Elbows, bends, tees, exchanger shells
- Lower temperature application, ET210s

Typically 10-15 measurement locations, 2-3 sensors per location

WT Sensor:

Industry challenges

Company

Solutions >

System outputs >

System benefits

Business case

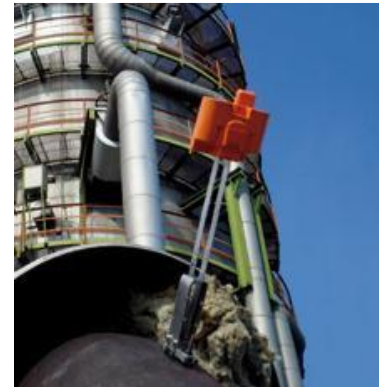
Technology

Deployment >

Applications >

Summary

- **Standard sensor** length accommodates
 - temperatures up to **600 °C**, e.g. refinery crude unit pipework
 - temperatures down to **-180 °C**, e.g. LNG plants
 - thick insulation or lagging
- **Compact** – short waveguide – version where extreme temperature capability not required
- Attached by Studs (full temperature range) or Clamps (up to approx. 250°C)
- Temperature and Material Compensation and AXC



ET210 sensors for easy installation in locations operating under 120 °C (250 °F)

PERMASENSE QUALITY DATA

INTRINSICALLY SAFE SENSORS

ULTRA-LOW POWER
CONSUMPTION

NON-INTRUSIVE

NO NEED TO REMOVE EXTERNAL
COATINGS

WirelessHART TECHNOLOGY



NO HEAVY CLAMPS /
CABLING REQUIRED

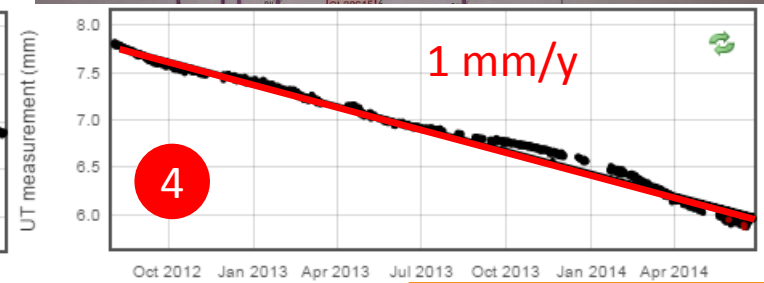
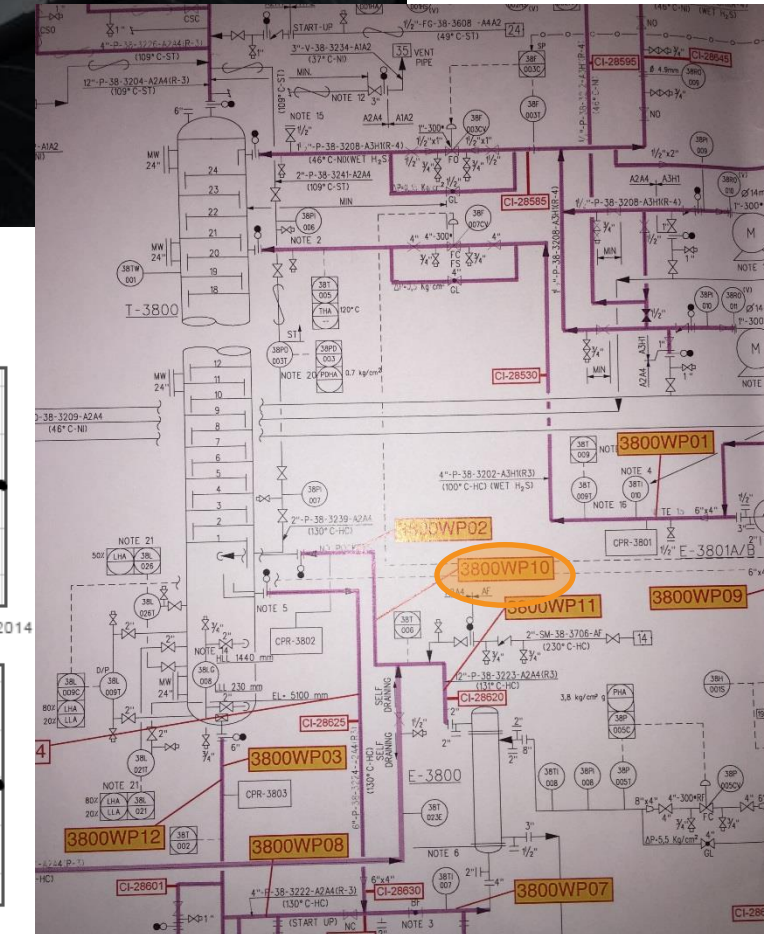
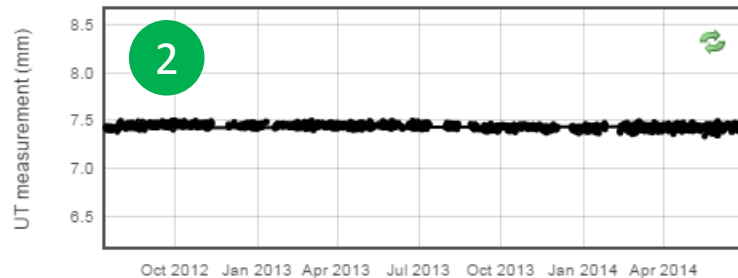
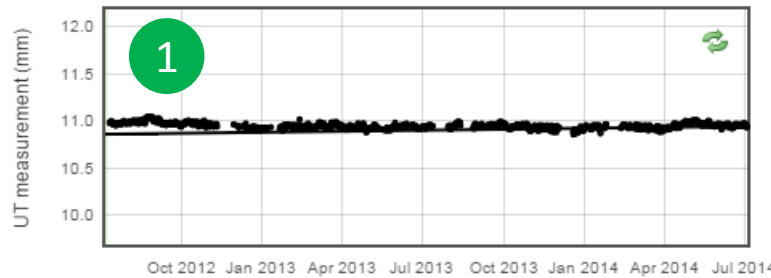
Quick and easy installation:
Magnetically – heat-resistant plastic
strap provides additional robustness



No need to remove coating:
Measures through up to
1mm of external coating

European Refinery Case Study: Amine Absorption and Regeneration

- Customer has four amine absorber/regeneration trains
- All trains have a very similar configuration
- Markedly higher corrosion in Train 4 shown on reboiler return sensor data
- Feed showed high CO₂ content – preferential routing of FCC sour gas to this train
- Feeds were redistributed to dilute the high CO₂ gas



Continuous high-quality data enables...

Industry challenges

Company

Solutions >

System outputs >

System benefits

Business case

Technology

Deployment >

Applications >

Summary

- **Confidence in asset management**
 - early identification of corrosion / erosion activity
 - understanding of causes – correlation with process condition changes
 - better forecasting of attainment of retirement thickness
- **Optimisation of prevention / mitigation strategies**
 - correlation of trends with inhibition strategy
- **Insight into impact of feedstock decisions (refineries)**
 - rapid feedback with changes in crude slate
- **Cost-effective, *safe* measurement**
 - no cost of repeat measurements
 - no shutdown for measurement
 - no personnel exposure to high-risk locations



Financial return from investment in Permasense technology

Industry challenges

Company

Solutions >

System outputs >

System benefits

Business case

Technology

Deployment >

Applications >

Summary

- SYSTEM PAYBACK TIME

Months to a Year

**SAVE
\$100,000s**

- **REDUCE OPEX SAFELY**
Enhance Maintenance Planning
Optimise Corrosion Control

Days to Weeks

**GENERATE
\$\$ MILLIONS**

- **OPTIMISE OPERATIONS**
Diversify Crude Basket
Maximise Availability
Enhance Refinery Flexibility

Very short

**AVOID
SPENDING/LOSSES
\$\$\$ TENS OF MILLIONS+**

- **AVOID MAJOR INCIDENTS**
Avoid Loss of Containment
Avoid Unplanned Outages



THANK YOU FOR YOUR ATTENTION



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Kuala Lumpur, Malaysia

Appendix 7

Perforation of plates in a Sulreen reactor

(S. Koller)

Sulfreen reactor damage

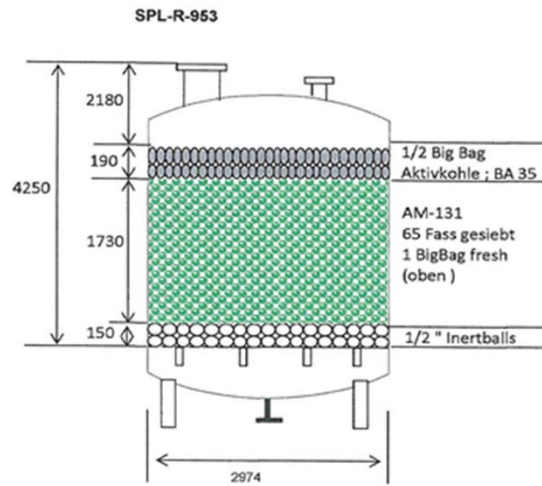


HOLBORN EUROPA RAFFINERIE
GMBH



-
1. Overview
 2. Description of the damage
 3. Damage analysis
 4. Conclusion / Discussion
-

Overview



Overview

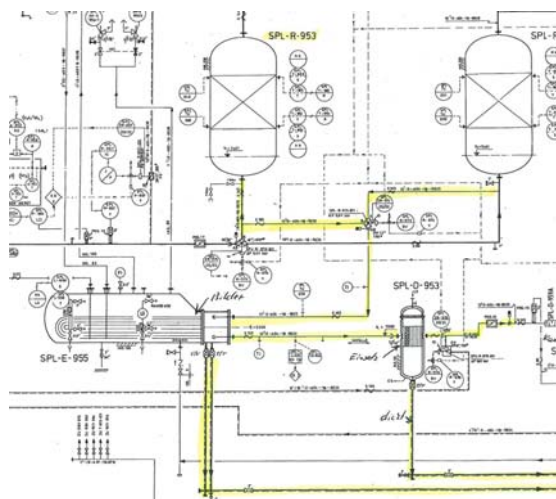


- Operation:
 - Tail gas from the Claus unit is cooled in the process gas cooler to ca. 135°C and passes through the sulphur separator before entering one of the two reactors.
 - Tail gas has usually 0,6 to 0,8 % H₂S and 0,3 to 0,4 % SO₂
- Regeneration:
 - For water and sulphur desorption a gas flow is established in the opposite direction with a temperature of ca. 320°C.
 - Regeneration gas has usually 0,2 to 0,3 % H₂S and up to 0,2 % SO₂

Overview - View into the reactor



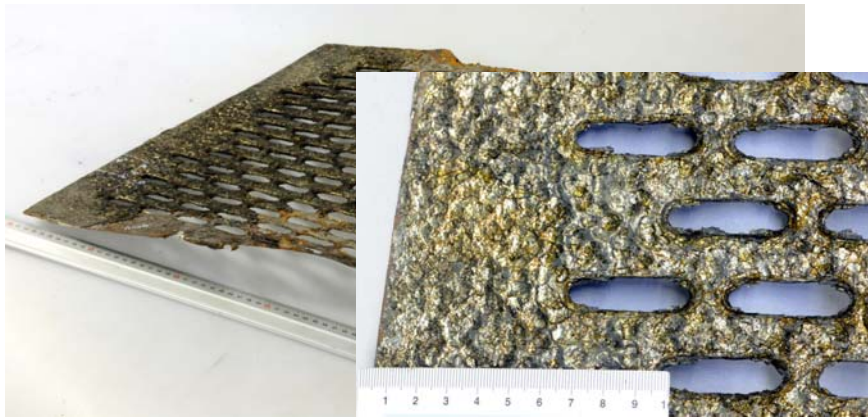
Overview - Cat in the reg-gas-system



Damage analysis



Distributor (support bottom)

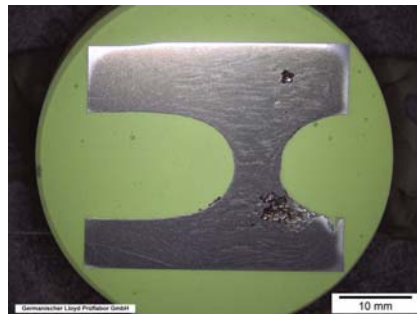
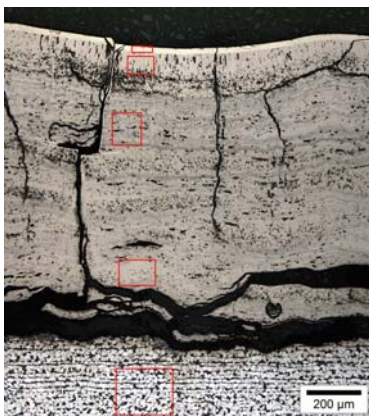


Damage analysis



Sulfidation

weak acid corrosion



Conclusion / Discussion



Material Selection:

1. High-temperature steel with sufficient corrosion addition and replacement every 10 years
2. High-temperature steel with TSA-coating
3. Stainless steel

Operating Parameter

1. Increase the process temperature to 150°C
2. Decrease the regeneration temperature to 315°C



Appendix 8

Sulfidation of a new ferritic stainless steel

(G. Marcolin)

EFC Meeting – Paris 2016



**A new ferritic grade
for Refining World**

Tenaris Overview



Serving the world's energy industry and other industrial applications

6.3

Million tons
of steel pipes

Annual
manufacturing
capacity

3.7 million tons
of seamless pipes

2.6 million tons
of welded pipes



OCTG



Premium
Connections



Offshore Line
Pipe



Hydrocarbon
Processing



Power
Generation



Industrial &
Automotive

16

Countries

Manufacturing
facilities

5

R&D
Centers

Worldwide

THOR™115



Demanding projects require high performing materials.
That's why Tenaris worked on the development and qualification of
a new ferritic steel product.



Born for Power Generation:

- Creep rupture strength equal and even better than T/P91
- Improved steam oxidation resistance when compared to T91/T92
- **Easy to manufacture**, put in service and weld

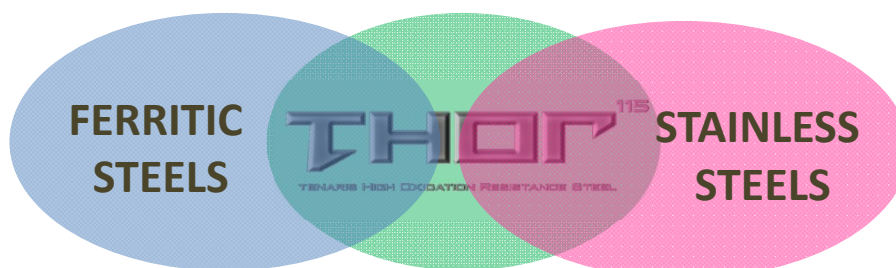
Nominal Cr content: 11%

THOR™115 Opportunities



Chemistry identified as suitable for Refining corrosive environment

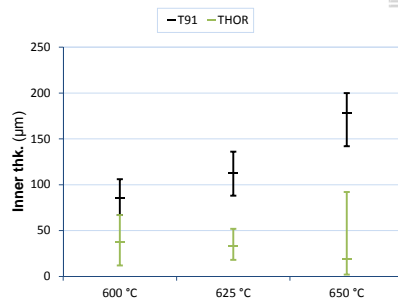
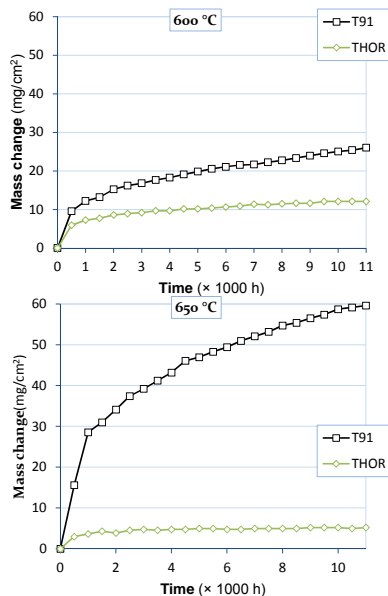
- Chemical composition in the middle of Ferritic and Stainless families
- Identification of fields of application of those materials
- Study of THOR behavior to understand strengths and weaknesses coming from Ferritic Steels and Stainless Steels



Test Plan Definition

- Loops identification
- Corrosion mechanisms
- Parameters identification
- Laboratory conditions definition
- Test equipment for dedicated trials
- Benchmarking with materials used in the identified service conditions

Steam Oxidation

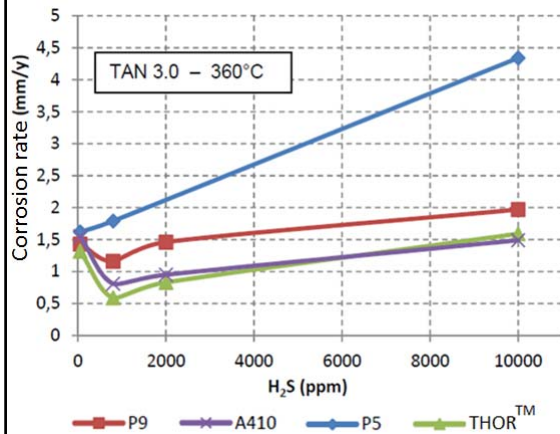


- Tests performed at Oak Ridge National Lab
- 11000 hours reached
- Benchmarking with T91



- THOR shows significantly better resistance vs P91 in terms of mass loss and oxide growth
- Mass change and scale thickness similar at 600°C and 650°C for THOR

High temperature sulfidation and naphthenic attack

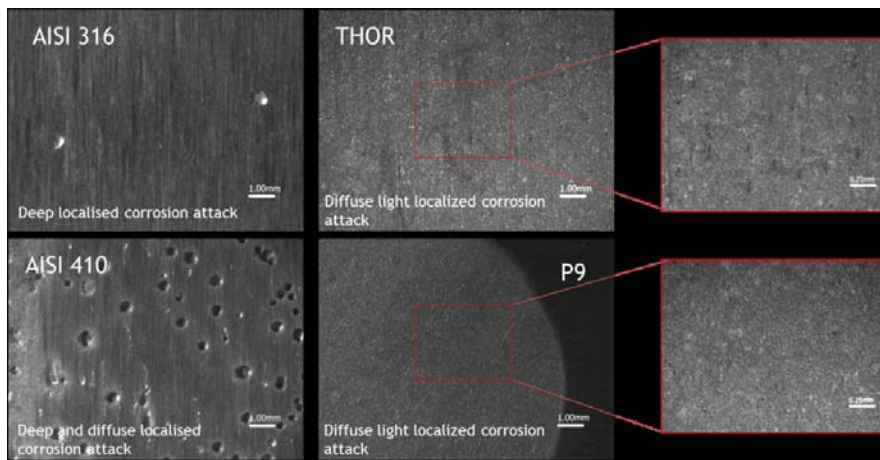


- Pilot Plant in Venezia Teconologie that allows to simulate operative conditions without fluid degradation
- Comparative tests
- Tests in naphthenic and sulphidic environment typical of refinery streams before desulphuration
- Expected trends confirmed after weight loss analysis at different H₂S contents

