

# **Appendix 1**

## **List of participants and excused persons**

**Participants EFC WP15 meeting 15<sup>th</sup> April 2008 Leiden**

<b>Name</b>	<b>Surname</b>	<b>Company</b>	<b>Country</b>
Beucler	Valerie	Nalco Energy Services	FRANCE
Claesen	Chris J	Nalco	BELGIUM
de Bruyn	Hennie	Borealis AS	NORWAY
Deves	Jean Marie	Axens - IFP Technology Group	FRANCE
Gonzalez-Barba	Maria Luisa	Baker Petrolite	SPAIN
Kiiski	Arto	Neste Jacobs Oy	FINLAND
Lorenz	Maarten	Shell Global Solutions International B.V.	NETHERLANDS
Loukachenko	Natalia	Arcelor Mittal	FRANCE
MeLampy	Michael	Hi-Temp Coatings Technology	USA
Munier	Michel	Axens - IFP Technology Group	FRANCE
Nordstrom	Sofi	Nynas AB	SWEDEN
Parr	Dennis	iicorr Limited	UK
Pugh	John	BP	UK
Reynolds	Steve	PPLEU	UK
Richez	Martin	Total	FRANCE
Ropital	François	IFP	FRANCE
Tabaud	Frederic	BP Rafinaderij Rotterdam B.V.	NETHERLANDS
van Roij	Johan	Shell Global Solutions International B.V.	NETHERLANDS
Vanacore	Mario	Nalco	ITALY
Verstijnen	Wim	Shell Nederland Raffinaderij B.V.	NETHERLANDS
Wännman	Lennart	Nynas AB	SWEDEN
Wilms	Dimphy	Applus RTD Benelux	NETHERLANDS

**Excuses received for the EFC WP15 meeting 15<sup>th</sup> April 2008 Leiden**

<b>Name</b>	<b>Surname</b>	<b>Company</b>	<b>Country</b>
Authier	Sylvain	Exxon Mobil	FRANCE
Carroll	Richard	BG Group	UK
Davies	Michael	CARIAD Consultants	GREECE
Dean	Frank	Ion Science Ltd	UK
Farina	Carlo	CEFIT Corrosion Consultant	ITALY
Farina	Carlo	CEFIT Corrosion Consultant	ITALY
Fevrier	Philippe	Total	FRANCE
Floquet	Jean Pierre	Honeywell	BELGIUM
Groysman	Alec	Oil Refineries Ltd	ISRAEL
Hofmeister	Martin	Bayernoil Raffineriegesellschaft mbH	GERMANY
Holmquist	Martin	Sandvik Materials Technology	NETHERLANDS
Hucinska	Joanna	Gdansk Technical University	POLAND
Isaak	György	Env. & Corr. Manager	HUNGARY
Lunarska	Ellina	Institute of Physical Chemistry	POLAND
Maffert	Joerg	Dillinger Huttenwerke	GERMANY
Michvocik	Miroslav	SLOVNAFT	SLOVAKIA
Nielsen	Anni Visgaard	Statoil Refinery, Kalundborg,	DENMARK
Nolan	Peter	Advantica	UK
Pedersen	Iver Espen	StatoilHydro Research Center Rotvoll	NORWAY
Peultier	Jerome	Arcelor Mittal	FRANCE
Pritchard	Andrew	Corrosion & Fouling Consultancy	UK
Roberto Riva	Roberto	Eni R&M	ITALY
Rommerskirchen	Iris	Butting Edelstahlwerke GmbH&Co KG	GERMANY
Roumeau	Xavier	Total	FRANCE
Saarinen	Kari	Zerust Oy	FINLAND
Sargent	Margaret	Shell Global Solutions International B.V.	NETHERLANDS
Scanlan	Rob	Conoco	UK
Siegmund	Gerit	ExxonMobil Germany GfKorr	GERMANY
Smart	Smart	Serco Assurance F	UK
Trasatti	Stefano	University of Milan	ITALY
Zetlmeisl	Mike	Baker Petrolite	USA

# **Appendix 2**

## **EFC WP15 Activities**



## Presentation of the activities of WP15

### European Federation of Corrosion (EFC)

- Federation of 32 National Associations
- 19 Working Parties (WP) + 1 Task Force
- Annual Corrosion congress « Eurocorr »
- Thematic workshops and symposiums
- Working Party meetings (for WP15 twice a year)
- Publications
- EFC - NACE agreement
- for more information <http://www.efcweb.org>

EFC WP15 Spring meeting 15 April 2008 Leiden The Netherlands

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## EFC Working Parties

<http://www.efcweb.org>

- WP 1: Corrosion Inhibition
- WP 3: High Temperature
- WP 4: Nuclear Corrosion
- WP 5: Environmental Sensitive Fracture
- WP 6: Surface Science and Mechanisms of corrosion and protection
- WP 7: Education
- WP 8: Testing
- WP 9: Marine Corrosion
- WP 10: Microbial Corrosion
- WP 11: Corrosion of reinforcement in concrete
- WP 12: Computer based information systems
- WP 13: Corrosion in oil and gas production
- WP 14: Coatings
- WP 15: Corrosion in the refinery industry  
(created in sept. 96 with John Harston as first chairman)
- WP 16: Cathodic protection
- WP 17: Automotive
- WP 18: Tribocorrosion
- WP 19: Corrosion of polymer materials
- WP 20: Corrosion by drinking waters

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## EFC Working Party 15 « Corrosion in Refinery » Activities

<http://www.efcweb.org/Working+Parties-p-104085/WP%2B15-p-104111.html>

Chairman: Francois Ropital

Deputy Chairman: Hennie de Bruyn

The following are the main areas being pursued by the Working Party:

### Information Exchange

Sharing of refinery materials /corrosion experiences by operating company representatives.

### Forum for Technology

Sharing materials/ corrosion/ protection/ monitoring information by providers

### Eurocorr Conferences

### WP Meetings

One WP 15 working party meeting in Spring,

One meeting at Eurocorr in September in conjunction with the conference,

### Publications - Guidelines

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## Publications from WP15

- [EFC Guideline n°40 « Prevention of corrosion by cooling waters »](http://www.woodheadpublishing.com/en/book.aspx?bookID=1193) available from <http://www.woodheadpublishing.com/en/book.aspx?bookID=1193>

Update in relation with Nace document 11106 "Monitoring and adjustment of cooling water treatment operating parameters" Task Group 152 on cooling water systems

- [EFC Guideline n° 46 on corrosion in amine units](http://www.woodheadpublishing.com/en/book.aspx?bookID=1299)  
<http://www.woodheadpublishing.com/en/book.aspx?bookID=1299>

- [EFC Guideline n° 42 Collection of selected papers](http://www.woodheadpublishing.com/en/book.aspx?bookID=1295)  
<http://www.woodheadpublishing.com/en/book.aspx?bookID=1295>

- [EFC Guideline n° 55 Corrosion Under Insulation](http://www.woodheadpublishing.com/en/book.aspx?bookID=1486)  
<http://www.woodheadpublishing.com/en/book.aspx?bookID=1486>



- Future publications

- suggestions ?

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Tel: +44 (0)1223 891 358 Fax: +44 (0)1223 893 694 Email: [wp@woodheadpublishing.com](mailto:wp@woodheadpublishing.com)

## Corrosion under insulation (CUI) guidelines: (EFC 55)

Edited by **S Winnik, ExxonMobil, UK**

- guidelines cover inspection methodology for CUI, inspection techniques, including non-destructive evaluation methods and recommended best practice
- case studies are included illustrating key points in the book

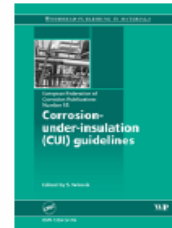
Corrosion under insulation (CUI) refers to the external corrosion of piping and vessels that occurs underneath externally clad/jacketed insulation as a result of the penetration of water. By its very nature CUI tends to remain undetected until the insulation and cladding/jacketing is removed to allow inspection or when leaks occur. CUI is a common problem shared by the refining, petrochemical, power, industrial, onshore and offshore industries.

The European Federation of Corrosion (EFC) Working Parties WP13 and WP15 have worked to provide guidelines on managing CUI together with a number of major European refining, petrochemical and offshore companies including BP, Chevron-Texaco, Conoco-Phillips, ENI, Exxon-Mobil, IFP, MOL, Scanraff, Statoil, Shell, Total and Borealis. The guidelines within this document are intended for use on all plants and installations that contain insulated vessels, piping and equipment. The guidelines cover a risk-based inspection methodology for CUI, inspection techniques (including non-destructive evaluation methods) and recommended best practice for mitigating CUI, including design of plant and equipment, coatings and the use of thermal spray techniques, types of insulation, cladding/jacketing materials and protection guards. The guidelines also include case studies.

ISBN 1 84569 423 6  
[ISBN-13: 978 1 84569 423 4]  
March 2008  
176 pages 234 x 156mm hardback  
**£115.00 / US\$230.00 / €170.00**

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		<a href="http://www.eurocorr.org">http://www.eurocorr.org</a>		
		Edinburgh 8-11 September 2008		
		Monday 8 September		
Monday 8 September 2008	09:00 - 09:15	<b>Opening Session: S. Lyon, M. Schütze</b>		
	09:15 - 09:30	<b>Welcome Address: Member of the Scottish Parliament</b>		
	09:30 - 09:45	<b>Cavallaro Medal Award Presentation</b>		
	09:45 - 10:30	<b>Invited Plenary Lecture: Infrastructure Maintenance - 100 Years of Painting the Forth Bridge, N.N.</b>		
	10:30 - 11:00	<b>Coffee Break</b>		
		Working Party 15 Business Meeting		
	11:00 - 11:25	WP15 Meeting		
	11:25 - 11:50	WP15 Meeting		
	11:50 - 12:15	WP15 Meeting		
	12:15 - 12:40	WP15 Meeting		
	12:40 - 14:00	<b>Lunchtime</b>		
		<b>Workshop Corrosion under Insulation</b>		
	14:00 - 14:25	1197 <b>Winnik/UK</b>	A corrosion under insulation prevention strategy - Experience with an approach aligned with the EFC WP 13 and WP 15 corrosion under insulation guideline	
	14:25 - 14:50	1045 <b>Scanlan/UK</b>	A refinery approach to address corrosion under insulation & external corrosion	
	14:50 - 15:15	1416 <b>MeLampy MA/USA</b>	Comparison of cost, application characteristics, service life, reliability and repairability of systems for preventing corrosion under insulation	
	15:15 - 15:40	<b>de Bruyn/N</b>	Risk-Based Approach to CUI Protection of Austenitic Stainless Steel	
15:40 - 16:15	<b>Coffee Break</b>			
16:15 - 16:40	Discussion			
16:40 - 17:05	Discussion			
17:05 - 19:30	<b>Poster Discussion / Poster Party with Beer and Pretzel</b>			