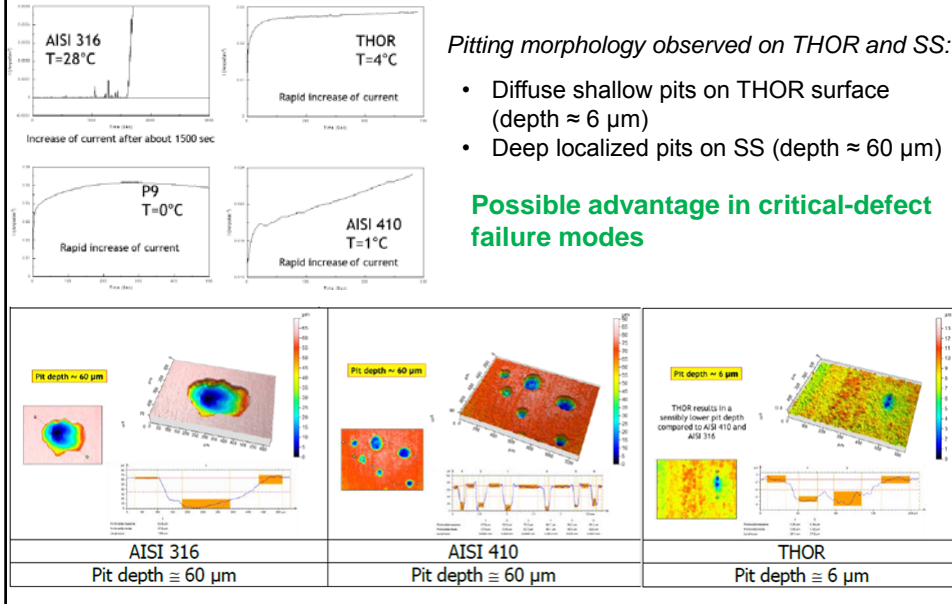
THOR SHOWS BETTER RESISTANCE vs P9, COMPARABLE TO A410

Chlorides – preliminary results

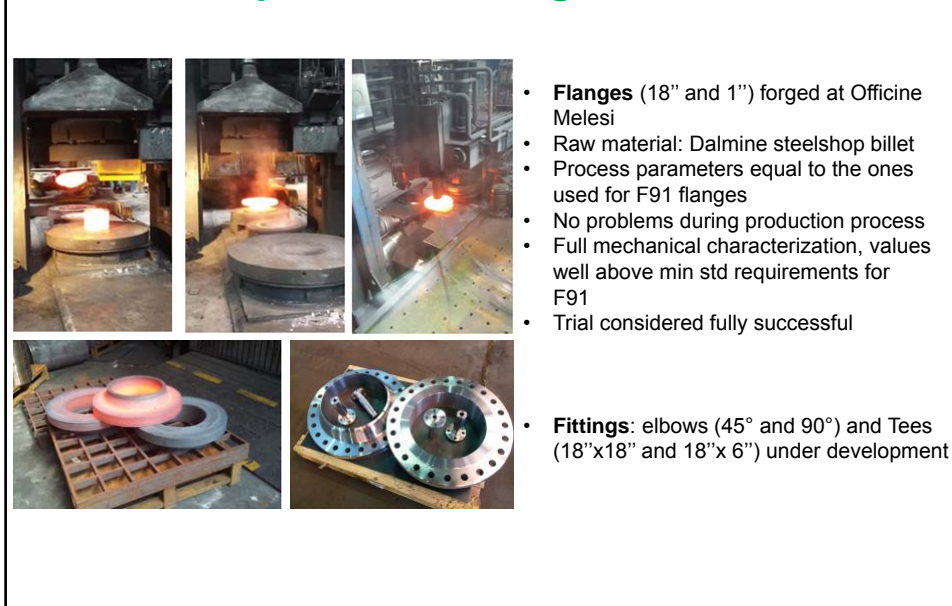
- Test Parameters as per ASTM G150
- Critical Temperatures for Pitting and Crevice
- Pitting morphology observation



Chlorides – preliminary results



Line components - Flanges



THOR™115 in International Standards



- *December 2014:* **Patent** published
- *April 2016:* **VdTÜV** approval received (Materialsheet 580, March 2016)
- *August 2016:* **ASME** code will be presented (ASME insertion expected by End 2016)
- *May 2016:* **ASTM** will be presented (A213, A335 will follow)

Next steps



- New tests to reinforce available data
- Identification of new testing conditions
- Test Loops
- Chemistry fine-tuning for special application

Any advice on test condition or test equipment is more than welcome!!!



Tubular Technologies. Innovative Services.



Thanks for your attention

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Product Engineer
Hydrocarbon Process Industry

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Appendix 9

CUI: utilization of Bayesian networks for prediction of location

(J. Rajek)



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TÜV
AUSTRIA

CUI
Corrosion under Insulation
& Bayesian
networks

J. Rajek
S. Jain

Mei 16

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TÜV
AUSTRIA

Motivation + Status

- ✓ CUI is a known threat to the process industry
- ✓ The problem areas are “out of sight”
- ✓ Mitigation is time consuming and expensive

1. Reduce incidents due to CUI – increase process safety
2. Save money for mitigation – do it more clever

Status:

- Feasibility done
- Cooperation partner list in development (EFC?)
- Literature research
- Basic modelling in progress

2

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Bayes' theorem

- ✓ Bayes' theorem describes the probability of an event, based on conditions that might be related to the event

$$p(B/E) = \frac{p(E/B) \cdot p(B)}{p(E)}$$

We want to know

We know, have data,
our hypothesis

- ✓ In other words...

Initial Belief $p(B)$ plus new Evidence = new and improved belief $p(B/E)$



Bayes' theorem

- ✓ Bayesian networks can...

- Make predictions under uncertainty
- Handle unknown data via probabilistic data
- Handle multiple inputs: Measurements, statistics, judgements

- ✓ Bayesian networks allow...

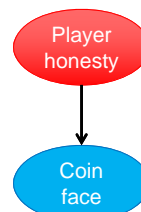
- Using distributions if hard data are missing
- Learning from small amount of information
- To find the most probable cause - and fix it



Example – coin flip



- ✓ I propose to flip a coin
 - ✓ You believe by 10%, that I am cheating = use double-headed coin
 - ✓ Evidence: On each flip “head” shows up
- How does this influence your believe that I cheat?



- ✓ We know:
 - Probability that I am a liar = 10% (assumption, hypothesis)
 - If liar: Probability for “head” = 100% (model, facts)
 - If honest: Probability for “head” = 50% (model, facts)
 - Flip coin: Check if head or tail (test, evidence)

$$p(\text{Liar}/\text{Head}) = \frac{p(\text{Head}/\text{Liar}) \cdot p(\text{Liar})}{p(\text{Head})}$$



You want to know this
=unknown



You use assumption and evidence
=known/assumed



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Example – coin flip



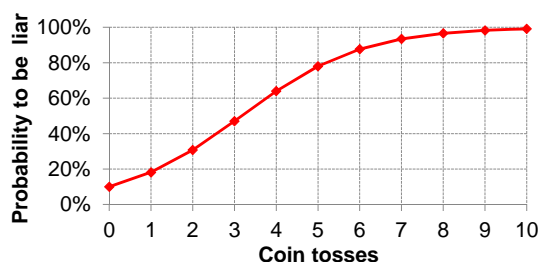
Initial assumption: $p(\text{Liar}) = 10\%$

1. test: $p(\text{Liar if head shows up}) = 18\% \rightarrow$ update assumption

$$p(\text{Liar}/\text{Head}) = \frac{p(H/L) \cdot p(L)}{p(H/L) \cdot P(L) + p(H/noL) \cdot P(noL)} = \frac{100\% \cdot 10\%}{100\% \cdot 10\% + 50\% \cdot 90\%} = 18\%$$

2. test + evidence from 1. test: $p(\text{Liar if head shows up}) = 31\%$

$$p(\text{Liar}/\text{Head}) = \frac{p(H/L) \cdot p(L)}{p(H/L) \cdot P(L) + p(H/noL) \cdot P(noL)} = \frac{100\% \cdot 18\%}{100\% \cdot 18\% + 50\% \cdot 82\%} = 31\%$$



- ✓ Falsify hypothesis
- ✓ Increase confidence

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Example - accident

- ✓ You are in the office and think about driving home
- ✓ Data you have:
 - You know the time
 - You see if it is cloudy or raining
 - You could hear a siren on the highway
- ✓ Should you drive home now?
- ✓ What is the chance for a traffic jam?

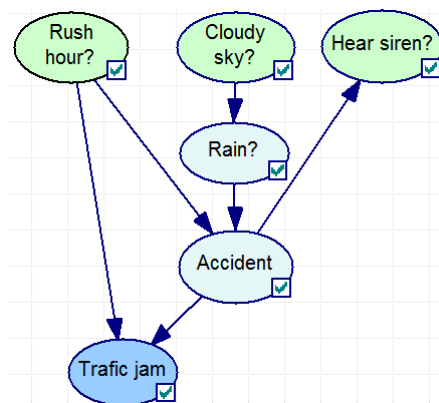
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Example - accident

- ✓ Possible Bayesian network



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Example - accident



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- ✓ Prior data, example
- ✓ Sources: Statistics, experience, calculations, measurements

	Cloudy sky?
clear	90%
cloudy	10%

	cloudy	clear
road wet	50%	5%
road dry	50%	95%

		yes		no	
Rush hour	Rain	road wet	road dry	road wet	road dry
Accident happened		70%	40%	60%	10%
No accident		30%	60%	40%	90%

Vertical sum = 100%

0

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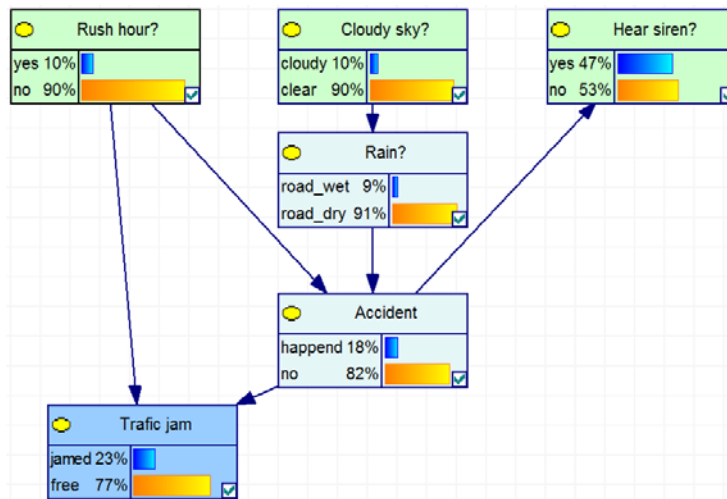


Example - accident



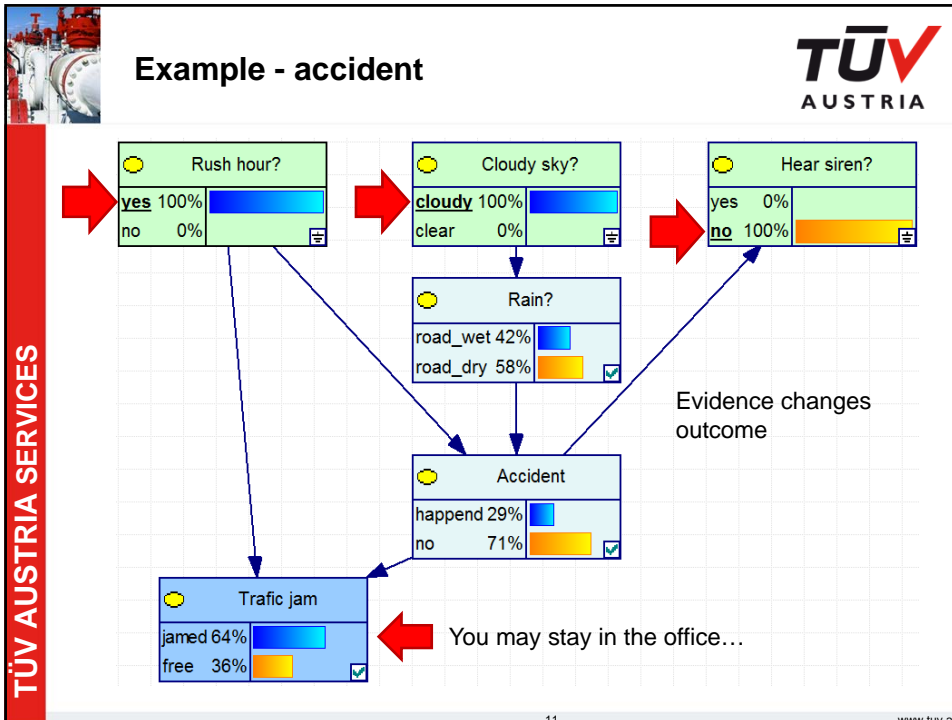
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- ✓ Prior data, no evidence



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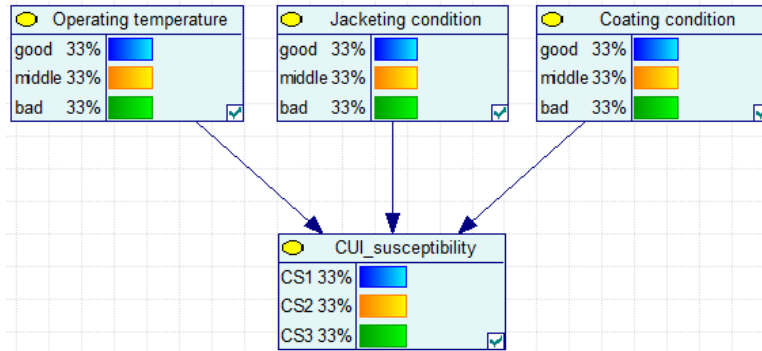
-
- CUI score tables as Bayesian network**
- TÜV AUSTRIA SERVICES**
- TÜV AUSTRIA**
- ✓ **Mimic CUI score tables (example)**
- Operating temperature influence on CUI?
 - Good (no influence): $T < -5C$ or $T > 175C$
 - Middle: $-5C < T < 50C$ or $110C < T < 175C$
 - Bad: $50C < T < 110C$
 - Jacketing condition
 - Good, middle, bad condition
 - Coating condition
 - Good, middle, bad condition



CUI score Bayesian network, start

✓ Even prior data „no knowledge”

- All influencing parameters evenly distributed
- Model predicts even CUI susceptibility



13

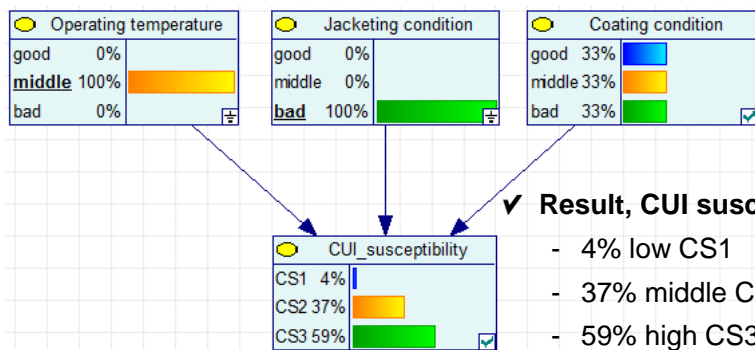
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CUI score Bayesian network, start

✓ Influence of prior data

- Operating temperature = 40C → middle
- Jacketing: Open on many places → bad
- Coating: ***Unknown***



✓ Result, CUI susceptibility

- 4% low CS1
- 37% middle CS2
- 59% high CS3

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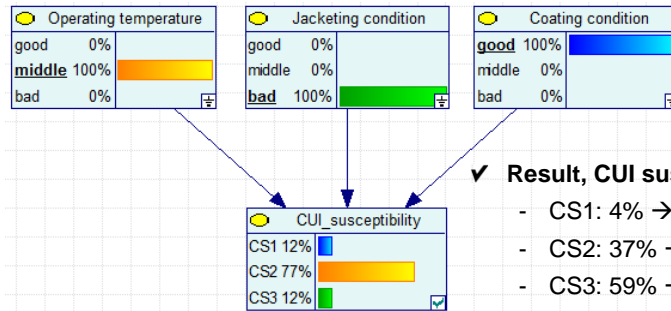
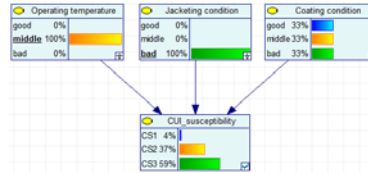
CUI score Bayesian network, start



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✓ Add evidence data about coating:

- Coating: Good condition
→ good



✓ Result, CUI susceptibility shift

- CS1: 4% → 12%
- CS2: 37% → 77%
- CS3: 59% → 12%

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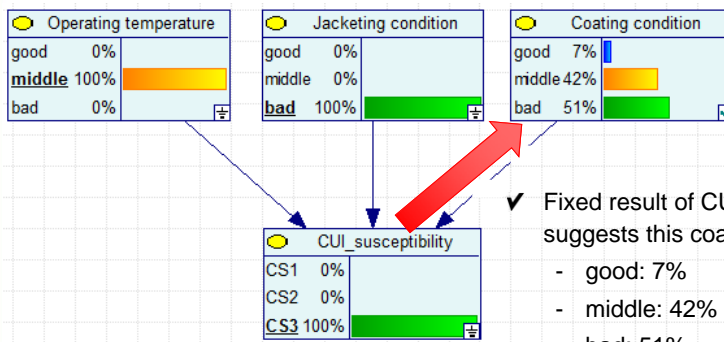
CUI score Bayesian network, start



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✓ Bring in evidence, inferring

- We found CUI condition 3 = "proof"
- What can we learn about our coating condition?



✓ Fixed result of CUI susceptibility suggests this coating condition:

- good: 7%
- middle: 42%
- bad: 51%

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Expectations



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- ✓ **Better utilization of existing data → learn**
 - Analyse data and find hidden patterns
 - Identify missing data-types
- ✓ **Better link of data**
 - Conditions, service and CUI over physical location
 - Confidence rating for prediction: Enough NDT data?
- ✓ **Outcome**
 - More precise prediction of severity AND location of CUI
 - More confidence in CUI inspection and testing program
 - Increased process safety
 - Optimised effort (safe money)

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Next steps



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1. **Collection of ALL data possible**
 - Finished CUI projects
 - Published data
2. **Test & improve model**
 - Compare model results with finished projects
 - Compare model results with score table approach
3. **Discuss results with industry partners, EEPC gets:**
 - Analysis of existing data
 - Proposal how to improve CUI assessments – if possible
 - Update score tables – if necessary
4. **Pilot project?**

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Thank you for your attention!



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Joachim Rajek
joachim.rajek@tuv.at

Software: GeNle
Decision Systems Laboratory
University of Pittsburgh



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Backup



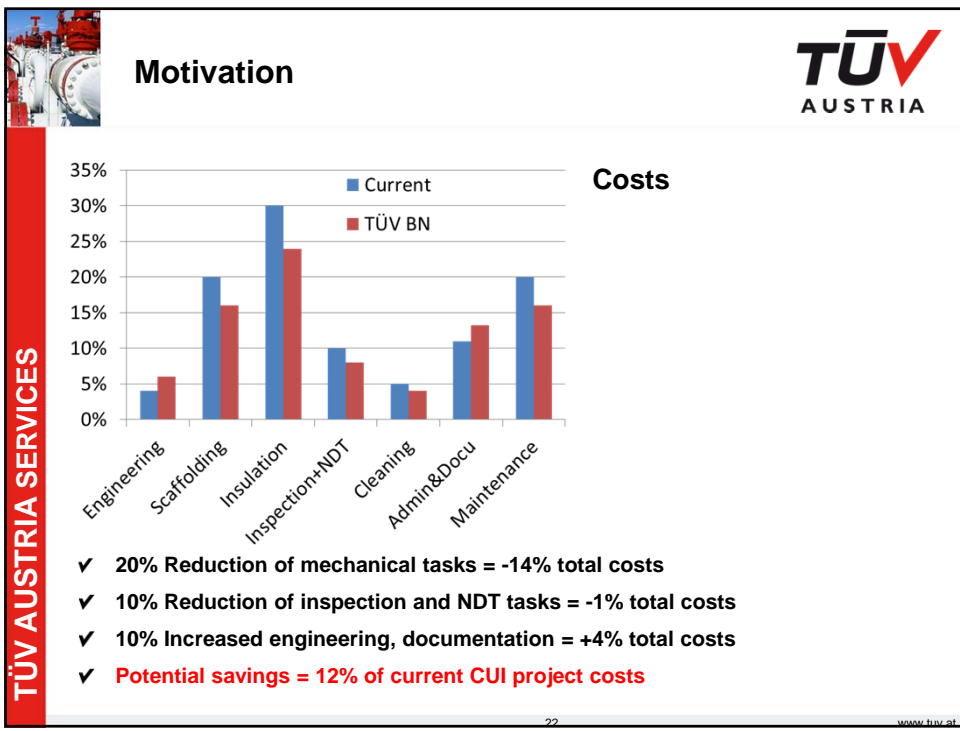
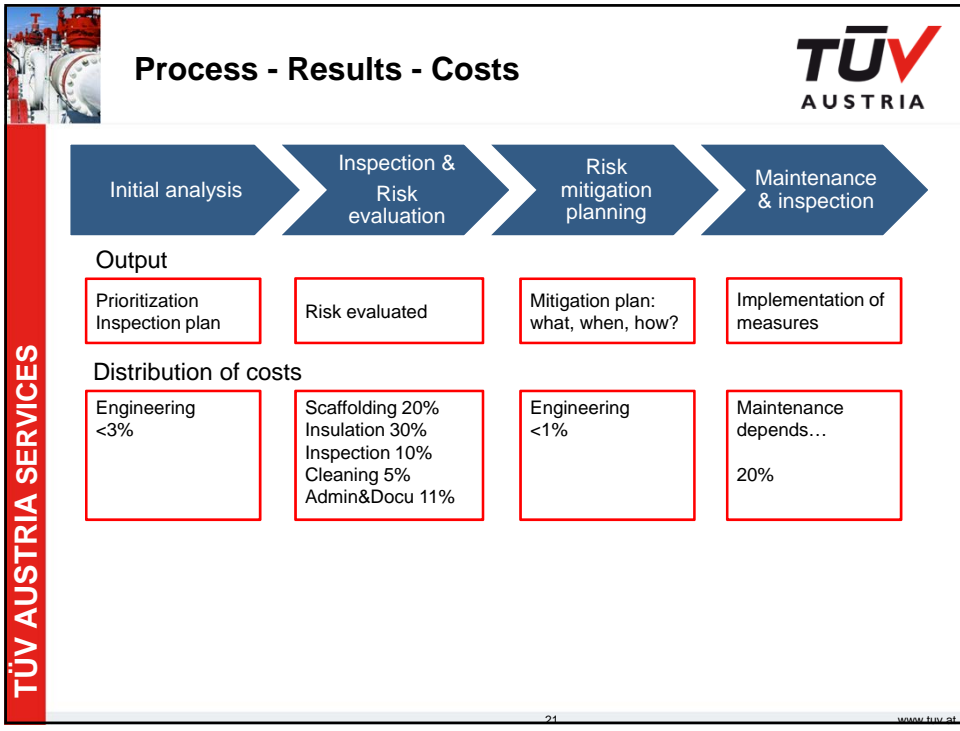
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✓ Backup...

May 2011


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The team




TÜV AUSTRIA SERVICES



Joachim Rajek
 Current: **TÜV Austria Services, Industry & Energy**
 Previous:
 OMV R&M, Head of Inspection & Integrity, Schwechat refinery

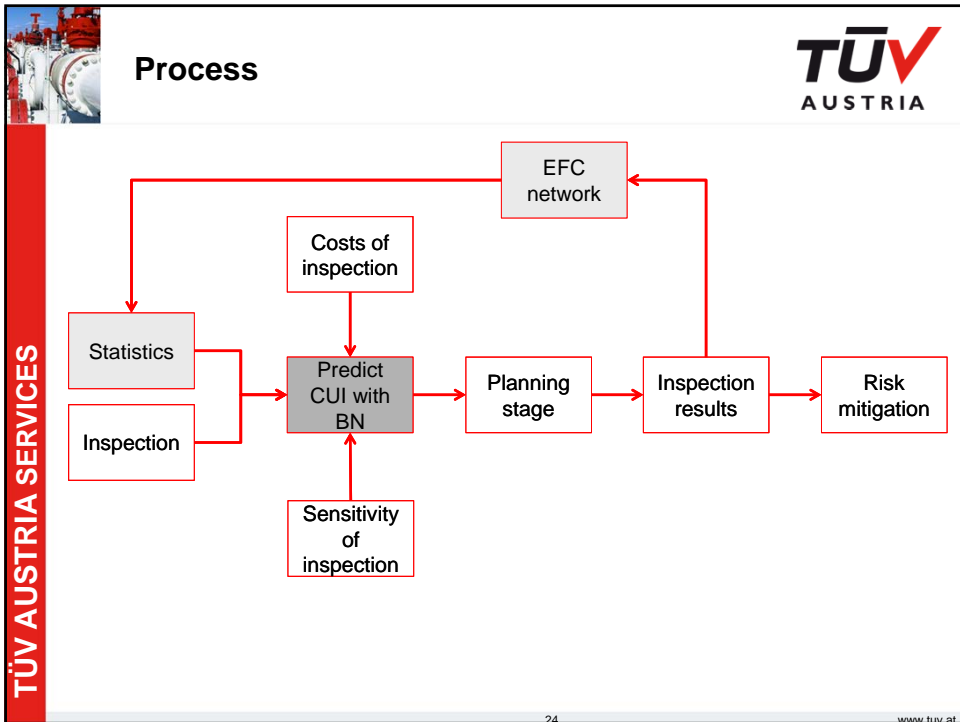
 PhD, mechanical engineering & economics



Swati Jain
 Current : **Senior Data Scientist/Independent consultant**
 Previous :
 DNV GL, oil&gas research engineer

 PhD, chemical engineering

May 16
23
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Bayes' theorem, cancer example

- ✓ 1% of population has this illness
 - ✓ Test sensitivity=99% → 99% of people tested are identified correctly
- What is the probability to have illness given the test result is positive?

	illness	No illness
Test +	99%	1%
Test -	1%	99%

P(ill)=1% Population has illness

P(T+)=1,98% of all test results are positive

$$p(C/T+) = \frac{p(T+/C) \cdot p(C)}{p(T+)} = \frac{99\% \cdot 1\%}{1,98\%} = 50\%$$

→ Test says "50% probability to have illness if test is positive"



Bayes' theorem, cancer example

✓ Initial data

P(ill)=1% Population has illness
 P(T+)=1,98% of all test results are positive

✓ After first test

P(ill)=50% According test
 P(T+)=50% of all test results are positive

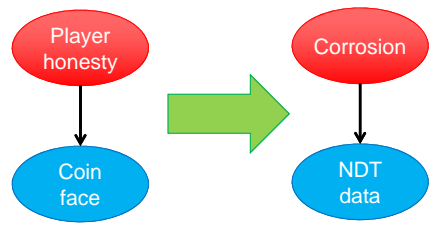
$$p(C/T+) = \frac{p(T+/C) \cdot p(C)}{p(T+)} = \frac{99\% \cdot 50\%}{50\%} = 99\%$$

→ 2nd test says "99% probability to have cancer if test is positive"

- ✓ Power of Bayesian theorem → Reduce uncertainty by using evidence and updating prior probability



Power of Bayesian networks



- ✓ Conditional dependence between variables through Bayes' theorem
- ✓ Bayesian networks can learn from small amount of information
- ✓ Bayesian networks make prediction under uncertainty
- ✓ Bayesian networks can help determine the most probable cause and fix it



Bayes' theorem, illness example

- ✓ 1% of population has this illness
- ✓ Test sensitivity=99% → 99% of people tested are identified correctly

	illness	No illness
Test positive	99%	1%

$P(\text{ill})=1\%$ Population has illness

- What is the probability to have this illness given the test result is positive?
- Test result: "50% probability to have illness if test is positive"
- Are we happy? No!

Mathematical reason:

Though the test is very precise (99%), the initial probability to be ill is only 1%
 → second test, we need data



Bayes' theorem, cancer example



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✓ First test

- Initial data: $P(\text{ill})=1\%$ Population has illness
- First test result: $P(\text{ill}/1. \text{ test positive})= 50\%$

✓ Second test

- Initial data: $P(\text{ill})=50\%$ = result of first test
- Second test result: $P(\text{ill}/2. \text{ test positive})= 99\%$

- ✓ **Power of Bayesian theorem** → Reduce uncertainty by using evidence and updating prior probability

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CUI score tables as Bayesian network



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✓ Mimic CUI score tables (example)

- Operating temperature influence on CUI?
 - Good (no influence), $T < -5C$ or $T > 175C$
 - Middle $-5C < T < 50C$ or $110C < T < 175C$
 - Bad $50C < T < 110C$
- Jacketing condition
 - Good, middle, bad condition
- Coating condition
 - Good, middle, bad condition

30

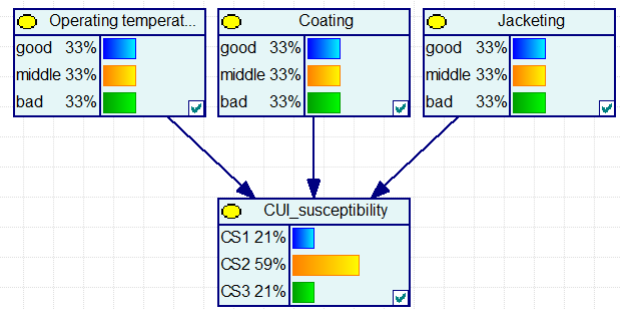
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CUI score Bayesian network, start

✓ Even prior data „no knowledge”

- All influencing parameters evenly distributed
- Model predicts 59% CS2, 21% CS1 or CS3



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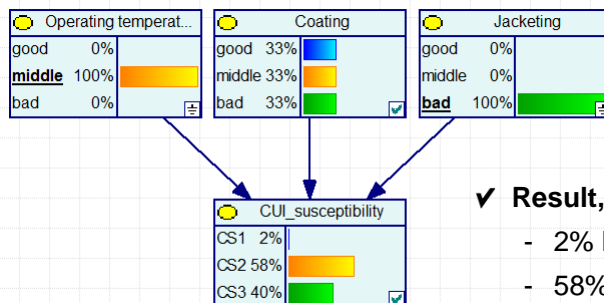
www.tuv.at



CUI score Bayesian network, start

✓ Prior data

- Operating temperature = 40C → middle
- Jacketing: Open on many places → bad
- Coating: *Unknown*



✓ Result, CUI susceptibility

- 2% low
- 58% middle
- 40% high

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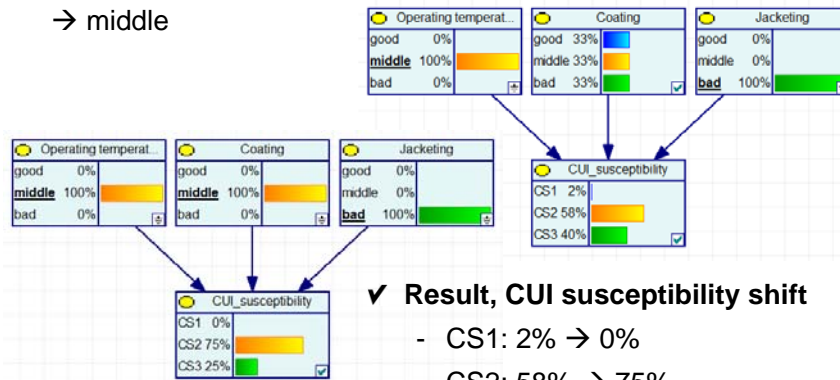
www.tuv.at



CUI score Bayesian network, start

✓ Add evidence data about coating:

- Coating: partly damaged
→ middle



✓ Result, CUI susceptibility shift

- CS1: 2% → 0%
- CS2: 58% → 75%
- CS3: 40% → 25%

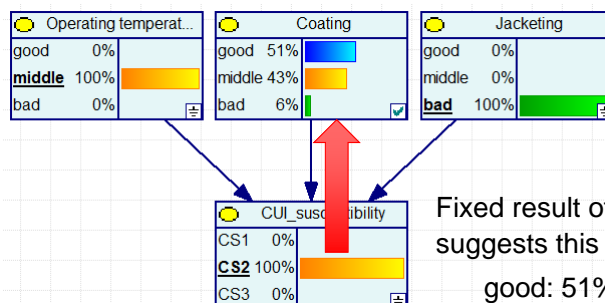


CUI score Bayesian network, start

✓ Bring in evidence, inferring

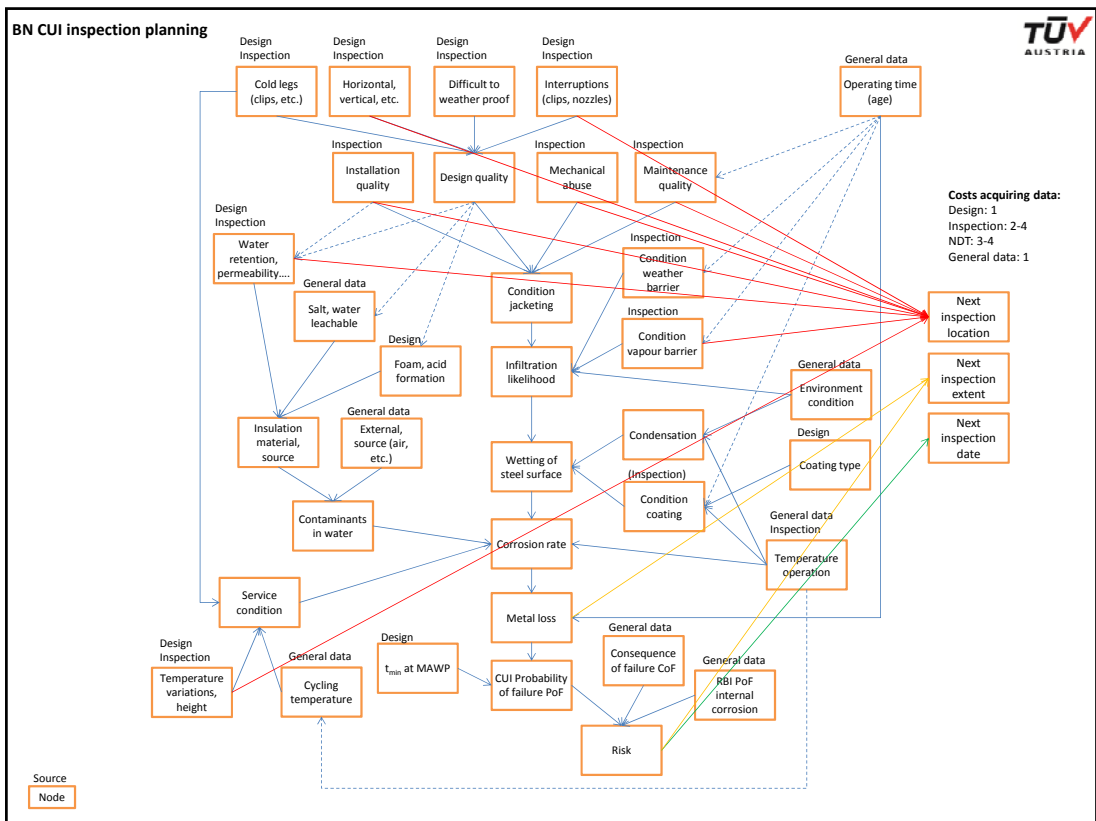
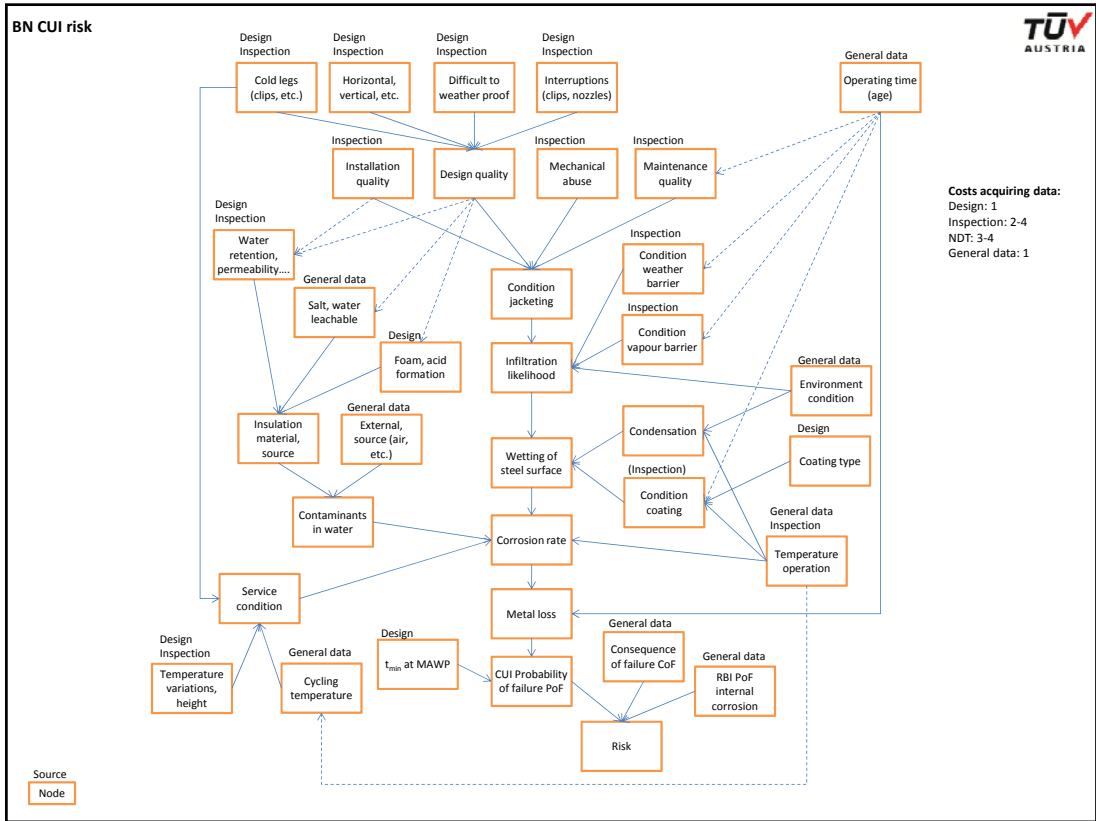
- We found CUI condition 2 "proof"

→ What can we learn about our coating condition?