<http://www.eurocorr.org>

Edinburgh 8-11 September 2008

Wednesday 10 September

W E D N E S D A Y 1 0 S E P T E M B E R	09.00 - 09:45	Invited Plenary Lecture: R.C. Newman/CDN
	09:45 - 09:55	Break for Changing Lecture Hall
		<b>Refinery Process Corrosion</b>
	09:55 - 10:20	1331 Eaton TX/USA Refinery corrosion by salt hydrolysis in opportunity crudes
	10:20 - 10:45	1284 Matsuo /UK Properties of sumitomo 347AP steel tube for use in desulfurizing plants in the petroleum refinery industry
	10:45 - 11:20	<b>Coffee Break</b>
		<b>Refinery Process Corrosion</b>
	11:20 - 11:45	1370 Bhattacharya GA/USA Corrosion and stress corrosion cracking of heat treated 2205 duplex stainless steel in caustic solution
	11:45 - 12:10	1071 Haug /D Reuse of water: Maintaining reliability by avoiding corrosion effects
	12:10 - 12:35	1394 de Freitas /BR Study of alternative systems for corrosion control in water cooling system operating in high concentration cycle
	12:35 - 14:00	<b>Lunchtime</b>
		<b>Refinery Process Corrosion</b>
	14:00 - 14:25	1121 Lyblinski OH/USA Corrosion protection of oil storage tank tops
	14:25 - 14:50	1252 Fleury /F Corrosion in amine solvents used for the removal of acid gases
		<b>Ethanol Biofuel Corrosion</b>
	14:50 - 15:15	1216 Albistur-Goni /E Stress corrosion cracking of carbon steel in ethanol-gasoline blends
	15:15 - 15:40	1367 Troßmann /D Corrosion of metals for automotive applications in ethanol blended biofuels
	15:40 - 16:15	<b>Coffee Break</b>
		<b>Refinery Inspection - Monitoring</b>
	16:15 - 16:40	1133 de Bruyn /N Training & Certification of operating inspectors: A review of Norwegian standards NS 415-1 & NS 415-2
16:40 - 17:05	1372 Dean /UK The utility of hydrogen flux measurement in refineries	
17:05 - 17:30	1145 Holdefer /D Reducing the effects of process corrosion "corrosion as a process variable"	
20.00 - 24:00	Conference Dinner at "The Hub"	



<http://www.eurocorr.org>

Edinburgh 8-11 September 2008

Thursday 11 September

T H U R S D A Y 1 1 S E P T E M B E R	09.00-9.45	Invited Plenary Lecture: Gerald S. Frankel OH/USA
	9:45 - 9:55	Break for Changing Lecture Hall
		<b>Refinery - Naphthenic Acid Corrosion</b>
	09:55 - 10:20	1437 Kane TX/USA Refining high acid crudes: When is an opportunity not an opportunity?
	10:20 - 10:45	1199 Invernizzi /I The effect of molecular structure on the Naphthenic acid corrosion occurrence
	10:45 - 11:10	1128 Groysman /IL New inhibitors for preventing naphthenic acid corrosion
	11:10 - 11:45	
	11:45 - 12:10	1248 Claesen /B 25 years experience of successful naphthenic acid corrosion inhibition
	12:10 - 12:35	1118 Sandu TX/USA Surface study of naphthenic acid corrosion inhibitors on carbon steel
	12:35 - 13:00	
	13:00 - 13:25	
	13:25 - 13:35	Break for Changing Lecture Hall
	13:35 - 13:50	Closing Remarks
	ca. 13:50	End of Scientific Programme





## EFC Working Party 15 plan work 2006-2009

- Task force on Corrosion Under Insulation - achieved
- Failure cases atlas : creation of a task force and a web page
- Cooling water treatment:  
Creation of a task group of WP15 members to cooperate with EFC WP1 and NACE on publication of update or new documents
- Session with NACE at Eurocorr 2009 in Nice on which topic ?
- Sessions with other EFC WP at Eurocorr 2009 in Nice on which topic?



## EFC Working Party 15: Future objectives of the group

### How to manage our working party meetings / Eurocorr sessions

#### · Eurocorr Sessions

✓ Implements of Eurocorr sessions or workshops with other WP and NACE (a workshop can be on a topic without formal presentation)

✓ Implication of young corrosion students, PhD at Eurocorr session with a dedicated poster session

#### · Working Party Meetings

✓ Future topics of task forces

✓ Facilitating student trainings outside their countries in our companies

✓ Presentation of UE funding projects in our area (if they are)

✓ Collaboration on Standard

#### Increase the collaboration with NACE

exchange of information on our activities - joint Eurocorr sessions



## **Appendix 3**

# **Case histories and 5 year track history of a liquid applied CUI coating**

**Michael McLampy**

**(High-Temp Coatings Technology)**

**EFC Working Party 15**  
**Corrosion Refinery Industry Meeting**  
**Oegstgeest/Leiden**  
15th April 2008 9h30 - 16h

## **Corrosion Under Insulation and Case Histories**

Michael MeLampy  
Hi-Temp Coatings Technology  
+1 978 635 1110  
[mmelampy@hitempcoatings.com](mailto:mmelampy@hitempcoatings.com)  
Steve Reynolds  
Performance Polymers Ltd  
+44 1367 242 732  
[steve@pplou.net](mailto:steve@pplou.net)

## Agenda

- A bit of history
- Insulation?
- Boiling water
- Coatings for Insulated Service
- Maintenance
- Case Histories
- Conclusions

“CUI occurs when the conditions ...  
...meet with unprotected steel.

The second critical point is that the  
REASON the steel was not adequately  
protected in the first place is because  
when our plants were built, industry did not  
understand that the environment under the  
insulation was going to be almost like  
immersion conditions (or worse)....

...So the correct type of coatings were not  
used. As a result, almost NONE of the  
surfaces under insulation in every single  
facility which are older than 15 years, are  
NOT adequately protected from CUI. CUI  
is a phenomena because of our  
ignorance.”

→ Monica Chauviere,

*Non-metallic Materials. ExxonMobil Research  
and Engineering AAEO/Materials, Inspection, &  
Support*

# Insulation Leaks .... Eventually





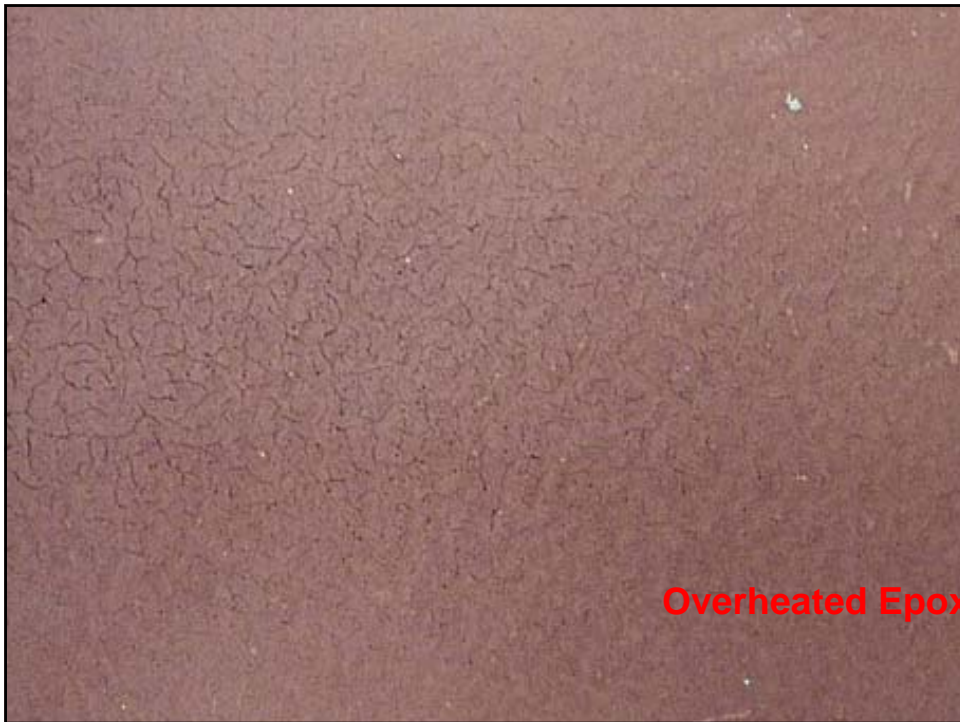








Novalac Epoxy  
After 3 months service  
Originally White  
Cracking  
Excessive DFT in places.



Overheated Epoxy



## Corrosion under insulation

### ■ Zinc Rich Systems – Galvanic Protection



## What are the Consequences of Leaking Insulation?



## Boiling Water

- For a coating to last a long time in needs to survive boiling water or close to boiling water.



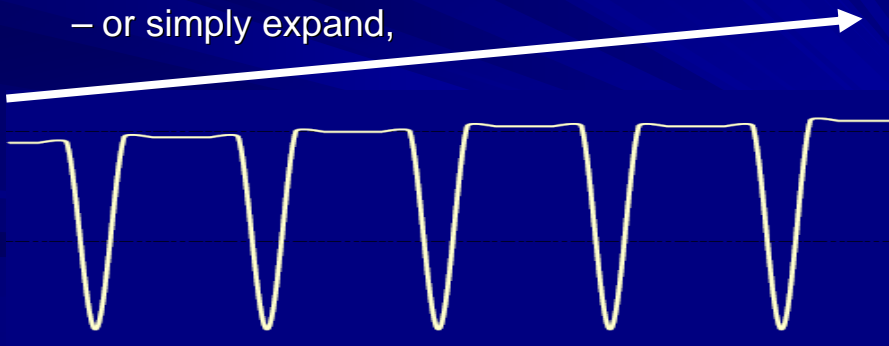
## Surfaces at Elevated Temperatures

- Require specialized coating systems
  - appropriate levels of surface preparation
  - coating application
  - coating materials
    - Allow for future maintenance
  - inspection procedures

- In maintenance situations,
  - Some systems can be applied while units are in service, often near or at the unit's peak temperature operation.
  - Some allow minimal surface preparation
  - Some have very long recoat window

## New Interest In High Temperature Coating Systems

- many older refineries and chemical plants
  - upgrade for higher quality products
  - improved efficiency
  - or simply expand,



	Inert Multi-polymeric Matrix	Thermal Spray Aluminum	Traditional Elevated Temperature Silicones	Multi-polymer Primer	Inorganic Zinc	Novalac Elevated Temperature Epoxies
Max Operating Temp	750°C	630°C	540°C	425°C	400°C	220°C
Max DFT per Coat (in Microns)	150	200	37	150	75	100
Recoat with self	Yes	No	Yes	Yes	No	Yes
Max DFT (in microns)	300 +		112	200		200 mils Max
						Recoat interval is critical
Single Component	Yes	n/a	Yes	No	No	No

	Inert Multi-polymeric Matrix	Thermal Spray Aluminum	Traditional Elevated Temperature Silicones	Multi-polymer Primer	Inorganic Zinc	Novalac Elevated Temperature Epoxies
Anodic Metal sacrifices in Electrolyte	No	Yes	Yes (Aluminum)	Yes (Aluminum)	Yes (Zinc)	No
Intermittent immersion in Salt Water	Yes	Fails	Fails	NR	Fails	Yes
Hot Apply °C	Yes	Yes	Yes	Yes	No	Yes
	260°C		93°C	120°C		150°C
Surface Tolerant	Yes	No	No	No	No	No
	SSPC SP-2					
Stainless Steel	Yes	No	No	Yes	No	Yes
Easy repair with Self	Yes	No	Yes	Yes	No	Yes
Protects at Ambient	Yes	Yes	No	Yes	Yes	Yes
Cryogenic Service	Yes	Yes	No	Yes	No	Yes