Fixed result of CUI susceptibility suggests this coating condition:

- good: 51%
- middle: 43%
- bad: 6%



Appendix 10

World Class Maintenance project on CUI

(R. de Heus)



Project "Corrosion under Insulation" (CUI)

World Class Maintenance
Presentation for EFC meeting April 26



Goals

Short term Goal CUI cooperation Dutch Industry


To create a proven self assessment tool for Dutch Industry based on EFC guideline.

Definite Goal CUI Dutch Industry

To Implement a "Best Practice" in the Industry to minimize the number of incidents with root cause "corrosion under insulation" .

To prevent reputational damage in the industry, unsafe situations and to decrease the total cost of ownership for CUI projects.







Development self assessment tool

Done

Self assessment is piloted at 4 plants and available for dutch speaking others

To do


- Translate assessment to english language and make it web enabled.
- Roll out to others via web

Next steps CUI

Proposed follower projects

- Development communication and trainingstools for Contractors/operators and technicians to prevent damage to insulation and increase awareness on shop floor.
- Develop standard, modulair CUI strategy format based on EFC in order to manage CUI more uniform and easy.
- Set up of network and advisory board to develop adequate monitoring techniques for CUI.
- Develop and implement a pragmatic RBI tool to predict high risk situations and for early detection and timely resource allocation
- Exchange feasibility and application of Thermal Spray Alumina for greenfield, brownfield.





Actual Situation and question to EFC

- Project on hold. Participating companies are not sure about next steps. WCM is looking around to involve more experienced participating parties in order to select best practices that will be broad accepted and to increase the rate of success. Uniform acceptance is a problem.
- The European Federation of Corrosion has produced the EFC guideline on CUI. What next steps on the guideline are foreseen at EFC?
- Can we, EFC & WCM, combine efforts to implement the guideline in dutch industry and/or can we transform the guideline to a NEN norm ?



www.worldclassmaintenance.com

Dit project is mede mogelijk gemaakt door een bijdrage uit het Europees Fonds voor Regionale Ontwikkeling in het kader van Op-Zuid.



Appendix 11

Failure cases

Material selection for corrosion protection

inside desalters

(S. Koller)

Desalter - Material and Corrosion

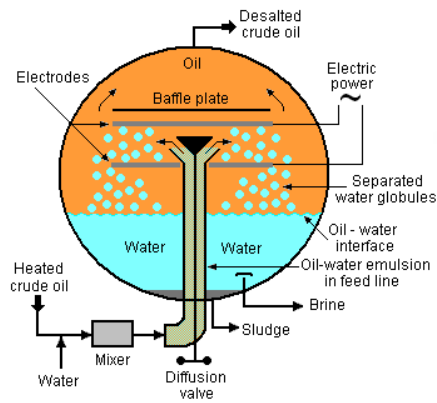


HOLBORN EUROPA RAFFINERIE
GMBH

Desalter



Electrostatic crude oil desalter



Cross-sectional view of Electrostatic crude oil desalter

Technical data

- Material: WStE47 (P460NH) (40 mm)
- Pmax: 50 bar
- T: 110 to 140 °C

Corrosion phenomena



Corrosion phenomena



Discussion of countermeasures

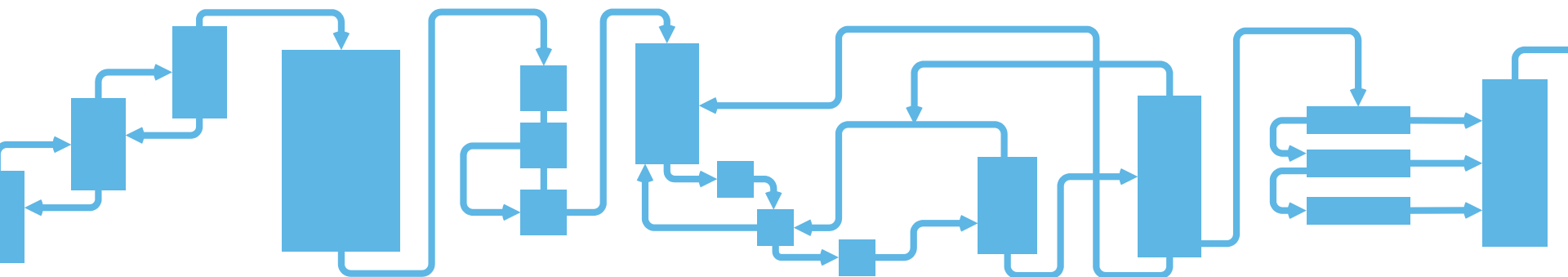


- Material selection
- Organic coatings/liners
- Metal plating/lining
- ...

Appendix 12

**Acceptable low level caustic concentration for
stainless steels to avoid SCC above 200°C**

(S. Shapcott)



Acceptable low level caustic concentrations (<0.5%) for stainless steels to avoid SCC above 200°C. Reflections on the new austenitic stainless steel caustic service chart in NACE SP0403-2015.

Stephen Shapcott

EFC WP15 - Spring 2016 - Paris

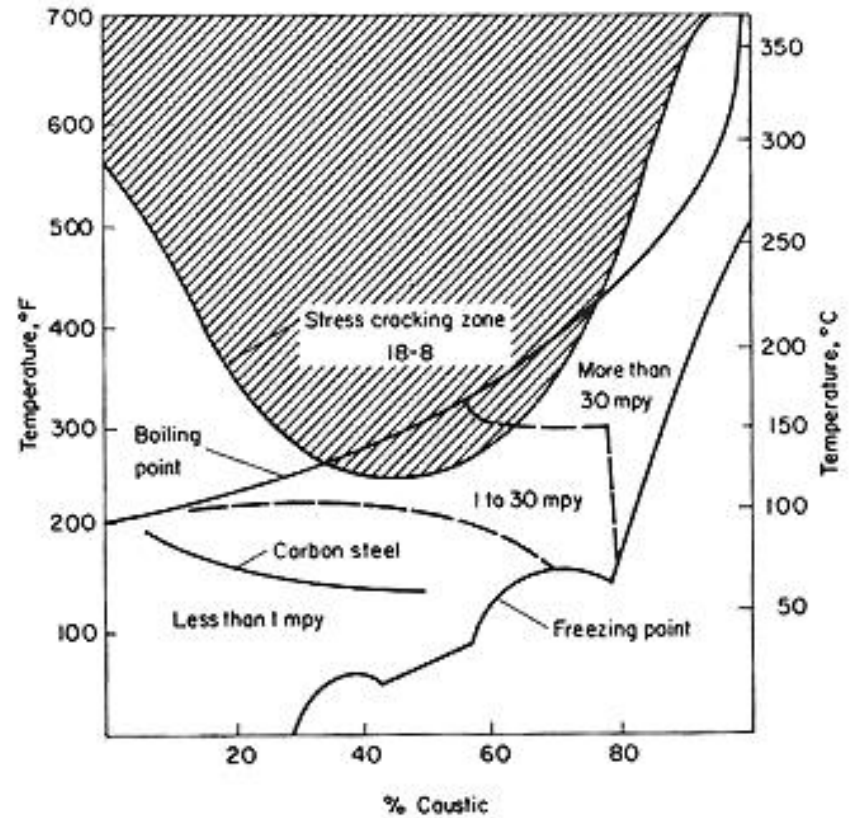


Johnson Matthey
Process Technologies

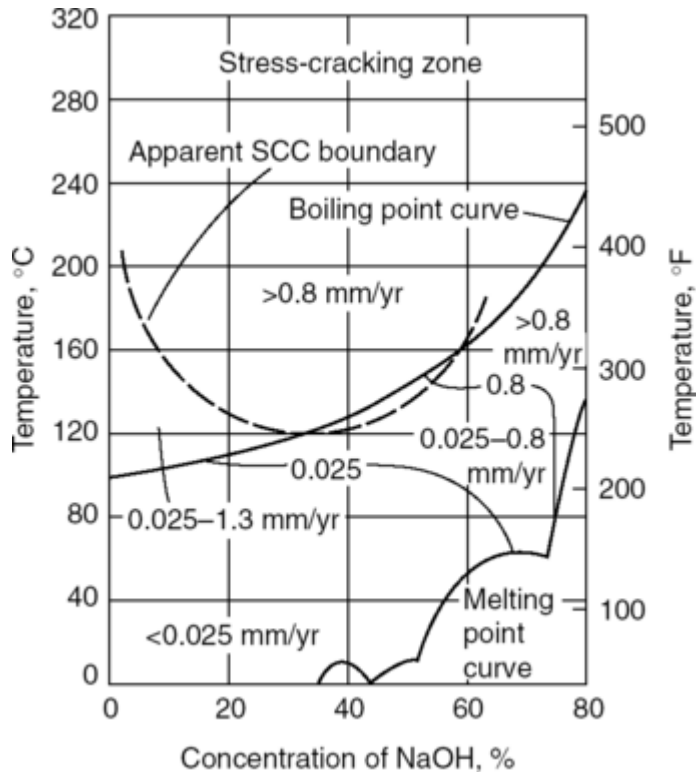
The beginning?

C P Dillon – 1986

Corrosion Control in the Chemical
Process Industries



Recent recommended practice



ASM Handbooks Online - 2016

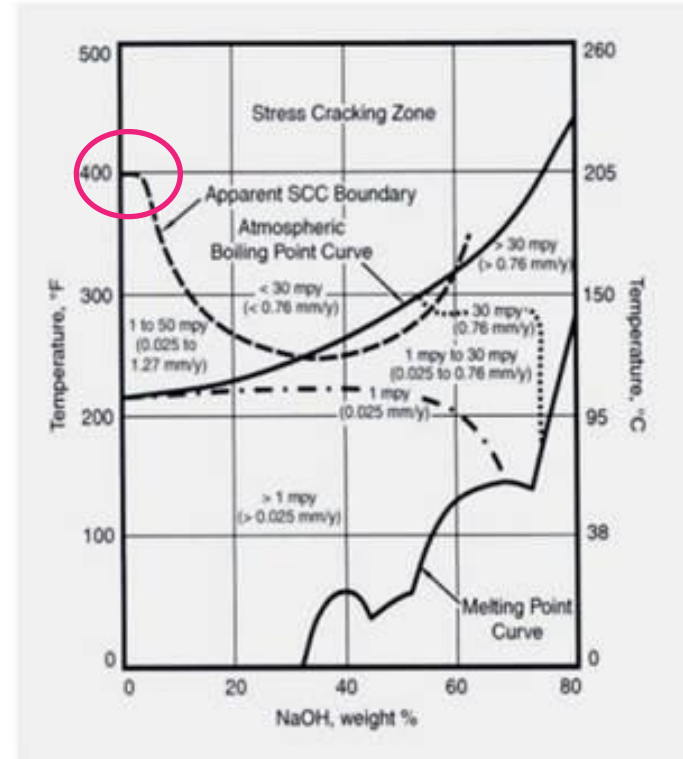


Figure 3. Caustic stress corrosion cracking susceptible regions of 300 series SS

MTI Bulletin - 2008



Current advice from NACE

NACE SP0403 – 2015

Certain technologies can recycle effluent water.

This effluent needs to be neutralised for use downstream, this is done with low level NaOH dosing.

Dosing leaves a slight residue of NaOH (1000-1500 ppm)

MTI curve is more conservative at very low concentrations, but less at higher concentrations

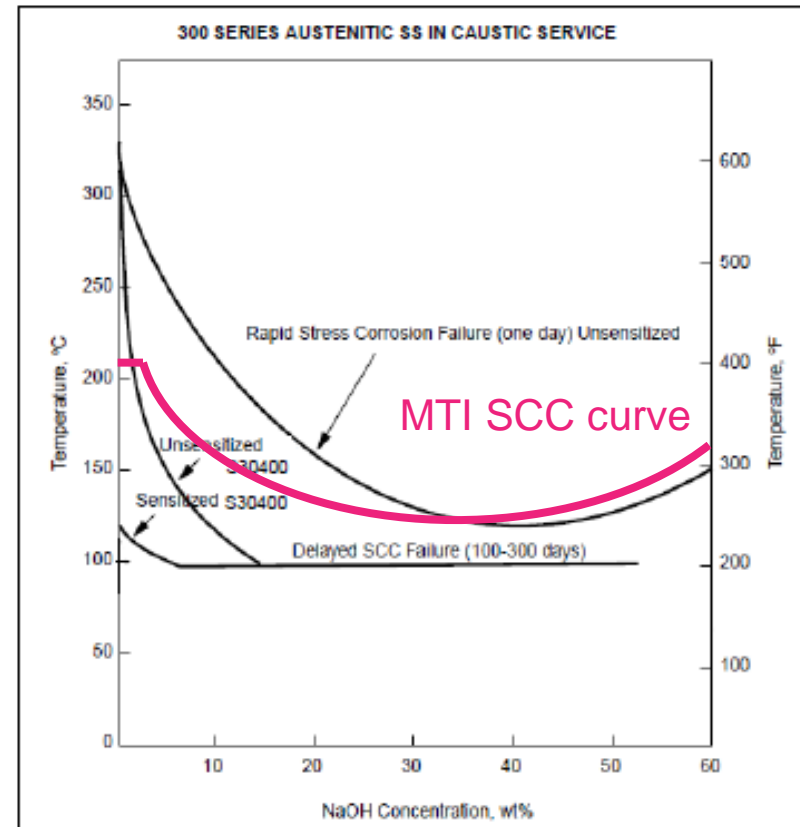
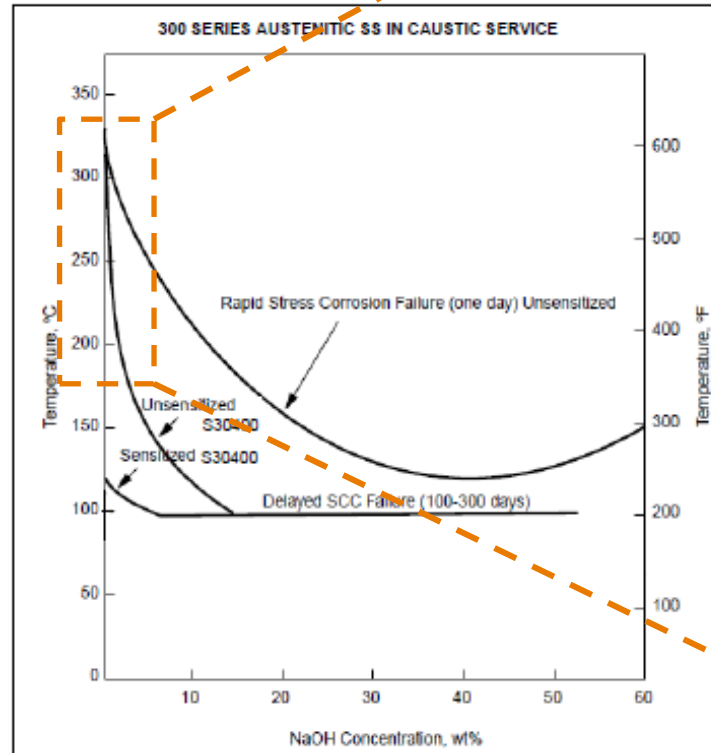
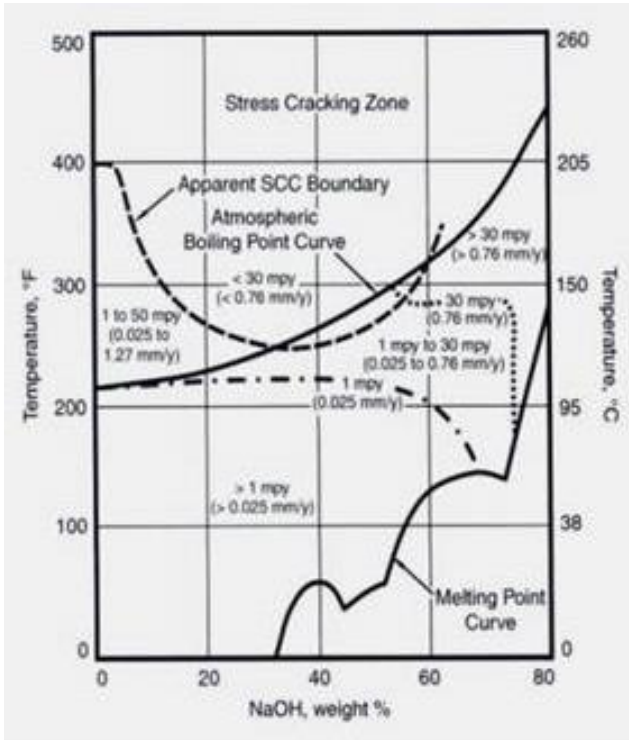


Figure 2: Caustic Service Chart for 300 Series Austenitic SS^{9,10}

Which to choose?

Given the proximity of 0.15% NaOH to the NACE line, and the conservatism of the MTI curve. Which is a prudent choice based on industry experience? Where does the NACE curve come from?



Duplex stainless steels for caustic service?

Questions for WP15

Which are the best duplex grades alloys to use?

Various sources indicate Mo may be detrimental to caustic corrosion performance. What are your experiences of this with low levels of caustic?

As NACE SP0403 recommends expensive nickel alloys once above the SCC curve for 300 series SS, how much improvement do duplex grades give?

Alloy vendors do not have data above boiling, and the literature is only qualitative.

Can excursions over an SCC boundary temperature initiate SCC?

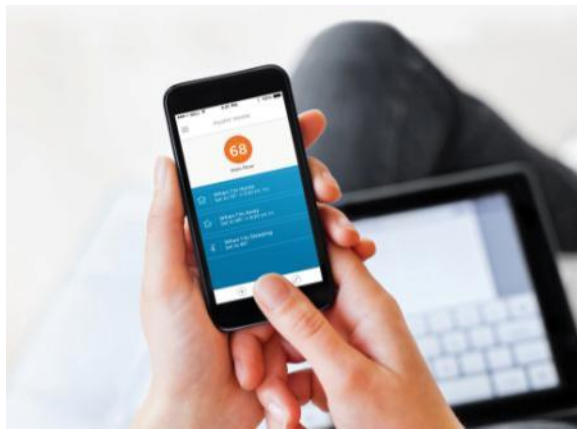
Thank you!



Appendix 13

Real time corrosion prediction and monitoring

(S. Kus)



Dr. Slawomir Kus
26-April-2016

REAL TIME CORROSION PREDICTION AND MONITORING

EFC Working Party 15 - Corrosion in Refinery Industry **Honeywell**

Agenda

- Traditional Methods of Corrosion Management
- Corrosion information flow – deficiencies in communication between Operation and Integrity/Corrosion Teams
- On-line, real time, corrosion management: Enhancements:
 - Step 1 – Monitoring
 - Step 2 – Prediction
- Architecture of real time monitoring/prediction integrated network
- Application example – Amine Unit
- Smarter Approach to Corrosion Management – on-line & real time
- Summary

Traditional Methods of Corrosion Management



Methods

- Inspection
- Coupons
- NDT techniques
- pH and other analytical parameters
- Corrosion is not quantified as it happens, it is found after it happens!

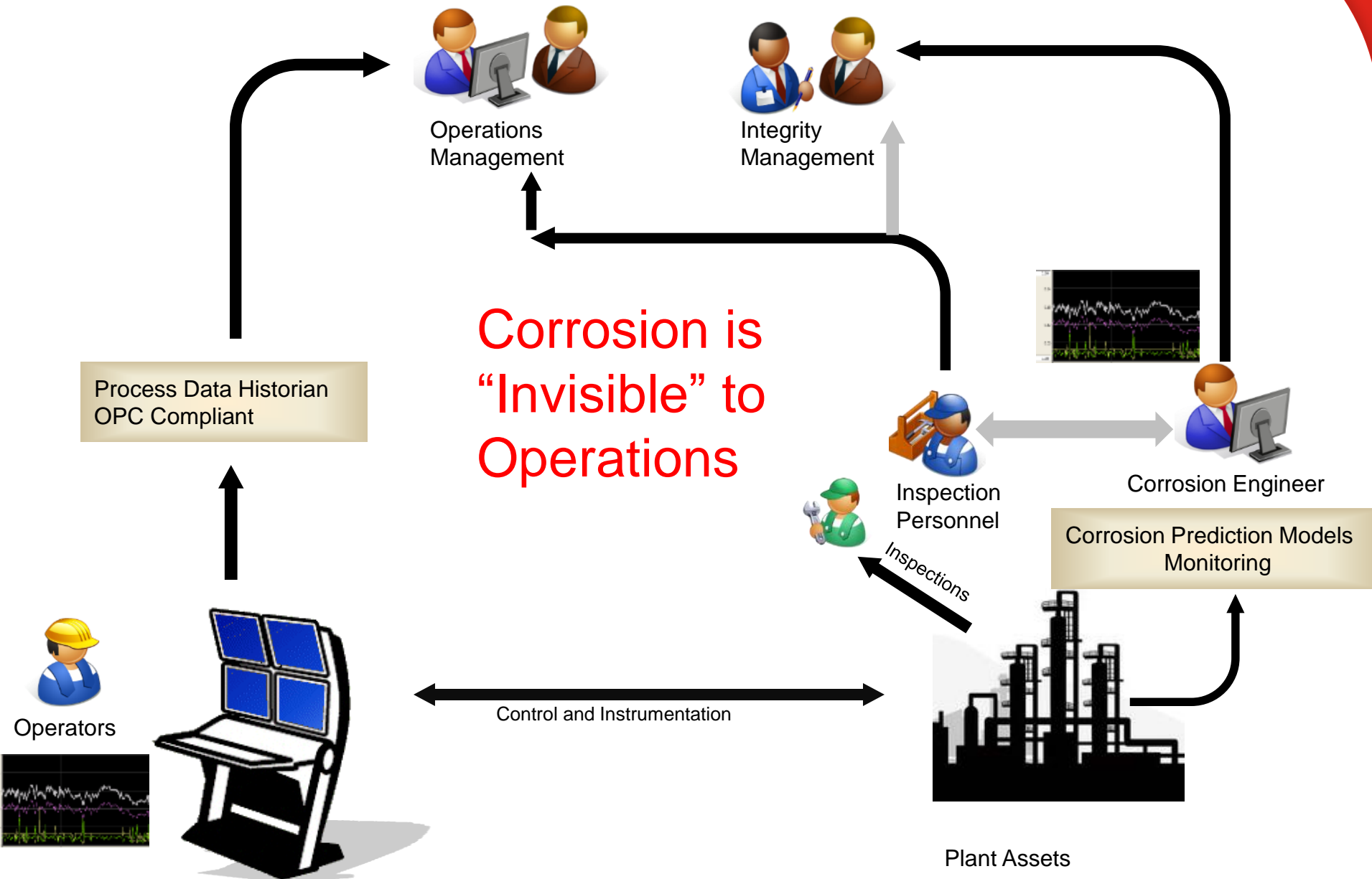


Limitations

- Reactive Response
- Historical based assessment
- Deviations and Peaks unknown
- Inspection Centric
- High Cost of Failure
- Inhibitor cost

Captures cumulative damage – reactive approach

Typical flow of corrosion information



On-line, real time, corrosion management

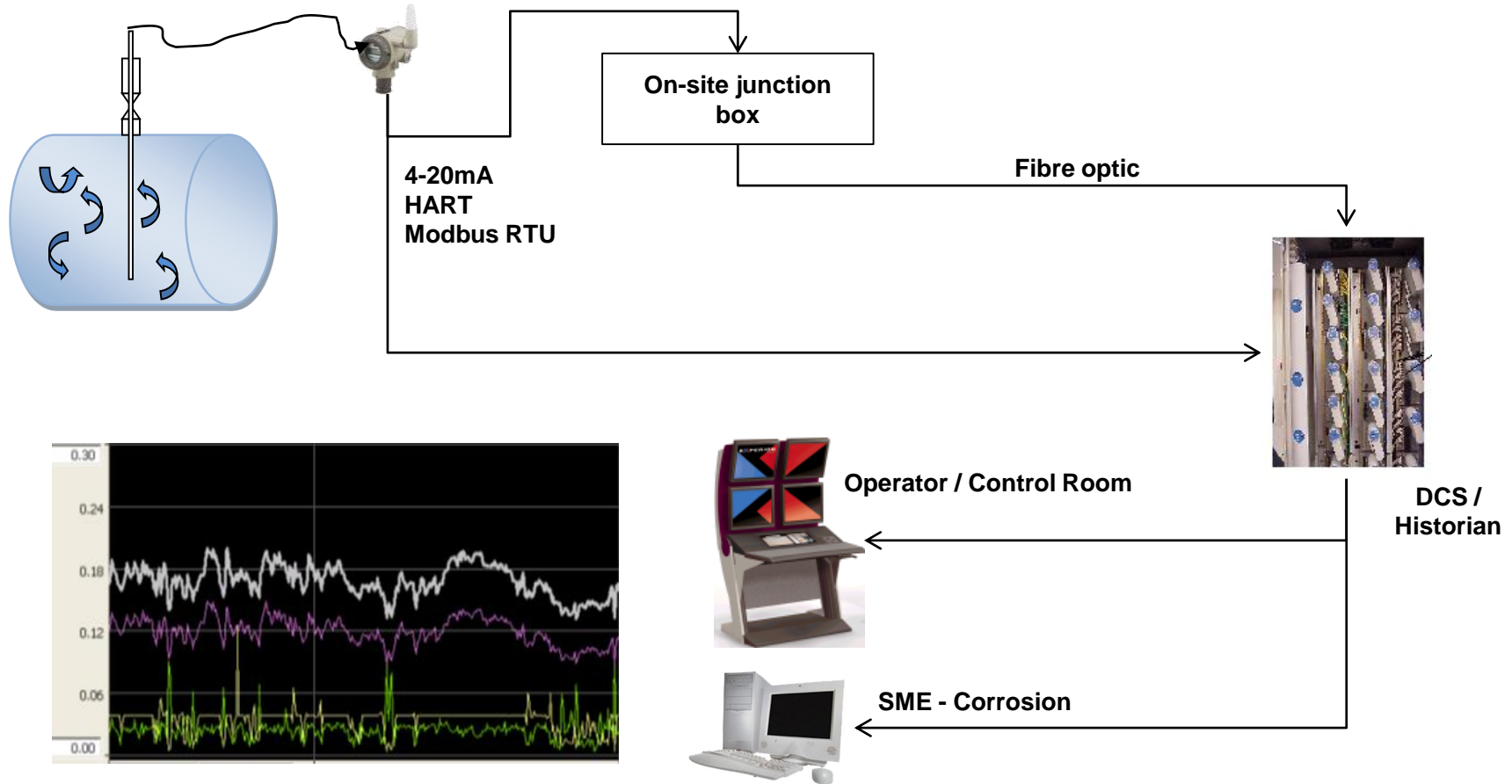
Evolutionary Enhancements – Step 1 - Monitoring



On-line, real time corrosion monitoring – typical setup

On-line ER, LPR, UT etc.

Data collection at intervals from minutes to hours



Corrosion visible in “dynamic” mode

Deficiencies of on-line, real time monitoring

- Very few monitoring locations
- Difficulties in proper selection of monitoring points and monitoring techniques
- Complicated integration with plant's DCS
- Complicated process-corrosion data analysis
- Operation does not see corrosion trends/alarms to be reliable
- Overall implementation and maintenance costs
-

On-line, real time, corrosion management

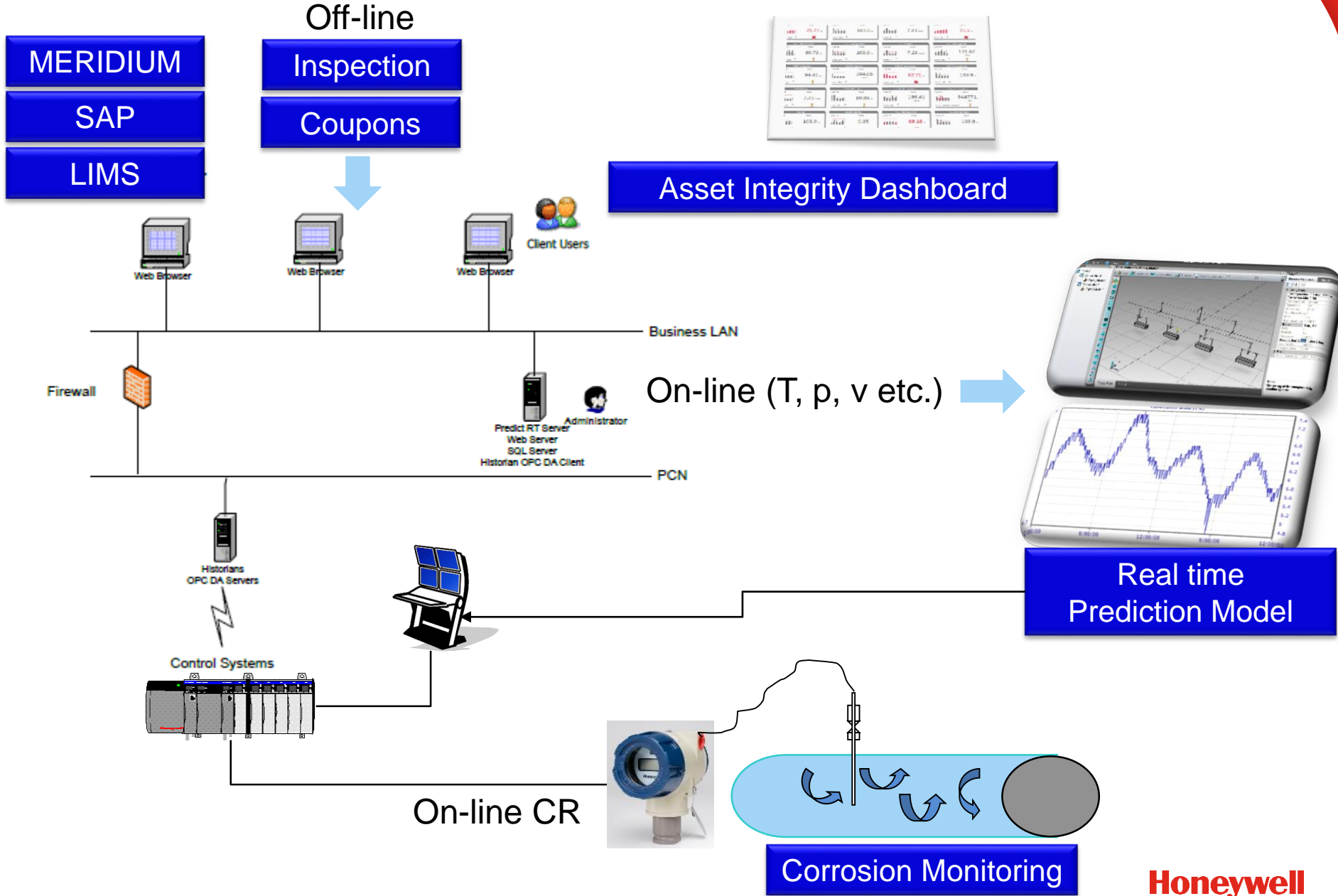
Evolutionary Enhancements – Step 2 - Prediction



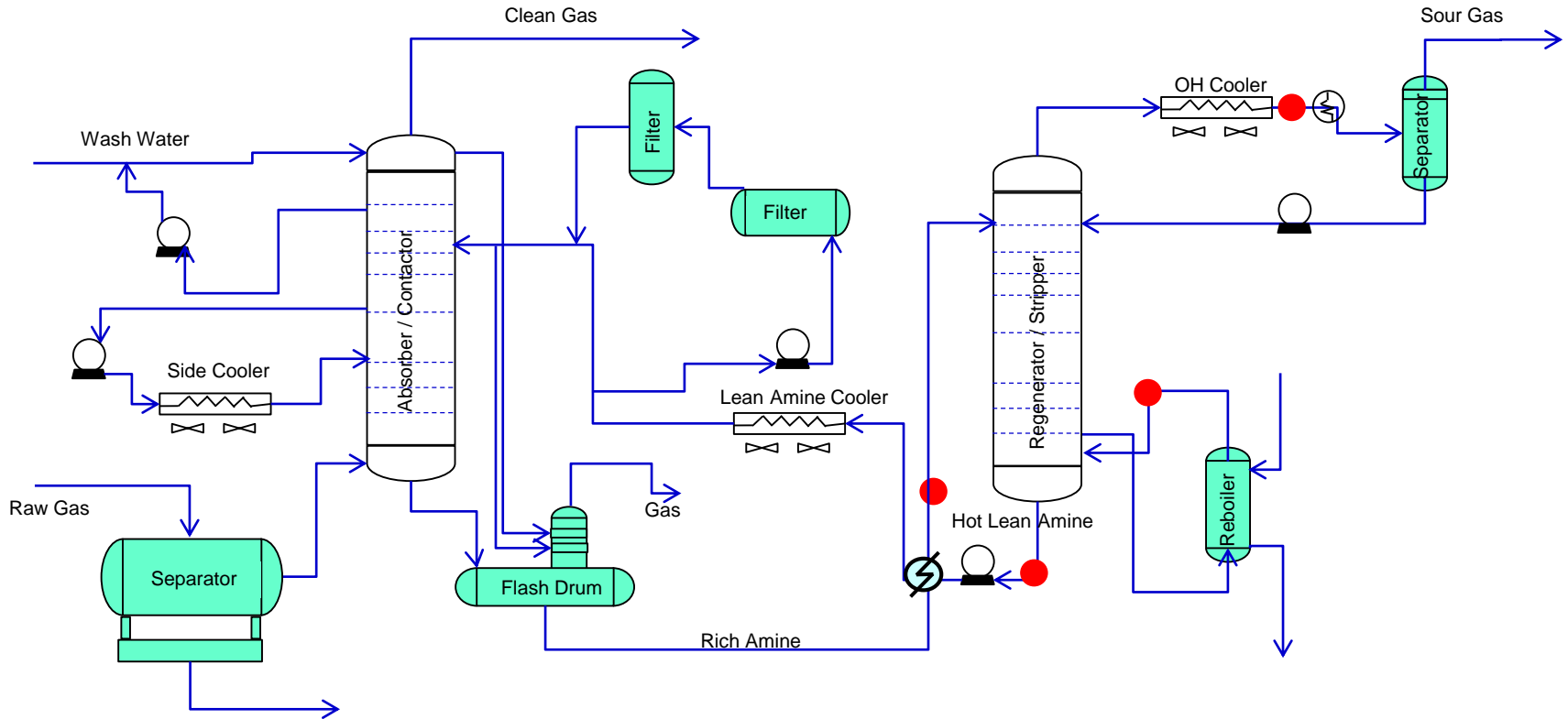
Why real-time corrosion prediction?