# Surface Preparation

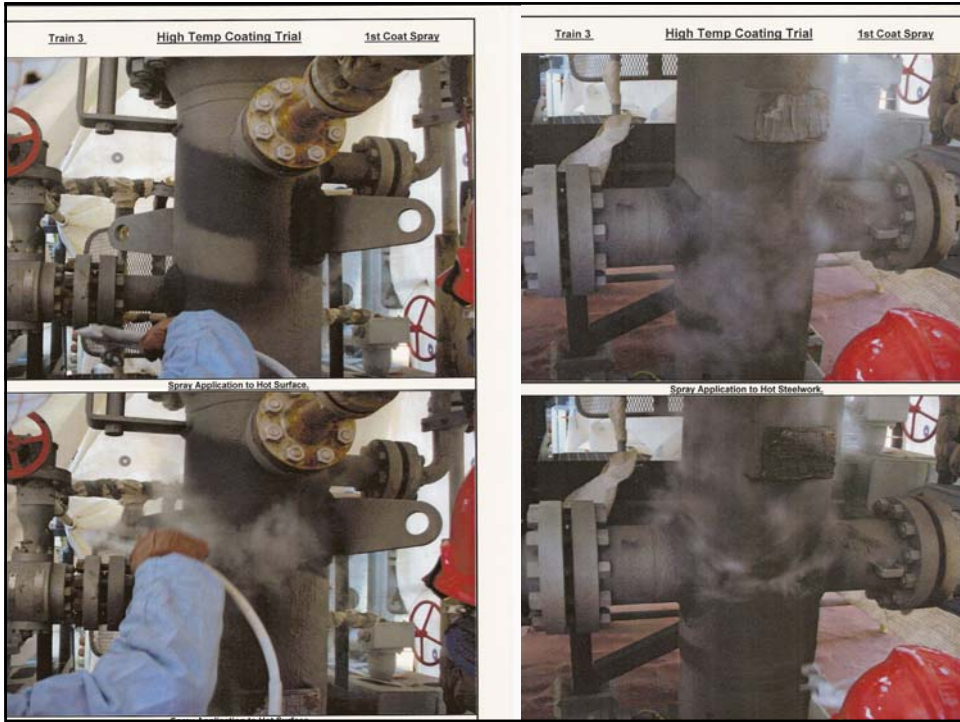
- Near White Metal Clean
- Power Tool Clean
- Hand Tool Clean

But really...

How clean can you get?

And at what cost?





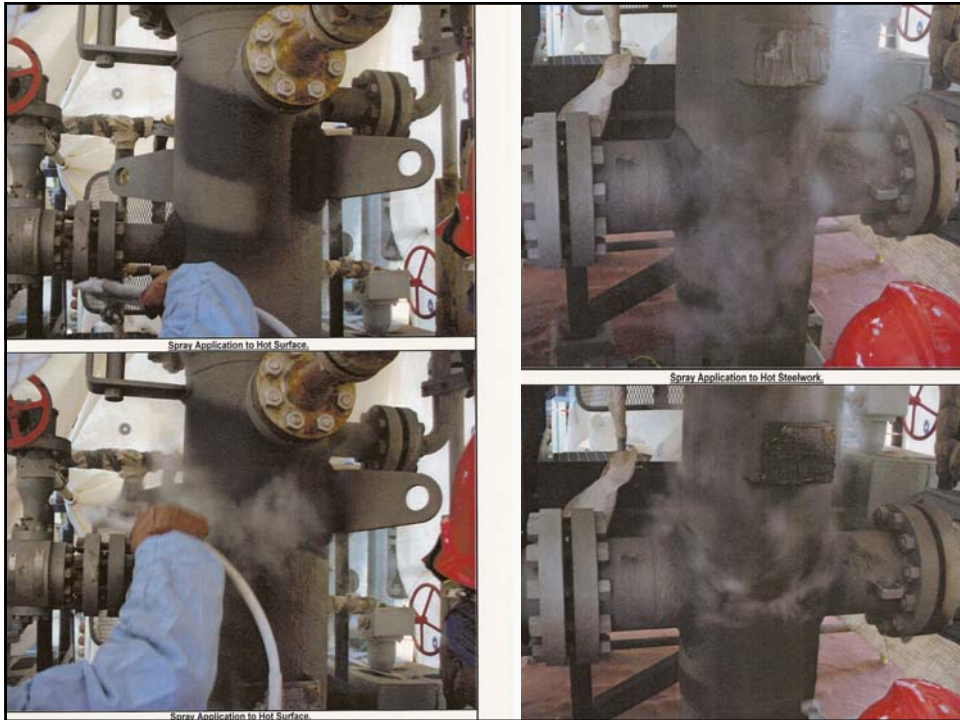
## Case Studies

1. Blast Cleaned – Off-Shore
2. Direct to rust
3. Other Examples

Coating applied to  
Regeneration heater that has  
cyclic temperatures ranging  
from  $-12^{\circ}\text{C}$  to  $160^{\circ}\text{C}$ . Once  
the coatings were applied  
several tests were done  
along with two follow up  
visits.







## Subsequent Visits

- 2<sup>nd</sup> Visit - Coating found acceptable after visual test for cracks, delaminations and other failure modes, Adhesion and Impact tests
- 3<sup>rd</sup> Visit Visual Inspection, insulated and put back into service.



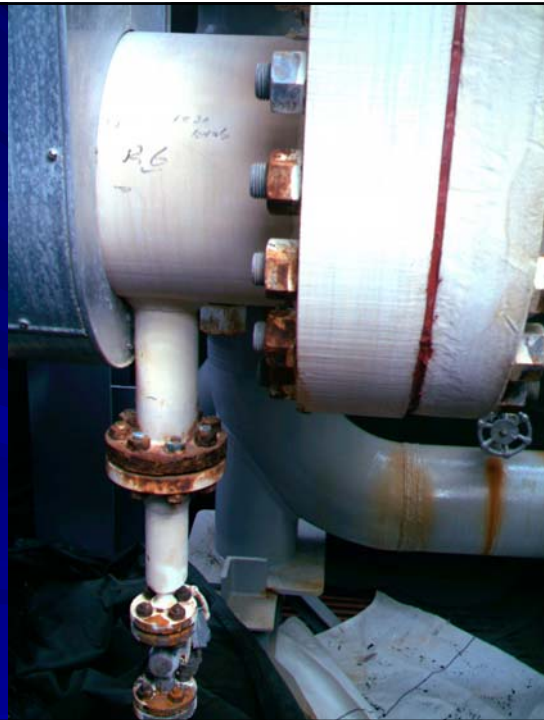


# Off Shore Nigeria (Shell Bonga)

- Nozzles
- Before
- Blast Cleaned
- Primer
- Topcoat



Before



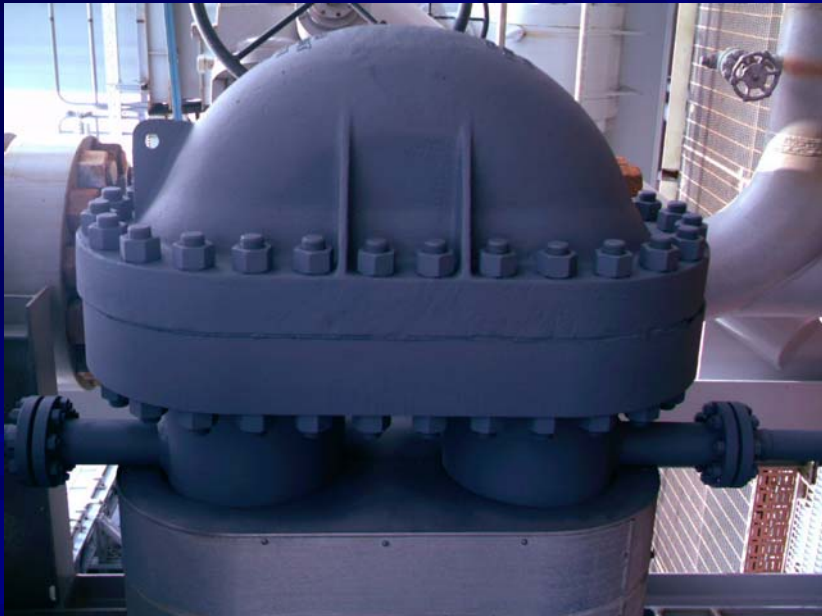


After Blast  
Cleaning



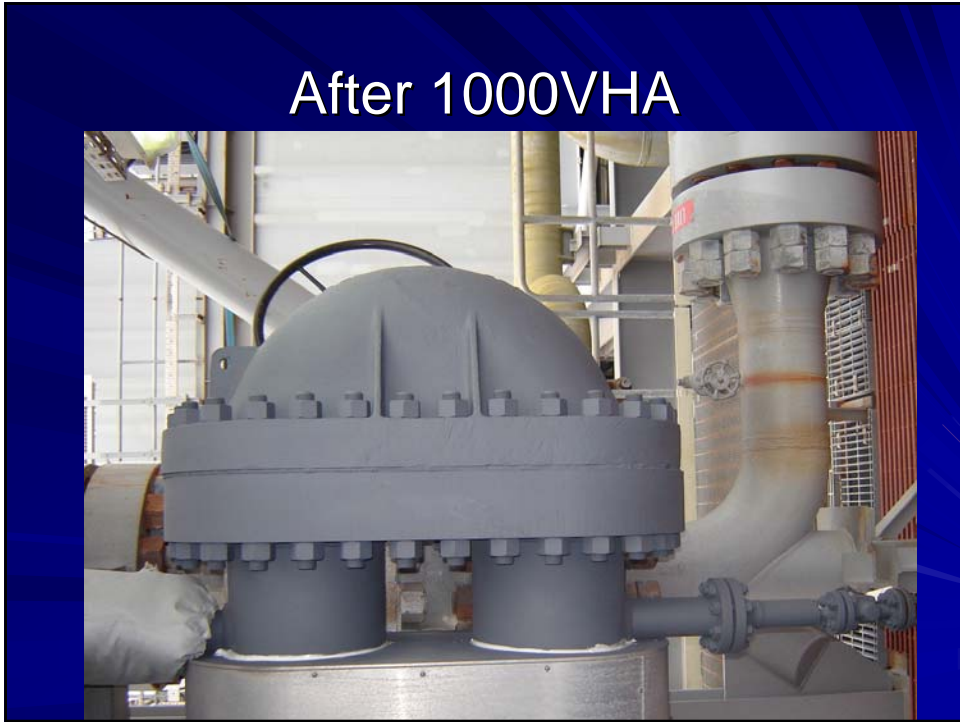


After 1027





After 1000VHA







Direct to Rust / Over Old Zinc



**DOW CHEMICAL, PLAQUEMINE, LA.,**

**Project:** New Construction and Plant Maintenance  
**Location:** Plaquemine, Louisiana  
**Facility Type:** Chemical Plant  
**Start Date:** 2003  
**Owner/contact:** Lynda Mink  
**Contractor:** Sipco/Protherm  
**Contact:** Johnny Thorning (B&H)  
**Engineering:** n/a  
**Coatings:** Hi-Temp 1027 10-15 mils DFT [250-375]Microns  
**Substrate;** Carbon Steel  
**Surface Prep:** Dry Abrasive Blast  
**Environment:** Chemical Plant  
**Project size:**

**PROJECT DESCRIPTION**

Multiple projects dating back to 2003. Hi-Temp Coatings products and systems are on the Dow Global Coatings specification.

**VALERO REFINING, GOOD HOPE, LA.,**

**Project:** Field Maintenance  
**Location:** Good Hope, Louisiana  
**Facility Type:** Petroleum Refinery  
**Start Date:** 2003 -- 2005  
**Contractor:** Mansfield / Brock  
**Contact:** Johnny Thorning (B&H)  
**Engineering:** n/a  
**Coatings:** Hi-Temp 1027 250-375 Microns  
**Substrate;** Carbon Steel  
**Surface Prep:** Dry Abrasive Blast  
**Environment:** Refinery / Chemical  
**Project size:** 15,000 sq.ft. / 1500 sq. M



**PROJECT DESCRIPTION**

Valero Refining, Good Hope, Louisiana, was one of the earliest users of Hi-Temp Coatings Systems for CUI (Coating Under Insulation) projects and for exposed elevated temperature plant equipment as part of turnarounds and routine plant maintenance since early 2003. The facility continues using Hi-Temp products, with a series of small insulated vessel exteriors being the most recent project.

**VULCAN CHEMICAL, GEISMAR, LA.,**

**Project:** New Construction and Plant Maintenance

**Location:** Geismar, Louisiana  
**Facility Type:** Chemical Plant (now owned by Oxychem)  
**Start Date:** 2003  
**Owner/contact:**  
**Contractor:**  
**Contact:** Johnny Thorning (B&H )  
**Engineering:** n/a  
**Coatings:** Hi-Temp 1027 10-15 mils DFT [250-375]Microns  
**Substrate:** Carbon Steel  
**Surface Prep:** Dry Abrasive Blast  
**Environment:** Chemical Plant  
**Project size:**

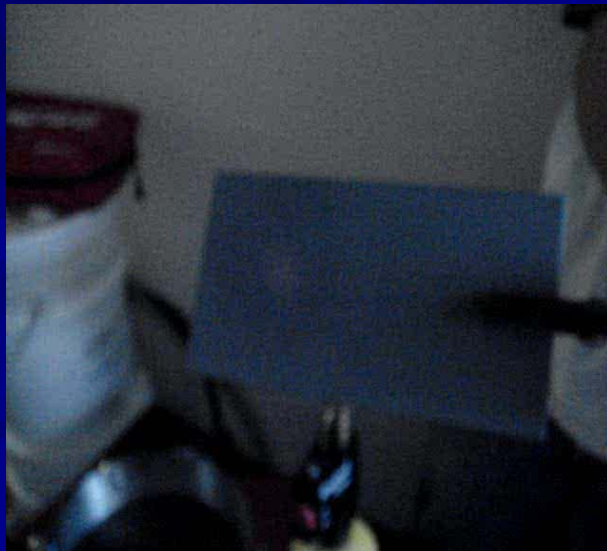
**PROJECT DESCRIPTION**

Multiple projects dating back to 2003.

## -Conclusions-

### Surfaces at Elevated Temperatures

- Require specialized coating systems
  - appropriate levels of surface preparation
  - coating application
  - coating materials
  - inspection procedures.



## **Appendix 4**

### **Unusual CUI failure**

**Hennie de Bruyn (Borealis Group)**



# Unusual CUI Failure

Hennie de Bruyn – Chief Engineer Material Technology



© 2008 Borealis AG

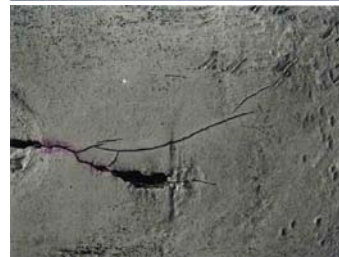
## Nitrate SCC of Carbon Steel

- Benfield (CO<sub>2</sub> –removal) unit

- Linz – Austria
- Material: St35
- Line: DN500 / 11mm WT
- Operating at: 120°C / 30 bar
- Lean caustic potash solution

- Insulation system

- Installed in 1974
- Pb-based primer; no topcoat
- Unspecified mineral wool insulation
- No maintenance



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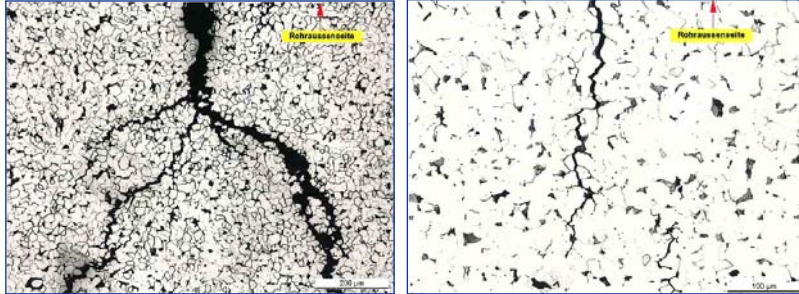
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15 April 2008

Nitrate SCC of Carbon Steel (EFC WP15 - Spring 2008 Meeting)



## Nitrate SCC of Carbon Steel



- Failure mode

- Intergranular cracking that initiated from the external surface
- Cracking occurred away from any welds or other structural discontinuities
- Main crack direction is longitudinal to the pipe
- Wet chemical analysis from surface debris indicate high nitrate levels

## Nitrate SCC of Carbon Steel

- Investigation

- Direct nitrate source not found
- Old insulation material contains high levels of leachable nitrate
- Possible atmospheric contamination from nearby nitric acid plant

- Mitigation

- Extensive insulation removal program + inspection
- Considering painting program / alternatively TSA

- Key questions

- Has anyone else experienced similar failures?
- Main reason: nitrate containing insulation or external contamination?
- Experienced TSA applicators in Austria?
- What about Al-foil wrapping? Sufficient electrochemical potential shift?

## **Appendix 5**

**Assessment of the resistance to HTHA of a 5%**

**Mo steel equipment**

**Martin Richez (Total)**

# Assessment of the resistance to HTHA of a 0.5 % Mo steel equipment

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EFC – WP15

# Presentation

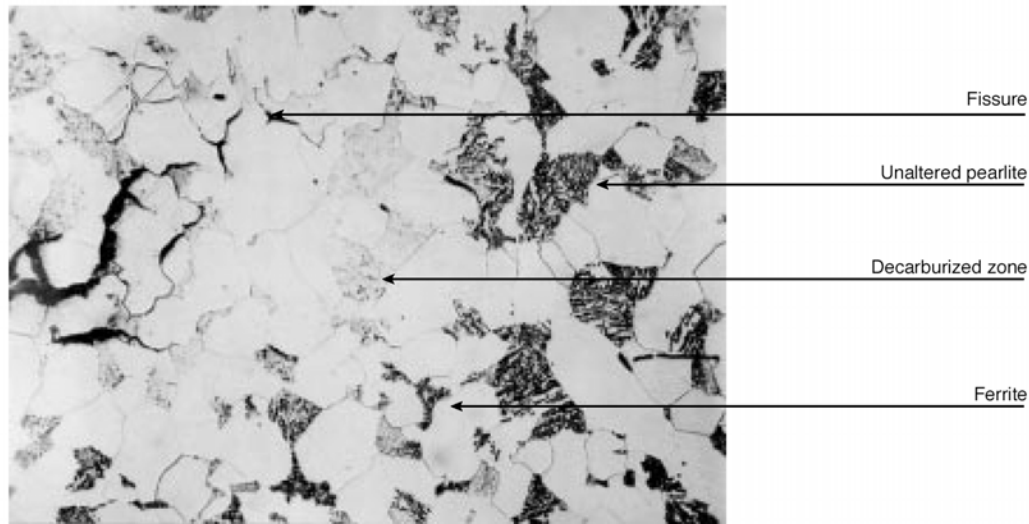
- ▶ **HTHA mechanism**
- ▶ **Failure cases**
- ▶ **Resistance of 0.5 % Mo steel to HTHA (Hot Temperature Hydrogen Attack)**
- ▶ **Management of 0.5 % Mo equipments**
- ▶ **Assessment of an equipment by direct evaluation of metal resistance to HTHA**

# HTHA Mechanism

- ▶ **Hot Temperature (temp > 220°C)**
- ▶ **Concerned units in refining :**
  - **Hydrotreatments (HDT, HDS, Hydrocracker...) – Pressure from 30 to 170 bars, temp. from 350 to 430 °C**
  - **Les reforming – Pressure from 4 to 35 bars, temp. from 490 à 550 °C**
  - **Steam reforming (hydrogene production) 40 bars, 850 °C**
- ▶ **2 different types of attack :**
  - **Surface decarburation**
  - **Internal decarburation**

# Décarburation interne

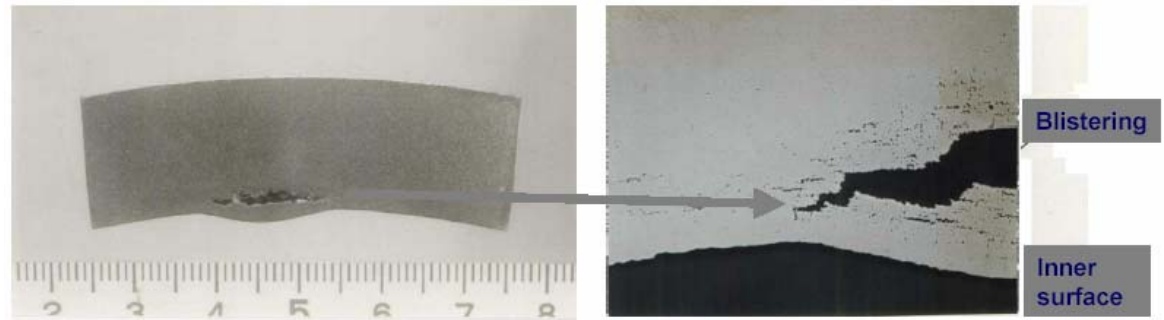
- ▶ Hydrogen ingress penetration in the metal
- ▶ Reaction with carbon to form methane  $\text{CH}_4$
- ▶ Trapping of methane dans les discontinuities and at grain boundaries
- ▶ Internal stresses leads progressively to micro void formation, then micro cracks and cracks.
- ▶ Resistance to hydrogen attack is linked to carbide stability.