- Intrusive corrosion monitoring cannot be implemented everywhere
- It is important to know how current corrosion situation may/will be developed over time
- Standard corrosion models are based on static, fixed input data
- Not possible for all process configurations to be predicted at design stage
- Increasing throughput may shift corrosion beyond designed scenarios
- Integrity Operating Windows, if left “static”, may become irrelevant

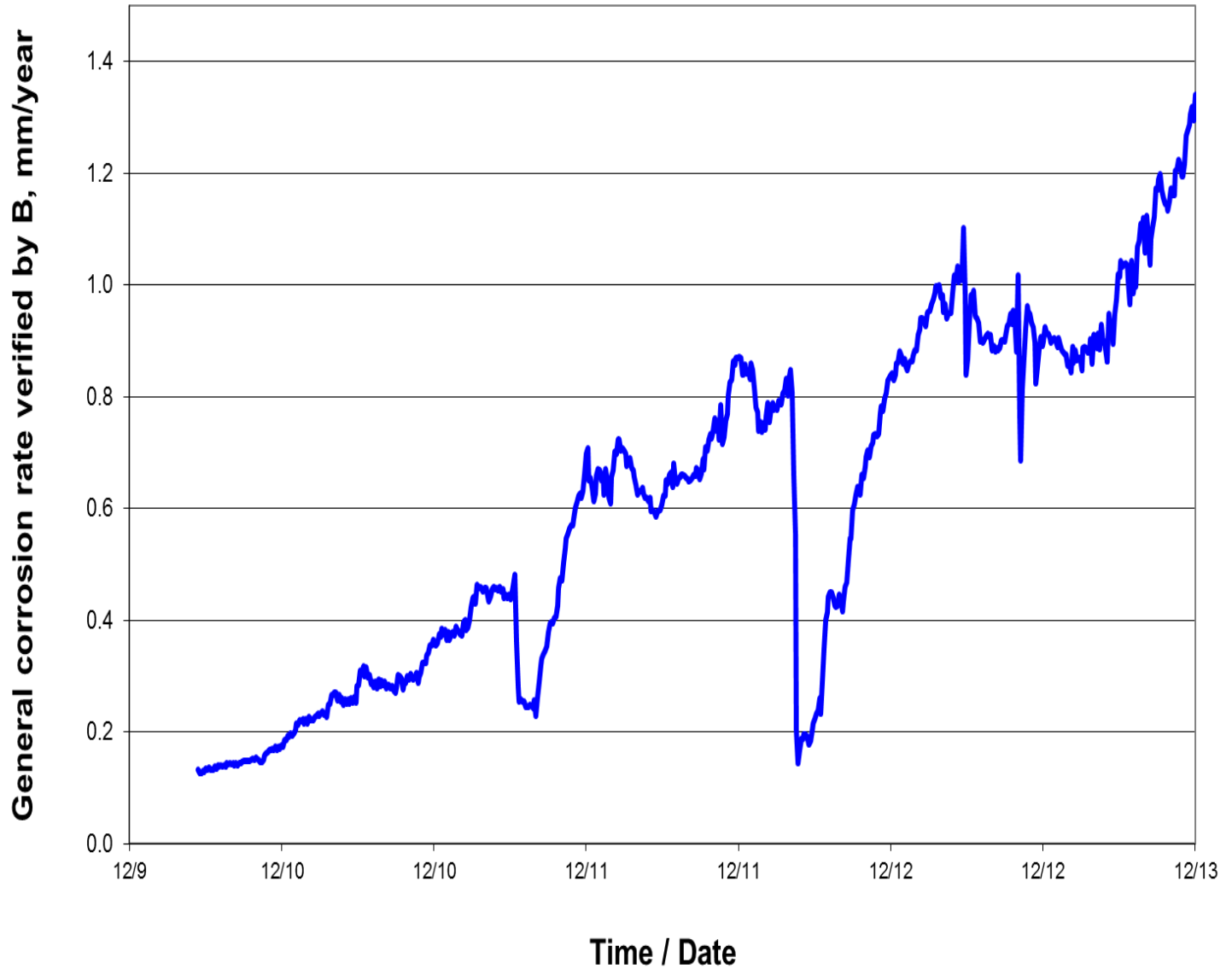
Architecture of real time monitoring/prediction



Amine unit – real time monitoring points (example)



Amine unit – real time monitoring points (example) 11



Environment

Solvent: MEA DGA DEA MDEA

Solvent Concentration: %

Total Pressure: kPa(g)

Temperature: °C

H2S Loading: mole H2S / mole solvent

CO2 Loading: mole CO2 / mole solvent

Total Gas Loading: mole total gas / mole solvent

Impurities: wt%

Application

Pipe Inner Diameter: mm

Design Life: years

Corrosion Allowance: mm

Pipe Roughness:

Custom Roughness: mm

Process Stream Conditions

Custom Wall Shear Stress:

Type of Flow: Horizontal Vertical

Configuration:

Pre-sets

Process Flow Rates and Properties

Vapor Properties

Flow Rate: Standard Actual

m3/d(N)

Density: kg/m3

Viscosity: Pa-s

Liquid Amine Properties

Flow Rate: m3/d

Density: kg/m3

Viscosity: Pa-s

Service
Point
Data Source
Preferences
Help
License

+ Add Point
✎ Edit Point
👤 Enable/Disable Point
👤 Validate Point
✔ Show Point
🔄 Update Manual Values
👤 Import

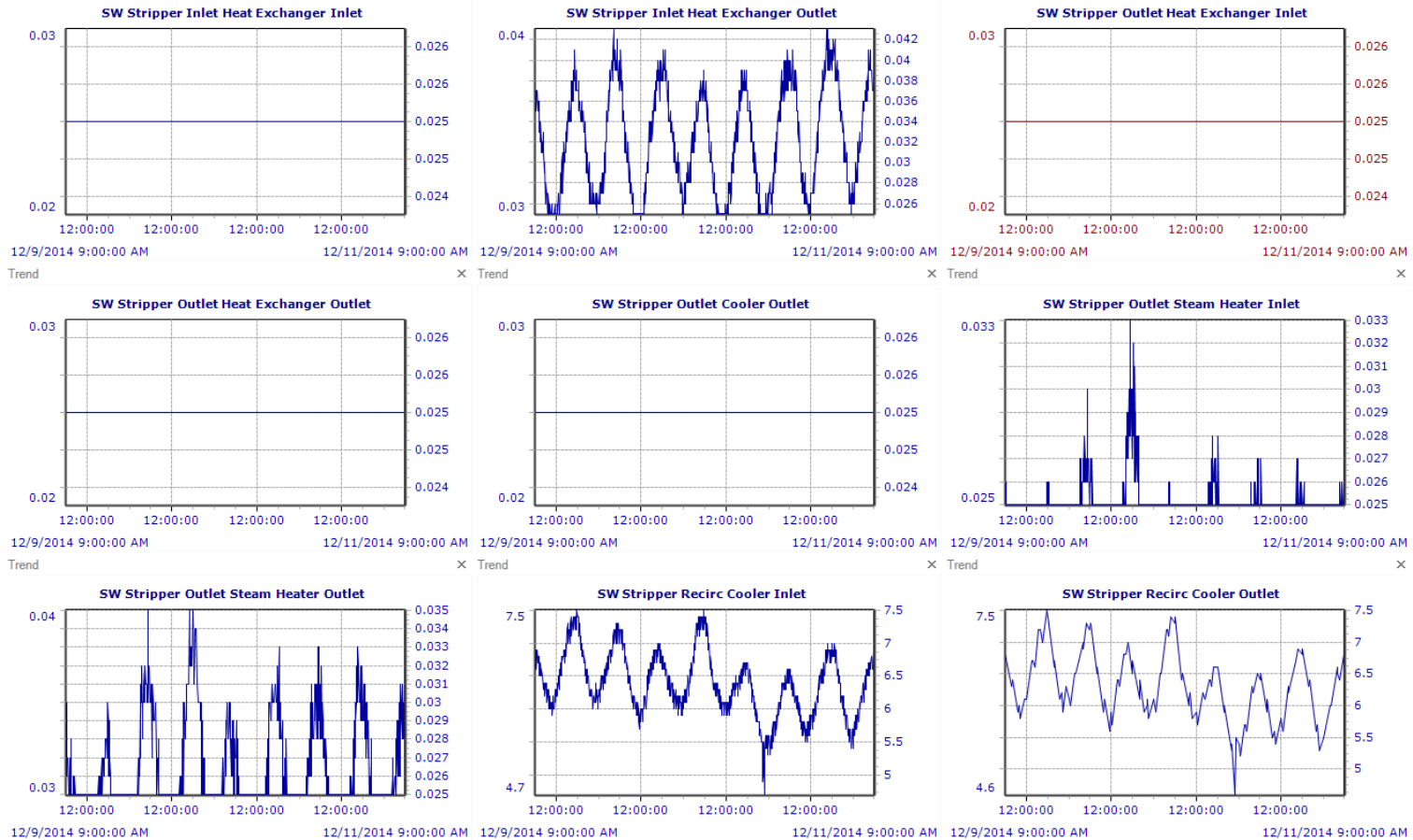
Point Specification
System Parameters
Environment Parameters
Corrosion Rates
Flow Results
KPIs
KPI Manual Values
Update Manual Values

Manual	Parameter Name	Tag Name	Manual	Parameter Name	Tag Name
<input type="checkbox"/>	Total Pressure (kPa(g))	T001.PRESS	<input type="checkbox"/>	Temperature (°C)	T001.TEMP
<input checked="" type="checkbox"/>	H2S (mole %)	NA	<input checked="" type="checkbox"/>	H2S Partial Pressure	NA
<input type="checkbox"/>	NH3 (mole %)	T001.NH3	<input checked="" type="checkbox"/>	NH3 Partial Pressure	NA
<input type="checkbox"/>	NH4HS(wt%)	T001.NH4HS	<input checked="" type="checkbox"/>	Free CN (ppmw(aq))	NA
<input checked="" type="checkbox"/>	Chemical Type	NA	<input checked="" type="checkbox"/>	Oil Type	NA
<input checked="" type="checkbox"/>	Chemical Conc.(ppmv)	NA	<input checked="" type="checkbox"/>	Type Of Flow	NA
<input checked="" type="checkbox"/>	Configuration	NA	<input checked="" type="checkbox"/>	Override Hydrocarbon	NA
<input type="checkbox"/>	Gas Flow Rate (m3/s)	T001.GFR	<input type="checkbox"/>	Gas Density (kg/m3)	T001.GAS_DEN
<input type="checkbox"/>	Gas Viscosity (Pa-s)	T001.GAS_VIS	<input type="checkbox"/>	SW Flow Rate (m3/d)	T001.SW_FR
<input type="checkbox"/>	SW Density (kg/m3)	T001.SW_DEN	<input type="checkbox"/>	SW Viscosity (Pa-s)	T001.SW_VIS
<input type="checkbox"/>	HC Flow Rate (m3/d)	T001.HC_FR	<input type="checkbox"/>	HC Density (kg/m3)	T001.HC_DEN
<input type="checkbox"/>	HC Viscosity (Pa-s)	T001.HC_VIS			

Back
Next
Finish
Edit

Point	Description	Location
T001	SW Stripper Inlet Heat Exchanger Inlet	\\Enterprise\Hydrotreater\Stripper
T002	SW Stripper Inlet Heat Exchanger Outlet	\\Enterprise\Hydrotreater\Stripper
T003	SW Stripper Outlet Heat Exchanger Inlet	\\Enterprise\Hydrotreater\Stripper
T004	SW Stripper Outlet Heat Exchanger Outlet	\\Enterprise\Hydrotreater\Stripper

Monitor “virtually” Corr. Rates at multiple locations



Example of on-line, real time corrosion modelling in Sour Water Stripper

Smarter Approach to Corrosion Management

- **Understand**

- Document damage mechanisms and corrosion loops
- Use **prediction models** to identify critical locations and variables that may impact asset integrity
- Establish **effective** Integrity Operating Windows
- Plan and design to avoid corrosion failures

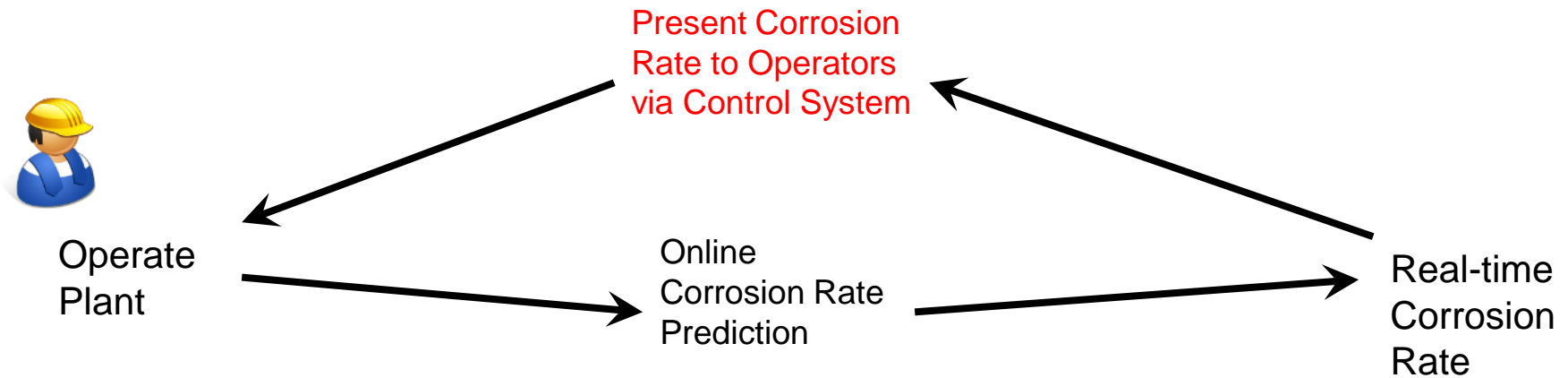
- **Monitor Real-time**

- Regard corrosion as a **process variable**
- Correlate process changes to corrosion
- Know and correctly quantify **current** health of assets; everyday
- Incorporate Boundary and Deviation Management
- Leverage Predictive Models to Detect Abnormal Events
- Review the effectiveness of Integrity Operating Windows on preventing excessive damage

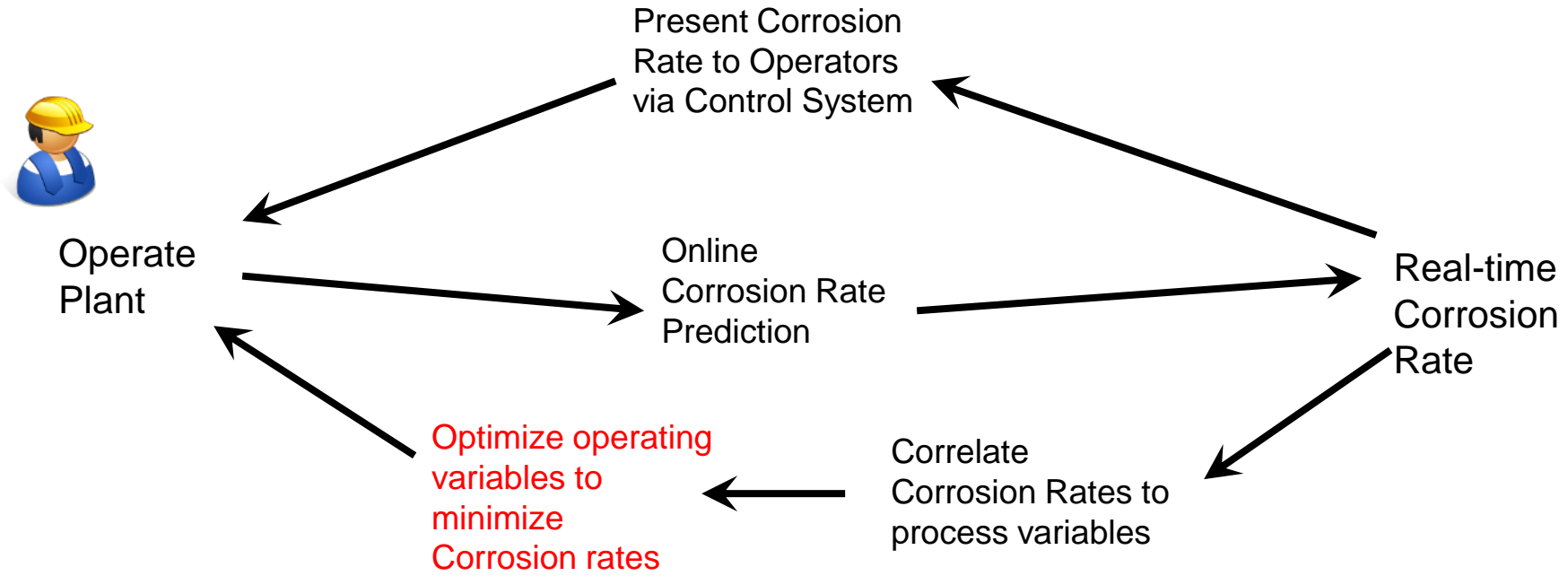
- **Focus Resources**

- Alert Operators, Engineers and Managers of abnormal conditions as they occur.
- Focus resources where they are needed most at the appropriate time.

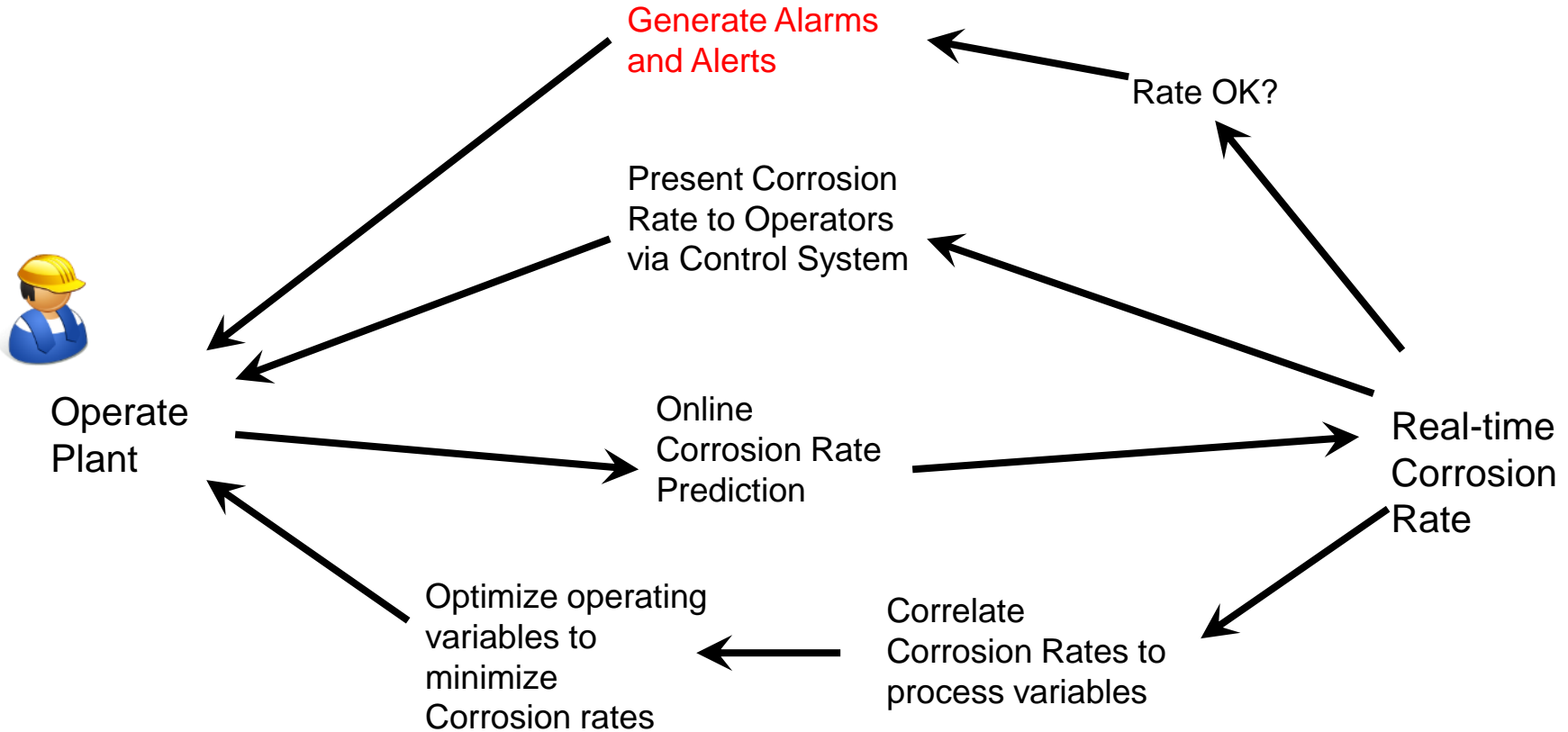
Step 1: making corrosion **visible** to Operation



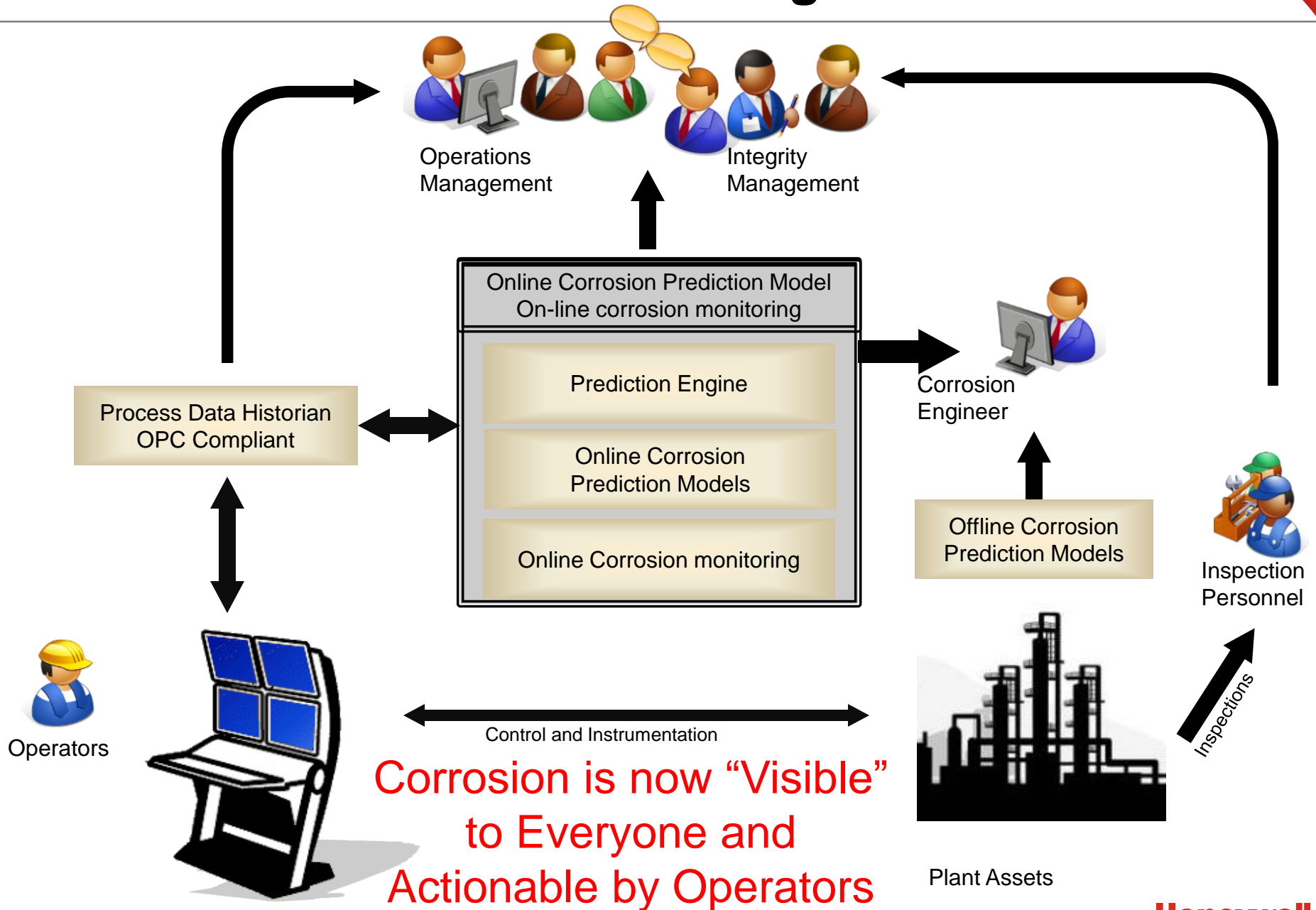
Step 2: making corrosion **relevant** to Operation



Step 3: making corrosion **actionable** by Operation



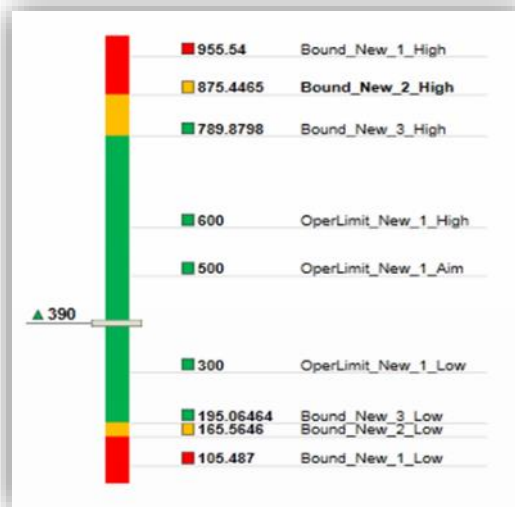
Real-Time Predictive Modeling Solutions



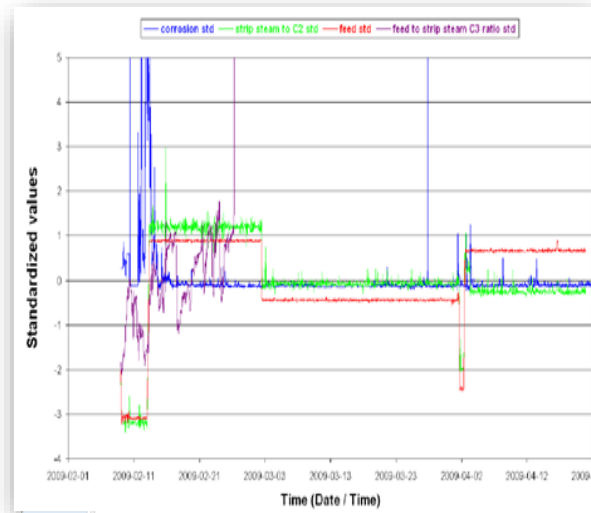
Corrosion is now "Visible" to Everyone and Actionable by Operators

On-line, Real-time Integrity Management

Set and Prioritize IOW Boundaries



Monitor & Predict On-line, Real time



Analyse and act before incident



Parameter: CDU Top temperature
 Tag: IOW-TI-CD1-001
 IOW type: Critical
 Damage mode: NH_4Cl & HCl corrosion
 Risk Rank: 20 (Medium-High)
 Target L/H: 118/138°C

$T = >115$
 Real time predicted CR = 15mpy
 On-line measured: 13.8mpy

Operator's Action:
 Adjust the OVHD T within 1 hour
 Corrosion Engineer notification:
 High risk for OVHD corrosion
 Management Dashboard feedback:
 Rising Cost of Inspection

Summary

- Effective corrosion management requires timely corrosion information
- On-line, real time corrosion monitoring cannot cover all corrosion areas within given process unit
- Traditional, corrosion prediction can statically highlight a corrosion hot-spots within the process unit, but cannot reflect their changes upon process dynamics
- Modern, DCS-integrated real time corrosion models facilitate continuous analysis of multiple process streams
- Based on on-line, real-time modelling data it is possible to predict future locations for corrosion monitoring - e.g. change of throughput may shift corrosion to new hot spots
- Integration with operations / boundary management systems (including IOW alarming and management) and plant KPI dashboard makes corrosion visible to Plant Management

Appendix 14

Naphtenic crude management

(A. Pothuaud)

GE Power & Water
Water & Process Technologies

How CFD Audit Helped Refinery to Manage Naphthenic Corrosion

EFC – La Defense – 26th April 2016

Alain POTHUAUD



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Why HAC in 2016

Crude oil price is low at # 42\$/bbl and should remain low in 2017

Discount between “classical” and “High TAN” crudes is also low

But

- Independent companies can increase their benefits
- Integrated companies can prepare the future with serenity

Predator Objectives

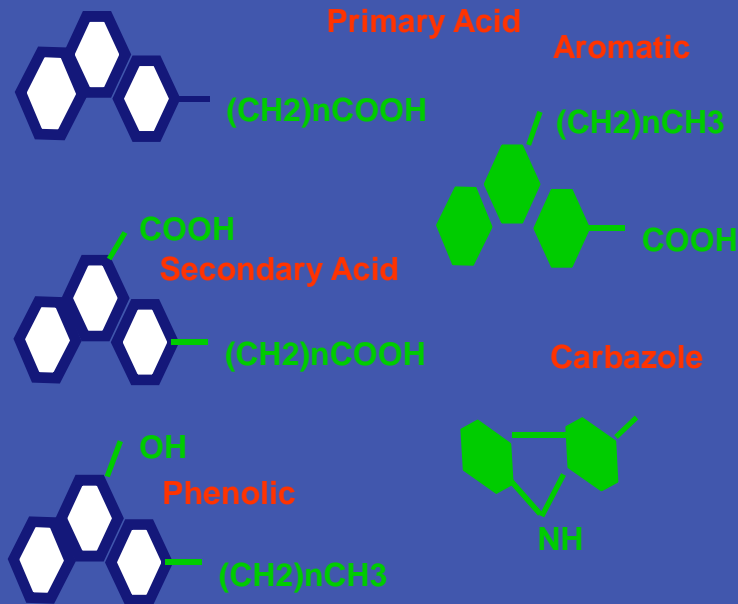
Provide the refining industry with an increased and unique level of reliability while processing discounted or difficult crude feedstock

Predator is a 3 fundamental pole program :
Prediction, Protection, Detection

 **Risk must become an opportunity**

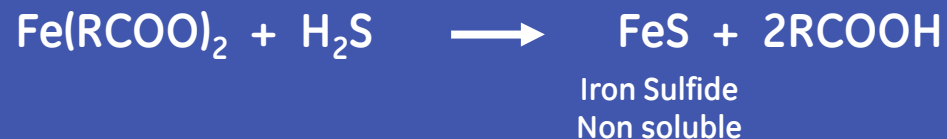
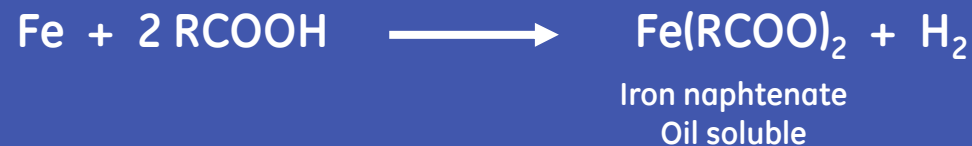
Naphtenic Corrosion

Different species of cyclic compounds (12 to up to 30 Carbons)



Corrodent if

- T° between 220 and 400 $^\circ C$
- Turbulence area
- Velocity
- Metallurgy
- TAN
- NAN
- TAN/S



Protection

Product

➤ *Predator chemical – Low P product*

- Low phosphorous patented product
- Low fouling potential P product
- Different phosphate esters working in synergy
- Recommended especially with low S crude blends
- Recommended also for high temperature sulphur corrosion

Fe + P = Iron Phosphate film

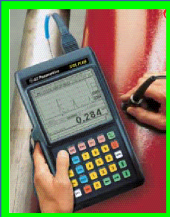
Detection



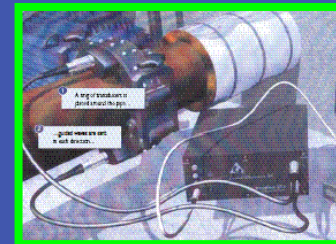
Rightrax



PEC



Hi Temp UT



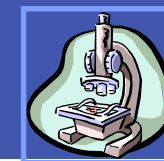
Guided Acoustics



Hydrogen Diffusion



Probes / Coupons



Fe / Ni Ratios

Integrated Measurement

GE W&PT

Third party

Current

Coverage

Local (more sensitivity)

Global (less sensitivity)

Rate Measurement

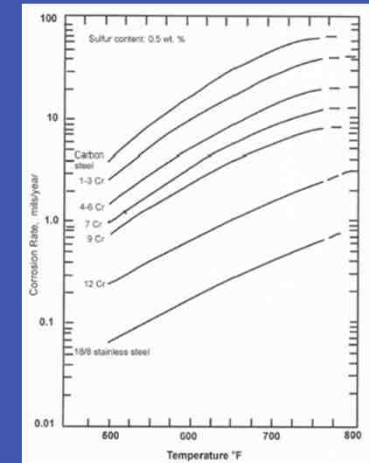
Challenge is Prediction

- TAN is increasing
- S is decreasing
- Better understand where to put representative and adapted corrosion monitoring tools
- Audit conducted on-site to better understand impacts of
 - Unit configuration
 - Naphthenic crudes
- Select right monitoring tool ifo available place
- Design chemical treatment

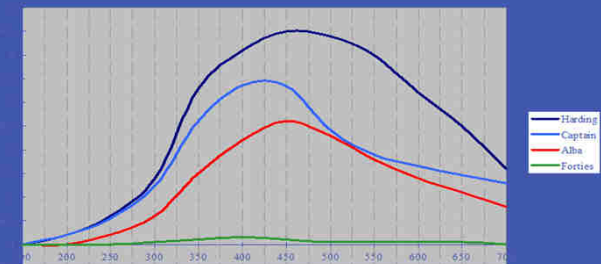
Corrosion Prediction

Different models try to predict corrosion

- McConomy curves
 - Applied to S corrosion
 - Nap Acid corrosion is not taken into account
- API 581
 - Estimated corrosion for different metallurgies
- Crude characterization
 - Based on distillation curve of Nap'Acids
- All define independent informations



Acidity



- Need to correlate unit configuration and naphthenic acidity with specific unit parameters

Solution For Processing Opportunity Crudes

Crude Characterization

- **NAN & TAN**
- **Fingerprint & Corrosion**
- Characterization Sulfur etc.
- Corrosion Model



Asset Determination

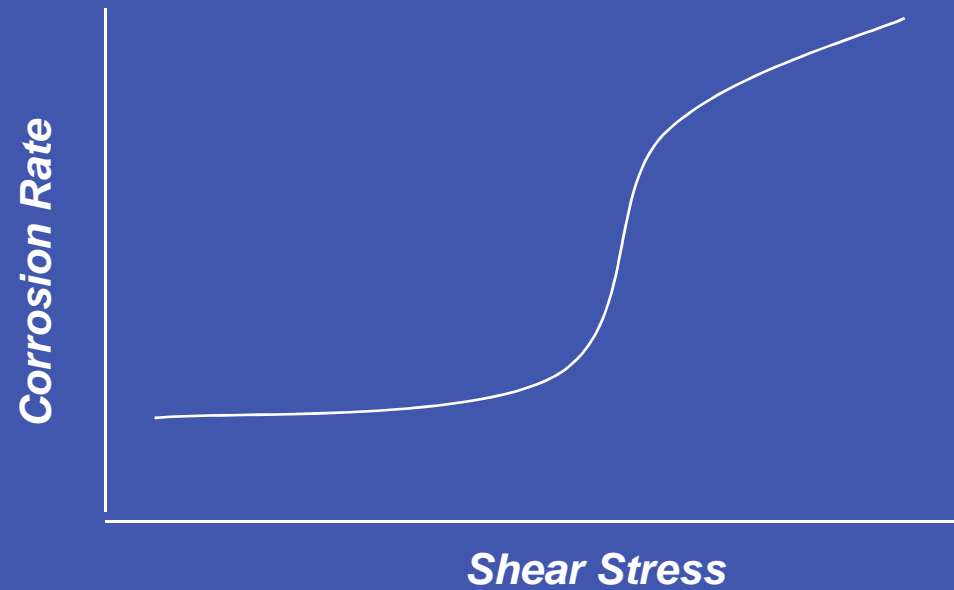
Flow, Pipe Size, Velocity, Temperature, Pipe Configuration and Spatial Relationship, Pipe Geometry, Shear Stress, Metallurgy,



Assessment Output

- Highlight critical areas
- Identify hot spots for corrosion
- Analysis requirements/schedule & location
- Chemical injection requirements /chemistry & application
- Monitoring requirements/schedule, location & sensor type

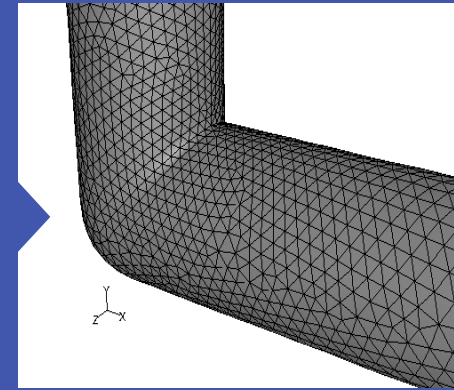
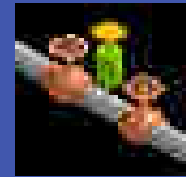
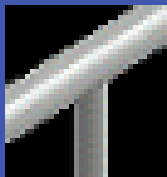
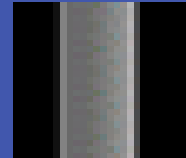
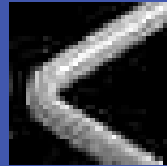
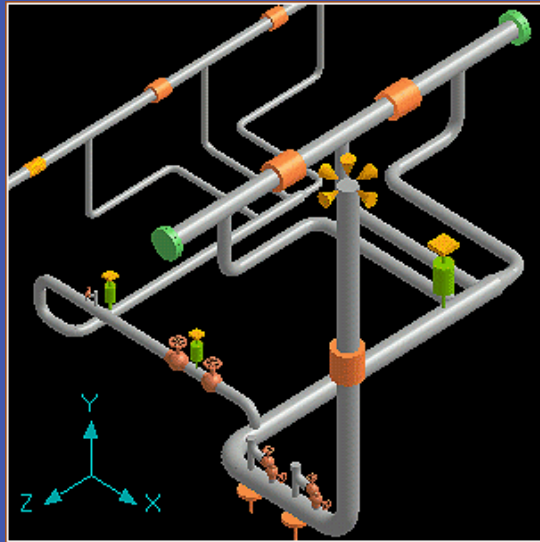
Corrosion & Shear Stress



Shear Stress is critical to understand Corrosion

- Need to identify hot spots based on Shear Stress calculation

Computational Fluid Dynamic Purpose



Refinery piping configuration

Pipeline configuration, components and geometry

Corrosion models developed for individual components through CFD

- Shear Stress calculation
- Hot spot identification
- Monitoring Placement

Computational Fluid Dynamic - CFD

Proprietary GE software technology incorporates various process and operating variables to predict specific areas of higher corrosion potential

Inputs

Characteristics of each element of turbulence (valve, reducer, pump, T, 90...)