# HTHA exemples

# Exemples d'attaque à chaud



Cross section of HTHA elbow

Figure 2 Blistering observed on C-0.5Mo nozzle flange used in Platformer unit

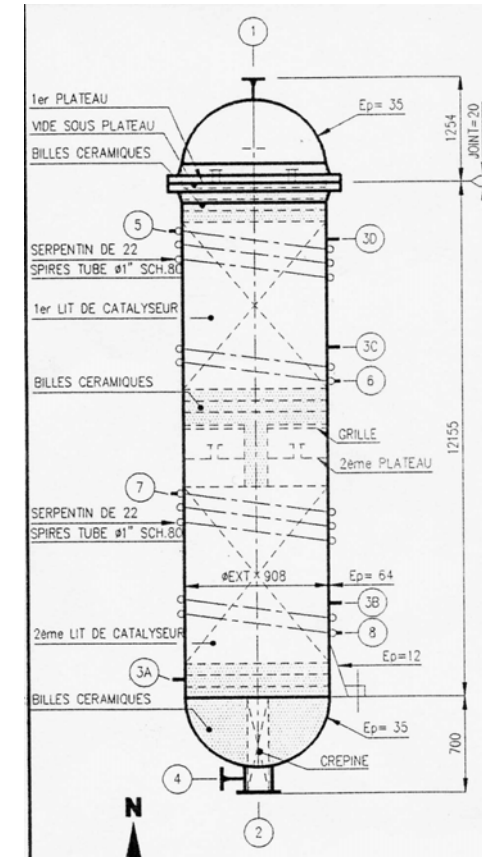
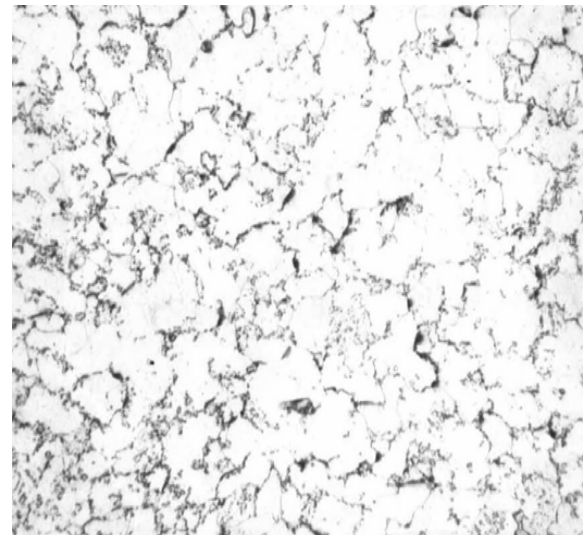
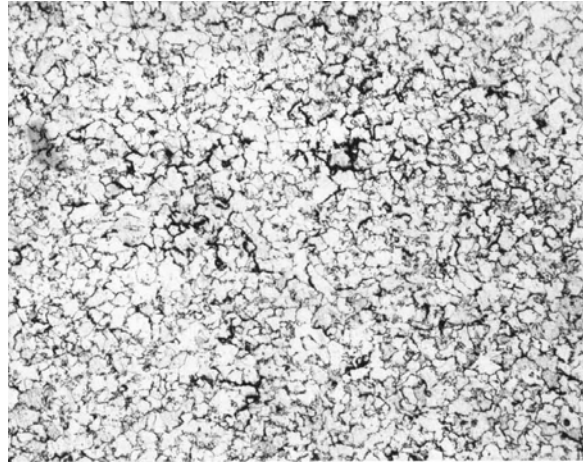
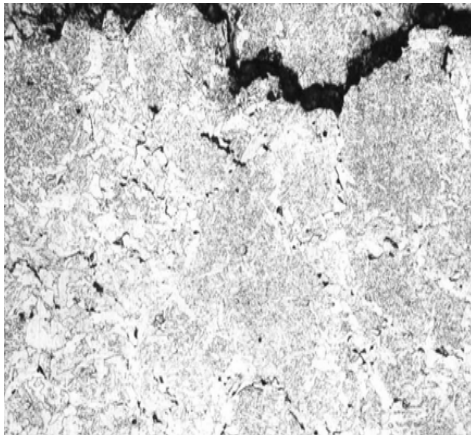
# Défaillance du à l'attaque à chaud



Réformeur 1989

# Exemple d'attaque dans le raffinage TOTAL (2004)

- ▶ Réacteur d'hydrofinissage d'huile
- ▶ A 204 Gr B (0,5 % Mo), ép. 64 mm
- ▶ 260 à 350 °C
- ▶ ppH2 78 à 136 bars





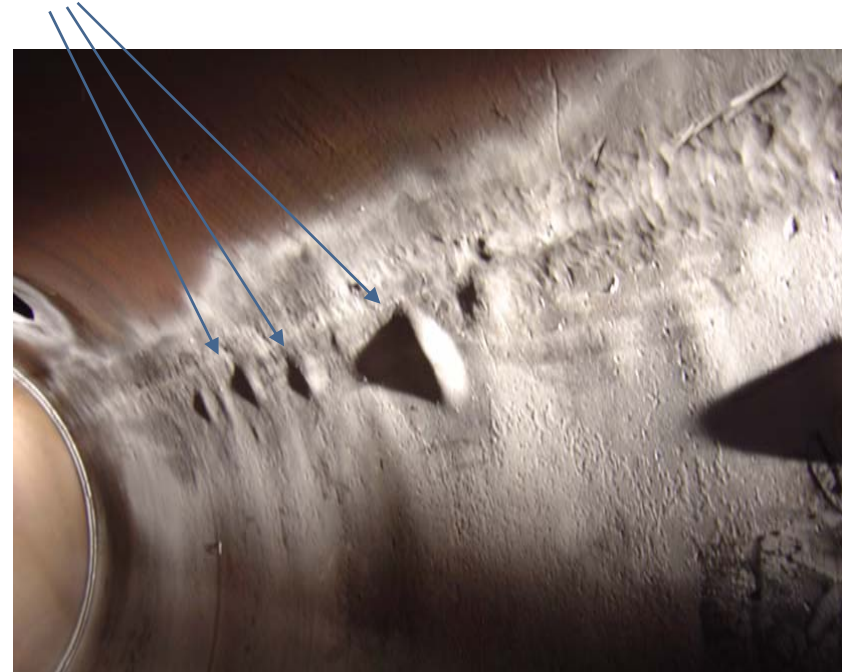
# Exemple d'attaque dans le raffinage TOTAL (2005)

- ▶ Fabrication 1967 – les 4 plus chauds: calandre A204 Gr.B (0,5 Mo) – TTAS – épaisseur 28mm
- ▶ Conditions de service pour les 4 plus chauds:
  - Température de service sortie calandre 420 à 450°C
  - Pression partielle d'hydrogène 12 à 18 bars
  - Durée d'exploitation 330 000 heures

- Fissures longitudinales le long soudure L2 sur calandre E5J



- Hernies  $\phi$  80/100 mm le long soudure L2 sur calandre E5E



# ATOCHEM Carling (2007)

- ▶ fuite d'un mélange d'hydrogène / hydrocarbures
- ▶ Une soudure circulaire de la tuyauterie 4'' (DN100) ép 6 mm ASTM A53 Gr.A est fissurée sur  $\frac{3}{4}$  de sa section (fissure débouchante sur toute sa longueur).
- ▶ Condition de fonctionnement : 40 bar – # 300°C – mélange H<sub>2</sub> + hydrocarbures ( 50% H<sub>2</sub> molaire pour P totale 40 bar)
- ▶ Mise en service de la tuyauterie: 1968



# Décarburation interne

- ▶ Elements like chromium, molybdenum, titanium, or niobium... are beneficial to resistance to HTHA by forming stable carbides.
- ▶ 0.5 % Mo non sufficient to stabilize carbides
- ▶ HTHA happen in 3 phases :
  - Incubation time, during which metallurgical evolutions are not detectables
  - Degradation phase, during which mechanical properties are decreasing
  - Then properties reach their final value (depletion of carbon)

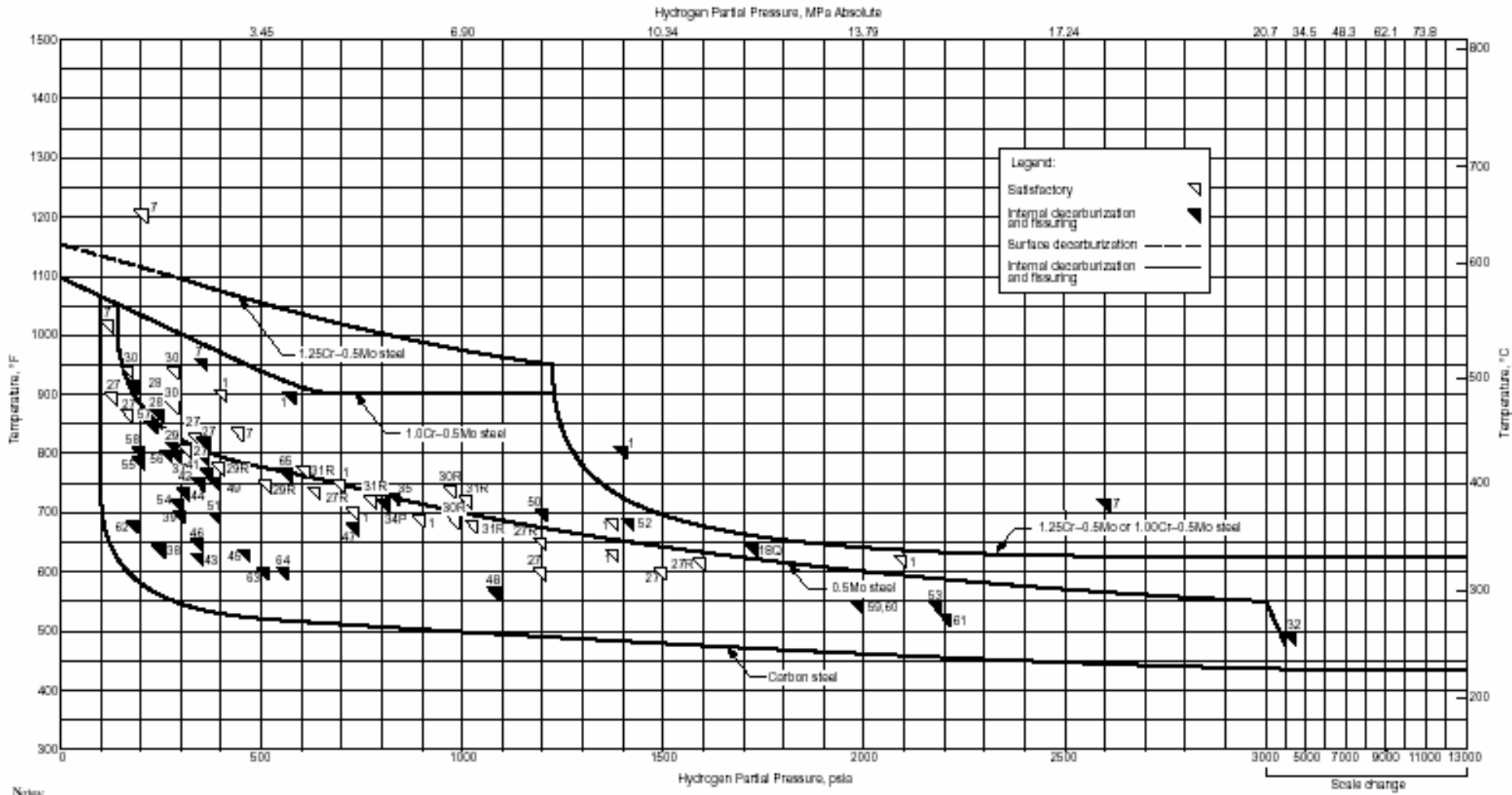
# Courbes de Nelson



# Courbes de Nelson

- ▶ Ces courbes empiriques représentent l'expérience de l'industrie du raffinage
- ▶ Premières courbes publiées par George NELSON dans les années 40
- ▶ Première publication dans l'API 941 en 1969
- ▶ Les aciers à 0,5 % de Mo sont abaissés de 33°C en 1977 suite à des expériences défavorables
- ▶ Dans l'édition de 1983, une note particulière est émise concernant des cas de HTHA sur des réformeurs catalytiques.
- ▶ De nouveau cas d'attaque sur l'acier 0,5 % Mo sont observés jusqu'à 110 °C en dessous de la courbe publié en 1977
- ▶ En 1990 les courbes concernant l'acier à 0,5 % Mo sont retirées
- ▶ L'expérience semble montrer que l'acier à 0,5 % Mo a une résistance très variable

# Courbes de Nelson publiées en 1977



Notes:  
1. The curves are based on the data of Nelson et al. (1977).