Existing system metallurgy

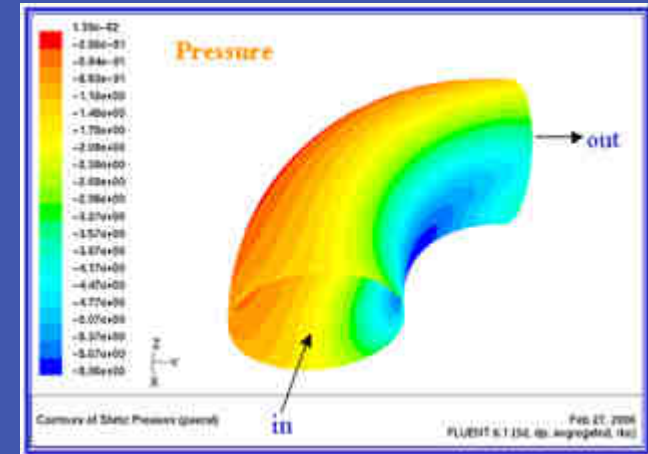
TAN, NAN, S, temperature, and line size for each node

Outputs

Calculates Shear Stress for each node

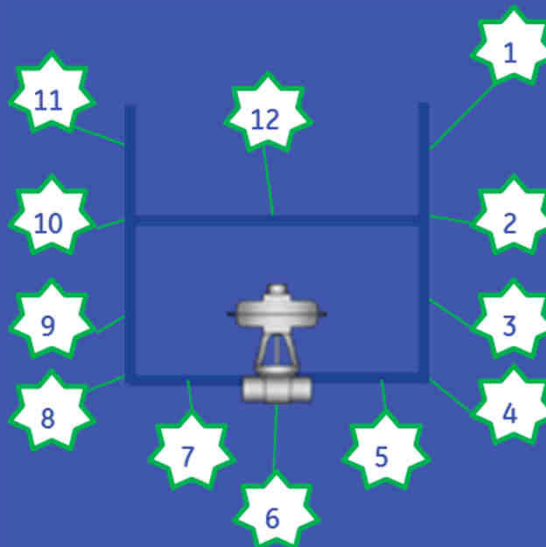
Calculates velocity for each node

☞ **PRIORITIZES EACH PIECE OF EQUIPMENT FOR ELEVATED CORROSION POTENTIAL**



Data Input

- All nodes of all circuits in CDU and VDU
- Need for isometrics and P&iD's
- Temp, pipe size, flow, metallurgy, acidity....
- Geometry and configuration
- Each of the element for each node (valve, by-pass...)
- Up to 250 lines for each circuit
- About 1 day at least per circuit to fill in all the data



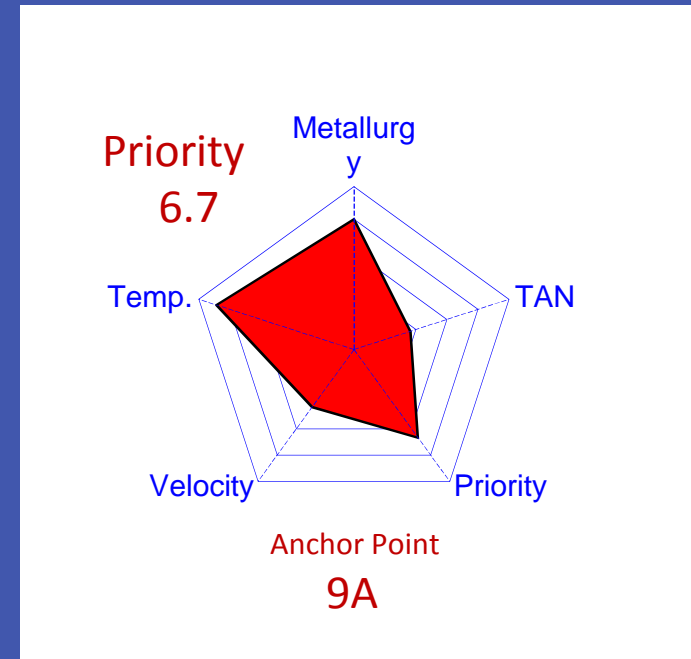
Data Output

Shear stress value

Rank higher priority hot-spots

Weight relative parameters
impact

- Define monitoring hot spots
- Determine inhibition needs



Conclusions

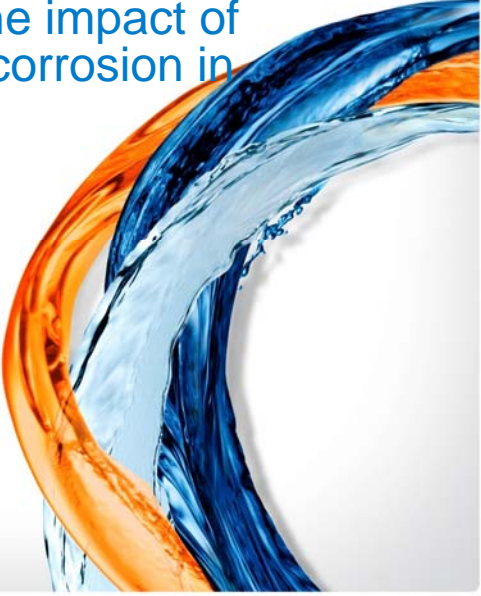
1. The highest priority hot-spot revealed by the project was not being monitored, inspected and was not getting fresh chemical
2. New corrosion monitoring equipment installed to cover weak spots
3. Explain un-expected corrosion issues

Appendix 15

**Use of chlorine dioxide to optimize bio-control
and reduce the impact of oxidizing biocide on
corrosion in cooling water systems**

(V. Bour-Beucler)

Use of chlorine dioxide to optimize bio-control and reduce the impact of oxidizing biocide on corrosion in cooling water system



Valerie Bour Beucler
EFC WP15 Spring meeting
La Défense 26/04/2016

Agenda

- ▲ Introduction
- ▲ Regulation and cooling system management
- ▲ Advantages of chlorine dioxide / hypochlorite
- ▲ Chlorine dioxide technology / Purate technology
- ▲ Chlorine dioxide PURATE case studies
- ▲ Questions

Cooling water system successful management

- ▲ Cooling water successful management
 - A good equilibrium between corrosion, scaling and MIC

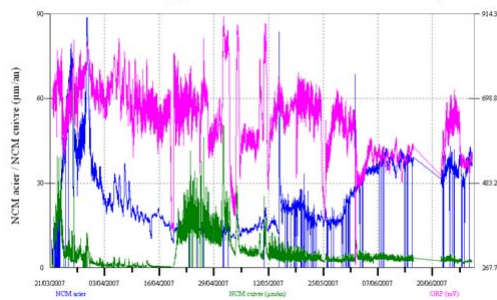


Regulation and cooling system

- ▲ Legionella Control and regulation
- ▲ Biocidal Product Directive / Regulation
 - harmonise the European market for biocidal products and their active substances.
- ▲ REACH (European Community Regulation on Chemical and their safe Use (EC 1907/2006))
 - It deals the Registration Evaluation Authorisation and Restriction of Chemical substances.
- ▲ The Future
 - Less non oxidizing biocides
 - More oxidizing biocide but with AOX control
 - Chlorine dioxide (ClO₂), a good alternative

Biomangement in cooling water system

Legionella results $< 10^3$ ufc *Lp* / l



Use of oxidizing biocides (bleach)

- ✓ Continuous injection
- ✓ Shock injection

- Corrosion on copper alloys
- Increase chlorides
- Increase pitting corrosion on carbon steel

Bio-population + Biofilm

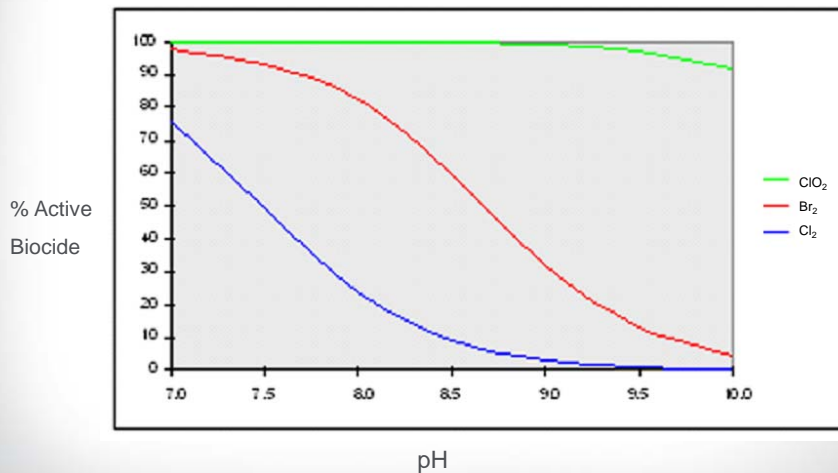
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Why is ClO₂ a Good Biocide?

- ▲ It's a gas that is highly soluble in water, diffuses into biofilms attacking the bacteria generating the biofilm
- ▲ It doesn't hydrolyze like chlorine gas or bleach: no loss of effectiveness at higher pHs
- ▲ It is non-reactive to most organics and ammonia
 - No loss of biocide effectiveness due to byproduct reaction
 - Doesn't react with the cooling water program
- ▲ It oxidizes (no chlorination)** - no byproduct AOX or THMs after hydrocarbon leaks

Typically 10-40% of the bleach requirement

Biocide Efficacy With Increasing pH



ClO₂ Reactivity

Reacts with:

- ▲ Reduced sulfur, iron, manganese
- ▲ Phenols
- ▲ Cyanide
- ▲ Mercaptan (thiol)
- ▲ Organic sulfides
- ▲ Secondary amines (R₂NH)
- ▲ Nitrite

Doesn't react with:

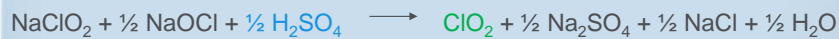
- ▲ Ammonia, 1^o amines
- ▲ Al-kane, kene, kyne
- ▲ Alcohols, glycols, diols
- ▲ Aldehydes, ketones, ethers
- ▲ Organic acids
- ▲ Unsub aromatics
- ▲ Azoles
- ▲ PBTC

Onsite ClO₂ Generation- The acid factor

- Industrial Scale technologies to produce chlorine dioxide that are gaseous chlorine-free require the use of a strong mineral acid to convert a salt into chlorine dioxide.
- Some technologies use hydrochloric acid while others use sulfuric acid.
- Both hydrochloric acid and sulfuric acid require special handling as they are industrial chemicals; as are sodium chlorite, sodium chlorate, hydrogen peroxide and sodium hypochlorite.

Some common gaseous chlorine free industrial processes

Chlorite



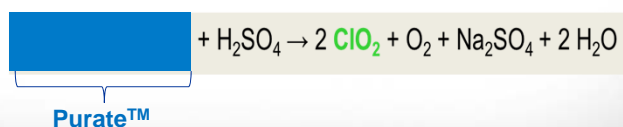
Purate[®]



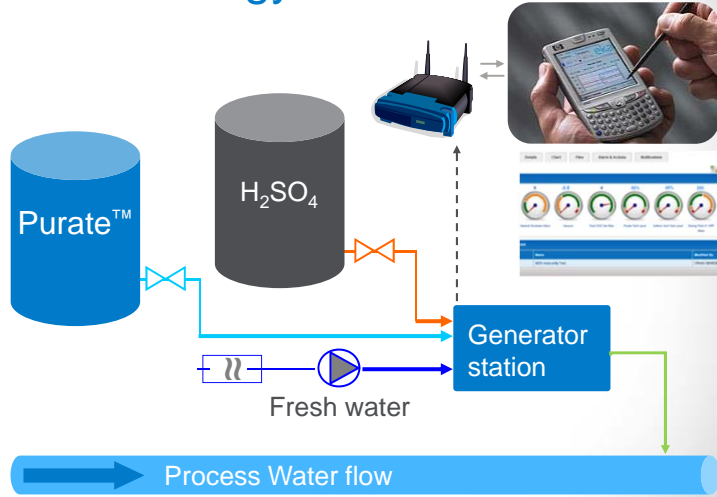
What is purate

On site on demand generation of chlorine dioxide in a water solution

- Chlorine dioxide
 - Powerful oxidation agent
 - Well proven
 - Reacts more selective with organic compounds than Cl₂ and Hypo
 - Gaseous molecule readily soluble in water



Purate™ Technology



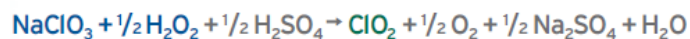
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Purate™ - ClO₂ Generator

Case studies

Sea water once-through cooling system

- ▲ Major refinery complex in Middle East
- ▲ Biofouling and heat transfer issues
 - Chronic leaks on heat exchanger of the large sea water once through cooling system (20 000 m³/h).
 - HE leaks due to under-deposit corrosion associated to biofouling
 - Historically Chlorine gas then Hypochlorite
 - A trial with chlorine dioxide (Purate Technology) has been done (6months)



Performance monitoring

- Safety, microbio visual (seawater strainers and fouling plate)
- Microbio testing
- Chlorine dioxide residual and ORP reading
- Sea water heat exchnager performance

Microbio counts and heat transfer monitoring

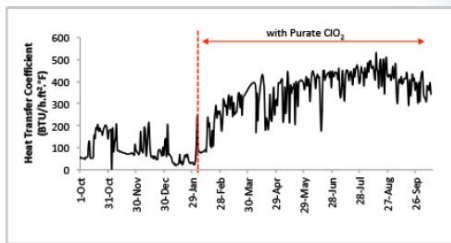


Figure 3: MED train C heat transfer coefficient improvement

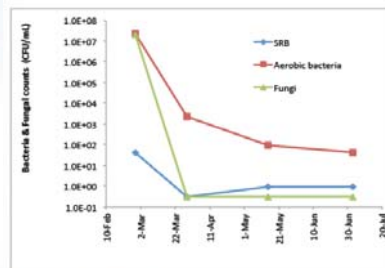


Figure 2: Microbial and fungal counts in seawater system

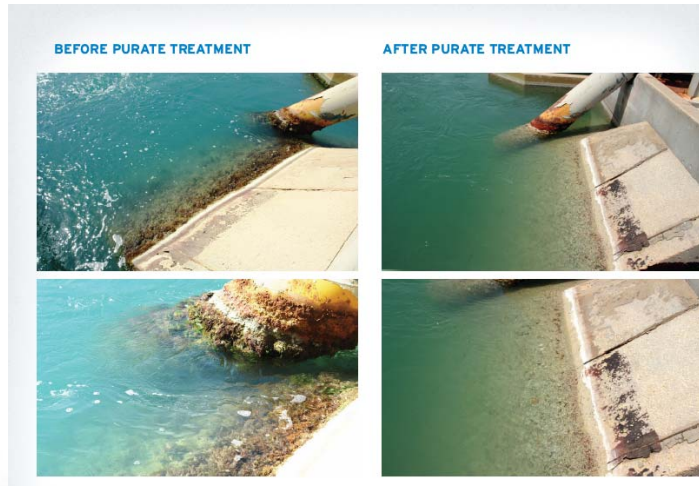
Results

ASPECT	IMPROVEMENT
Safety	A safer system for employees and environment
Monitoring and Control	Rigorous testing, additional ORP monitoring recommended
Analysis & Visual	Log 2 to Log 7 reduction in TVC, Cleaner Surfaces
P15-VDU strainer backwash frequency reduction	80%
P15-VDU HX Heat Transfer Coefficient	22%
MED-Heat Transfer Coefficient	70%

Figure 4: Impact of Purate technology on customer operation

Customer Impact	eROI™	Economic Results
Heat transfer rate improved by 70%; 85.72 MBTU/h. Natural gas savings equal to 700,000 MBTU/year.		Natural gas savings of \$4 million/Year
Above energy savings equates to 20 million Nm³ of natural gas, avoiding 37,000 tons of CO₂		
Reduced the cost of maintenance and increased the reliability of the heat exchangers in the plant		Savings worth \$ 0.5 Million/year

Visual results before and after purate

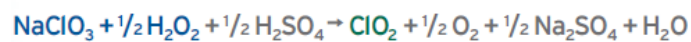


Sea water once-through cooling system

▲ Petrochemical plant in scandinavia

▲ Macrofouling and biofilm control

- Macrofouling and unplanned shutdown for heat exchangers cleaning
- Once through cooling systems is used for controlling the temperature of the process.
- Mussel and barnacles from sea water piping reduce flowrate and cause high pumping cost.
- Unplanned shutdown impacts the total cost of operation and reliability

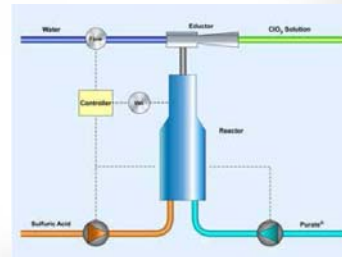


- Historically high level of Hypochlorite but without efficiency
- A trial with chlorine dioxide (Purate Technology) has been done
- Trial successful

Chlorine dioxide / Purate technology

Key benefits of ClO₂

- Chlorine free process - no AOX and THM formation
- Preferred oxidizer for removing biofilm and preventing biofilm formation
- Practically pH independent (pH 4-9)
- Very effective for controlling Legionella, Gardia, cryptosporidium etc.
- Effective against mussels



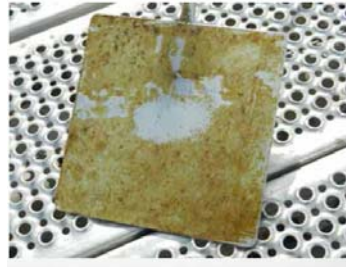
Results

Environmental Indicators	eROI	Economic Results
Cleaning including confined space entry no longer required several times per year	 SAFETY	Reduced reliability costs
No unplanned plant down time to clean the cooling system	 ASSETS	Cost saving of prevented down time: €2 Million per year €220000 for cleaning 8 heat exchangers, typically 2x per year

Results



With hypochlorite



With purate chlorine dioxide after 8 weeks

NALCO Champion
An Ecolab Company

Taking Energy Further™ 21

THANK YOU