# Résistance des aciers contenant 0,5 % de Mo

# Facteurs influençant la tenue à l'hydrogène

- ▶ Temps d'exposition
- ▶ Température et Pression (ppH<sub>2</sub>)
- ▶ Les méthodes de production (histoire thermique)
- ▶ La présence ou non d'une barrière à la diffusion de l'hydrogène
- ▶ Le niveau de contrainte

# Incidence d'un revêtement métallique

- Les aciers ferritiques ont un fort coefficient de diffusion de H<sub>2</sub> mais une faible solubilité, comparés aux aciers austénitiques
- Les revêtements en aciers austénitiques sont un obstacle à la diffusion de l'hydrogène dans l'acier. L'efficacité de cette barrière dépend :
  - L'épaisseur relative d'acier austénitique et ferritique
  - Du bon état du revêtement
  - De la température
- Les aciers ferritiques type 410, n'assurent aucune protection
- La pression effective sur la paroi peut être calculée par la formule suivante :

$$P_{Interface} = \frac{P_0}{\left[ 1 + 0.18 \frac{t_{Clad}}{t_{Base}} \exp\left(\frac{2984}{T}\right) \right]^2}$$

Exemple :

temperature	Thickness		PPH2	PPH2 Interface
	austenitic	Ferritic		
°C	mm	mm	bar	bar
390	2	70	35	16.3
390	4	70	35	9.4
390	3	20	35	3.0
300	3	20	35	1.0

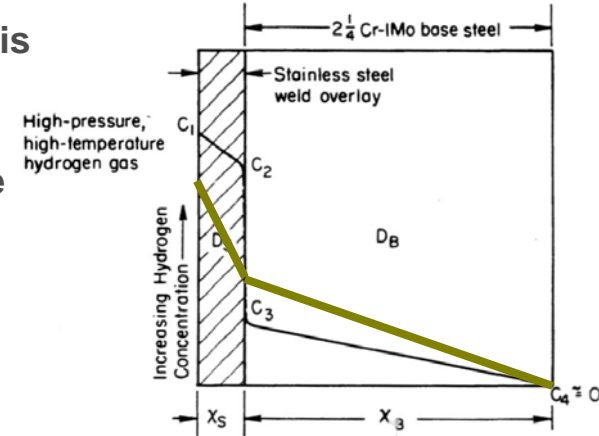
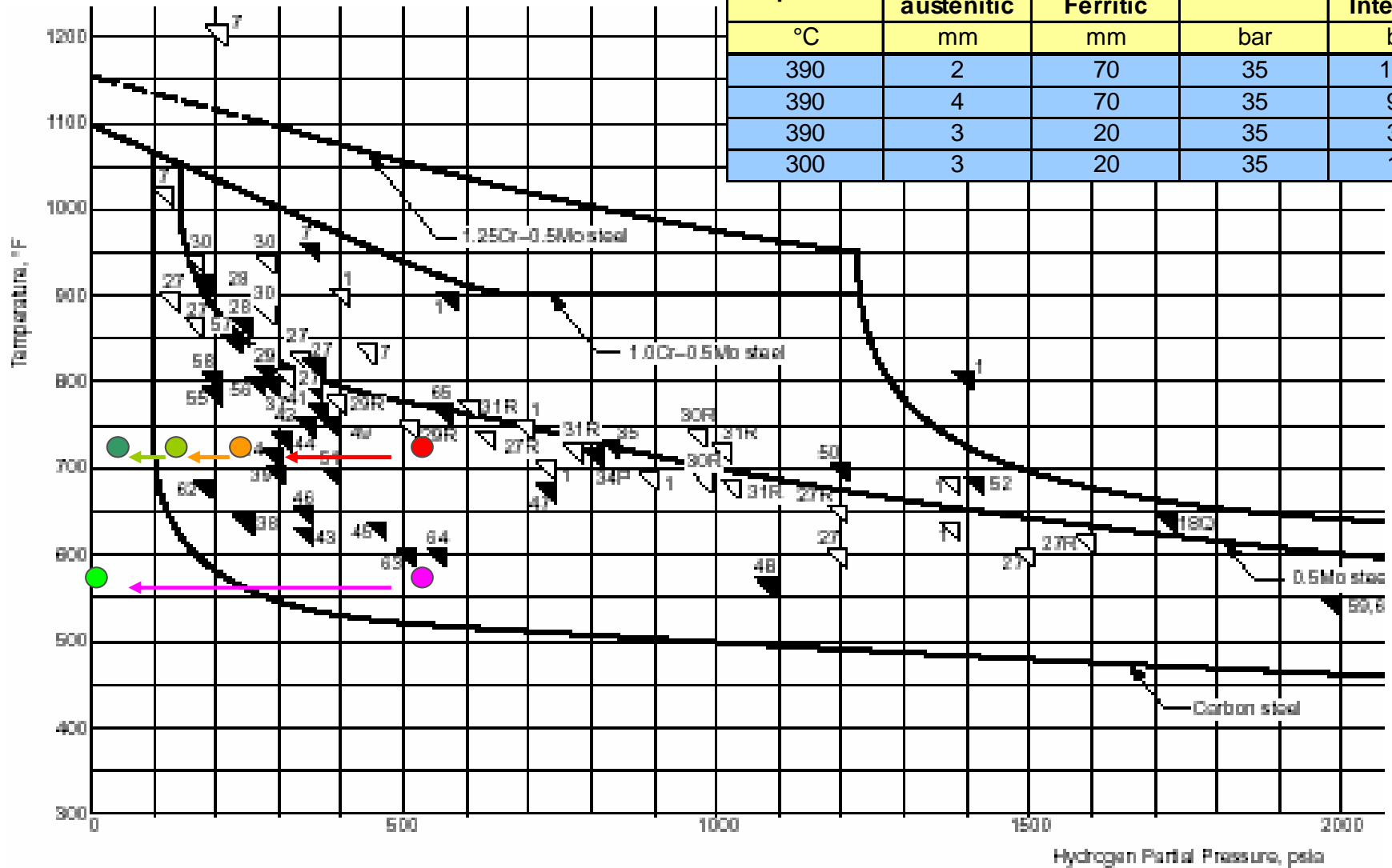


FIGURE 4. SKETCH SHOWING HYDROGEN DISTRIBUTION IN A STAINLESS STEEL-WELD-OVERLAYED 2-1/4Cr-1Mo STEEL VESSEL EXPOSED TO HIGH-PRESSURE, HIGH-TEMPERATURE HYDROGEN

# Incidence d'un revêtement métallique

temperature °C	Thickness		PPH2 bar	PPH2 Interface bar
	austenitic mm	Ferritic mm		
390	2	70	35	16.3
390	4	70	35	9.4
390	3	20	35	3.0
300	3	20	35	1.0



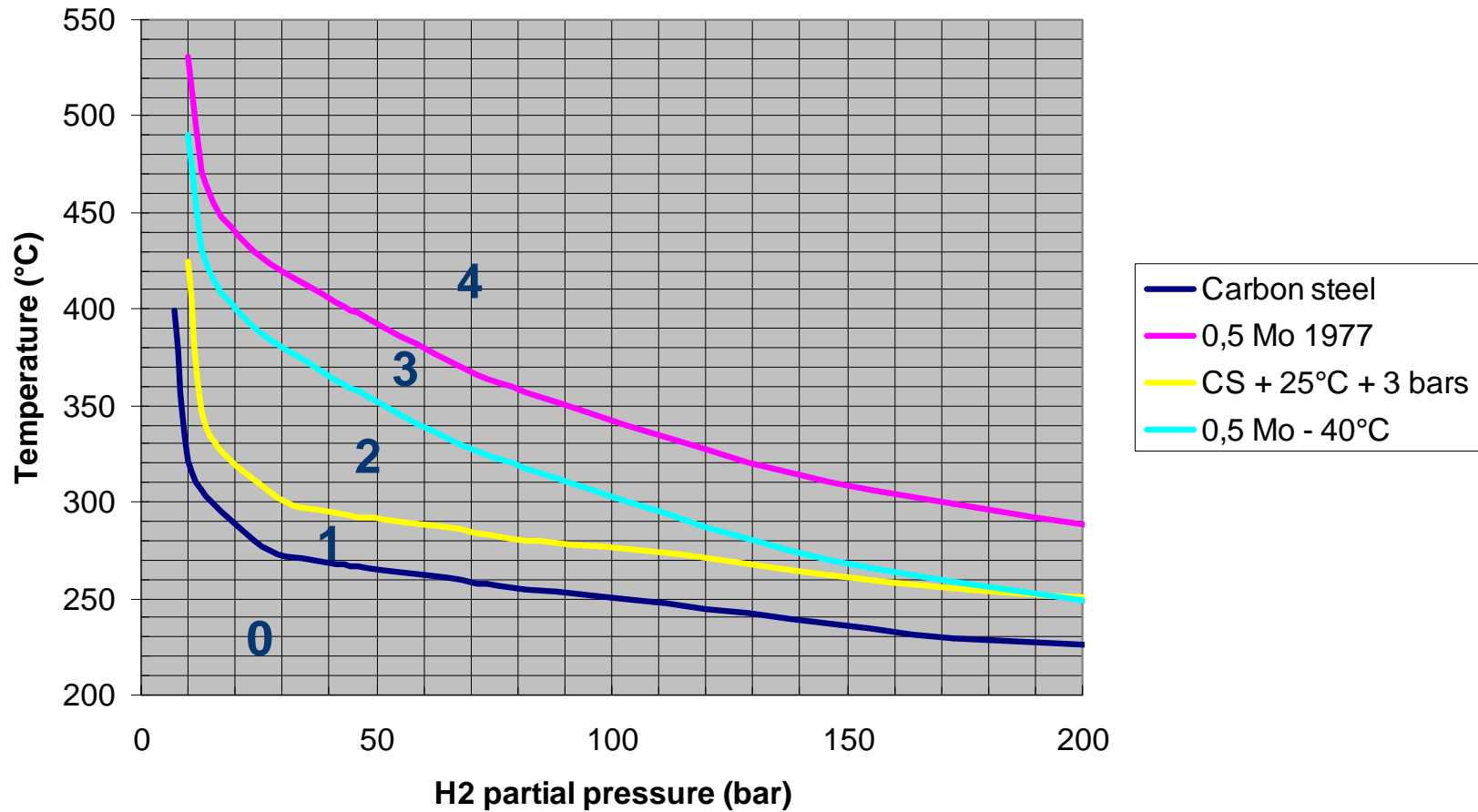
Notes:



# Management of 0.5 % Mo equipment

# Zones de risque

## Severity area for HTHA of 0,5 Mo



# Risk Area

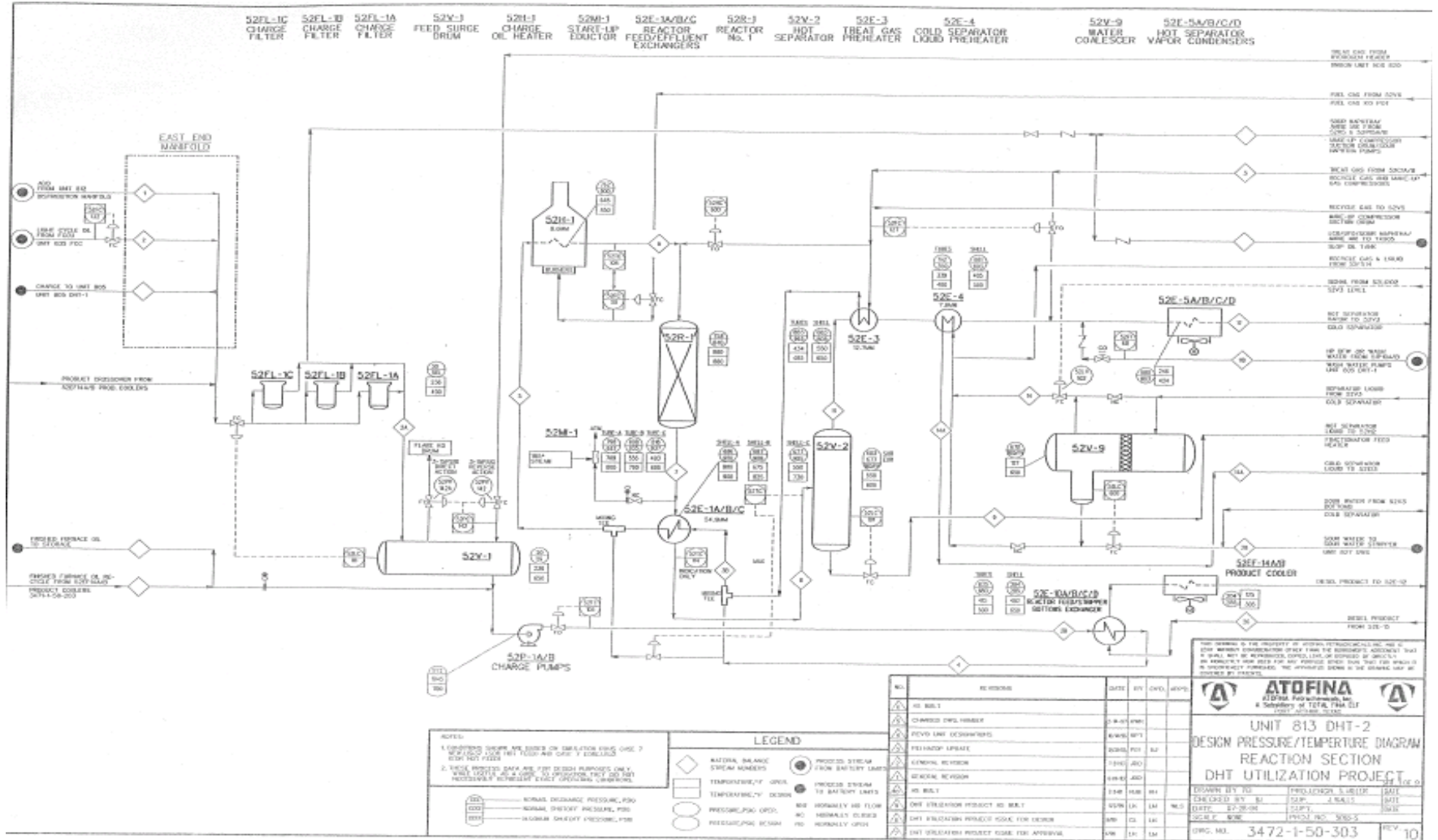
- ▶ **Area 0** – Below carbon steel curve. Degradation probability is considered as null
- ▶ **Area 1** – Between carbon steel curve and carbon steel + 3 bars and 25 °C. In this area no failure case on 0.5 % Mo steel has been reported. HTHA risk is considered as very low.
- ▶ **Area 2** – It is an intermediate area where risk is considered as low for equipment made of normalized steel and PWHT.
- ▶ **Area 3** – Between 1977 0.5 % Mo steel curve and 40°C below. In this area degradation risk is considered as high on old.
- ▶ **Area 4** – Located above 0.5 % Mo curve of 1977. This is an area where degradation risks are very high and unacceptable.

Secteur	Inspection HTHA	Remplacement	Sévèrisation des conditions procédé
Secteur 0	Néant	Néant	<p>Sévèrisation des conditions procédé autorisée si :</p> <ul style="list-style-type: none"> <li>- L'équipement est en zone 1 après prise en compte du clad.</li> <li>- Le clad est en bon état</li> <li>- L'appareil est traité thermiquement s'il se retrouve en zone 1.</li> </ul>
Secteur 1	<ul style="list-style-type: none"> <li>- Partielle si l'appareil est traité thermiquement.</li> <li>- Complète si l'appareil n'est pas traité thermiquement.</li> </ul>	<ul style="list-style-type: none"> <li>- En cas de détection d'HTHA et,</li> <li>- Systématique si tuyauteries</li> </ul>	<p>Sévèrisation des conditions procédé autorisée si :</p> <ul style="list-style-type: none"> <li>- L'équipement est en zone 1 après prise en compte du clad.</li> <li>- Le clad est en bon état</li> <li>- L'appareil est traité thermiquement</li> </ul>
Secteur 2	Complète	<ul style="list-style-type: none"> <li>- En cas de détection d'HTHA ou,</li> <li>- Si l'équipement n'est pas traité thermiquement ou,</li> <li>- Tuyauteries ou équipements peu inspectables</li> </ul>	<p>Sévèrisation des conditions procédé non autorisée sauf si :</p> <ul style="list-style-type: none"> <li>- L'équipement est en zone 1 après prise en compte du clad.</li> <li>- Le clad est en bon état.</li> </ul>
Secteur 3	Complète	<ul style="list-style-type: none"> <li>- En cas d'HTHA ou,</li> <li>- Si l'équipement n'est pas traité thermiquement ou,</li> <li>- Si le clad est détérioré ou,</li> <li>- En l'absence de clad protecteur ou,</li> <li>- Tuyauteries ou peu inspectable</li> </ul>	Sévèrisation des conditions procédé non autorisée.
Secteur 4	Néant (appareil à remplacer)	- A remplacer systématiquement et dans les délais les plus bref.	Sans objet (appareil à remplacer)

# Inspection for High Temperature Hydrogen Attack on a Carbon 0.5% Molybdenum Vessel

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# Process Service





# 52V-2 Hot Separator



# Vessel Specifics

- ▶ Built in 1962 - 45 years in Hydrotreater Hot Separator Service
- ▶ Design Pressure 803 PSI
- ▶ Operating Hydrogen Partial Pressure 587 psia
- ▶ Design Temperature 600 degrees F
- ▶ Operating Temperature 595 degrees F
- ▶ Shell 6' Diameter Height 21'1"
- ▶ Shell Material A204 grade C with .078" TP 410 Clad

# Scope

- ▶ To evaluate the future performance of 52V-2 , Lloyd's Register Capstone performed HTHA exposure testing on Scoop Samples removed from the OD of the vessel. Two scoop samples were removed from 4 locations. These locations consisted of the top and bottom heads and each of the two shell courses. Two test coupons were made from each scoop sample. One coupon per location was subject to a 100 hour test while the other coupon was subjected to a 200 hour test. This left one full Scoop Sample per location for additional testing as needed.
- ▶ L.R. Capstone also performed internal metallurgical examination using field metallography and replication, (FMR).
- ▶ Advanced Ultrasonic Backscatter Techniques (AUBT) were used to examine the base metal and weld joints were examined using angle beam spectrum analysis (ABSA).

# Test Procedure for Scoop Samples

Figure 1 and 2 show the Scoop Samples that were removed from the vessel OD surface. These were removed using LR Capstone's proprietary Scoop Sampler<sup>SM</sup>. Depth of the resultant divots were less than the maximum allowable based on local thin area calculations that were previously performed in the 1999 evaluation. Therefore, the excavations were blended to minimize local stress concentrations with no further repairs needed.

It can be seen that a total of eight Scoop Samples were removed from the vessel. Two Scoop Samples were removed from each of four locations; two test coupons were able to be removed from a Scoop Sample. One coupon per location was subjected to a 100 hour test while the other coupon was subjected to a 200 hour test. This left one full Scoop Sample per location for additional testing if needed. Sample locations are given below in Table 1.

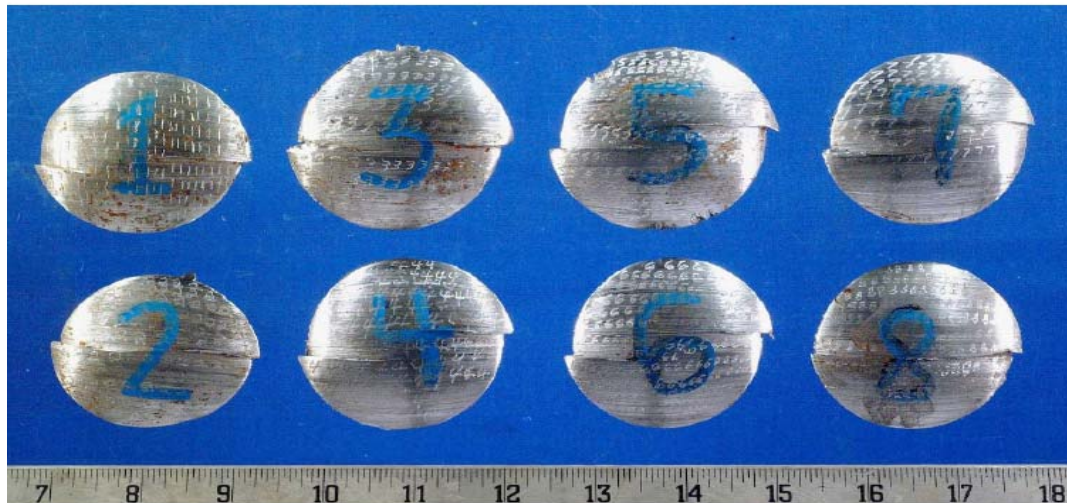
Table 1. Scoop Sample Locations

	Scoop Sample ID
Bottom Head	1 and 2
Bottom Course	3 and 4
Top Course	5 and 6
Top Head	7 and 8

# Scoop Samples



Figure 1. OD surfaces of the removed Scoop Samples.



# Test Procedure Cont.

The coupons were welded around their perimeter to seal them in place at the end of the test chamber, which had hydrogen inside and argon on the OD. The single side exposure testing was performed by exposing only one side of the coupon to high pressure hydrogen while the argon on the other side was set at a pressure equal to the hydrogen pressure. This balanced pressure approach minimized the potential for a leak or crack of the coupon to hydrogen chamber seal weld. Thermocouples were placed against the test coupon in the hydrogen chamber to monitor temperature during the test. Test temperature was set at 950 degrees F at a hydrogen pressure of 1215 PSIA.



# Carbon Steel Reference Sample

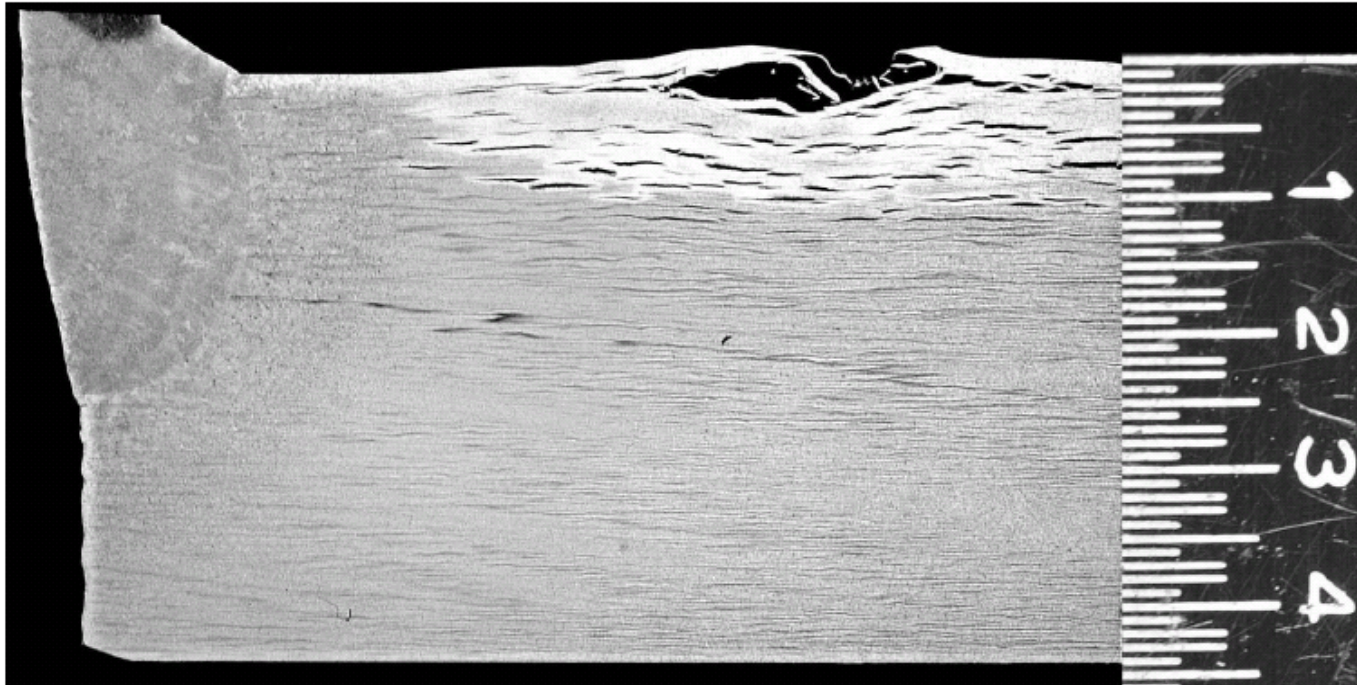


Figure 4: Macroscopic view of a carbon steel reference sample at the accelerated test conditions, 100 hour exposure. Decarburization depth is approximately 0.1" (23% through wall). Depth of cracking was approximately 0.13" (30% through wall). 3% Nital Etch, 6.7X.

## Nelson Curve

Included in this figure are the 100 hour incipient damage curves for carbon steel and C- 0.5Mo steel and past and future operating conditions. It can be seen that the test conditions are very severe and if susceptible they should have produced damage in the test coupons.

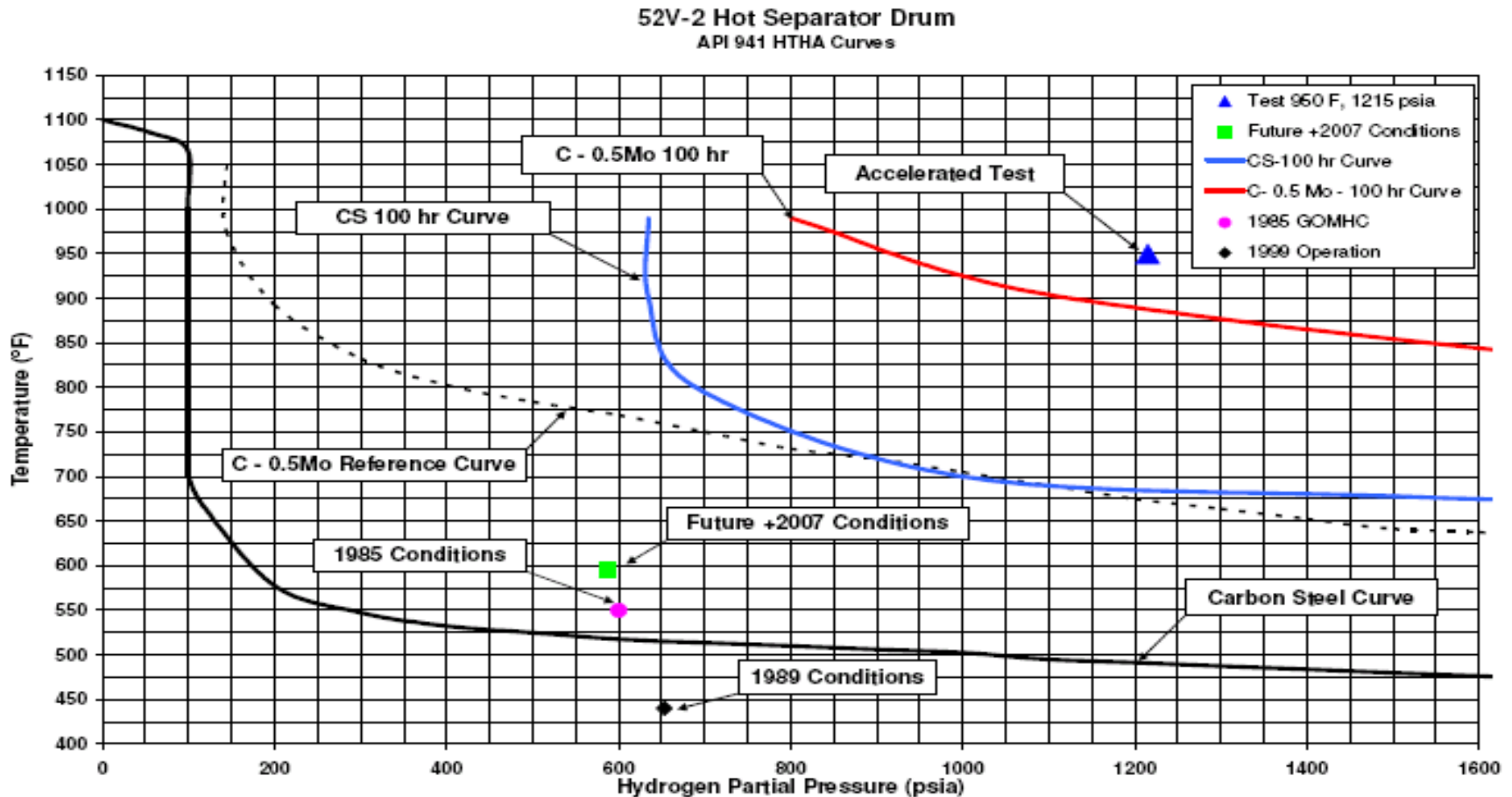


Figure 3: HTHA damage curves from API 941. Past to future operating conditions plus the accelerated test conditions are plotted to show severity of the various operating conditions.

# Pre-Exposure Microscopic View

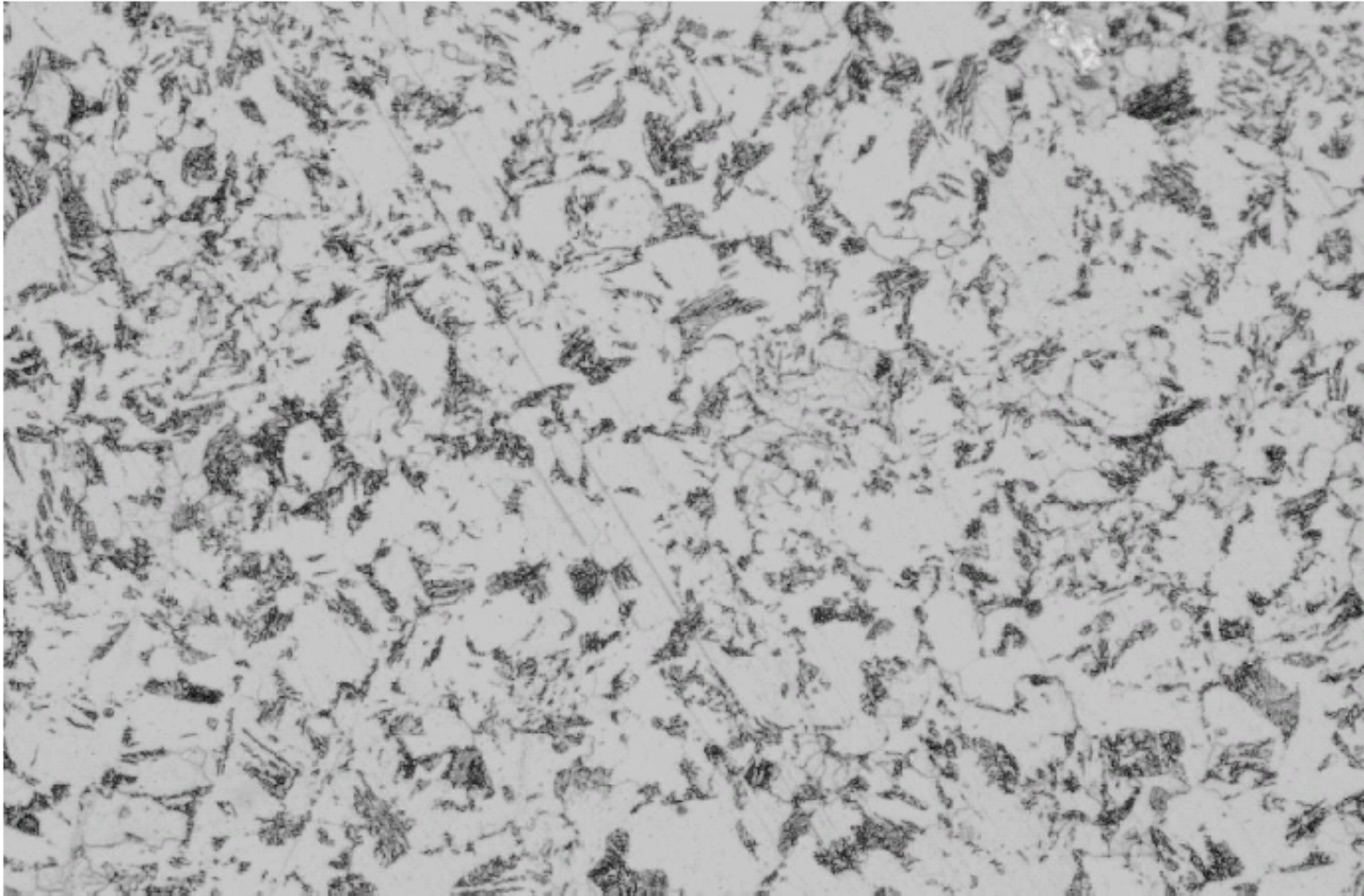


Figure 5. Scoop Sample 6, representative base metal microstructure, pre-exposure. 3% Nital Etch, 400X.



## Results:

No evidence of decarburization, methane voids, or microfissuring was found in any of the samples.

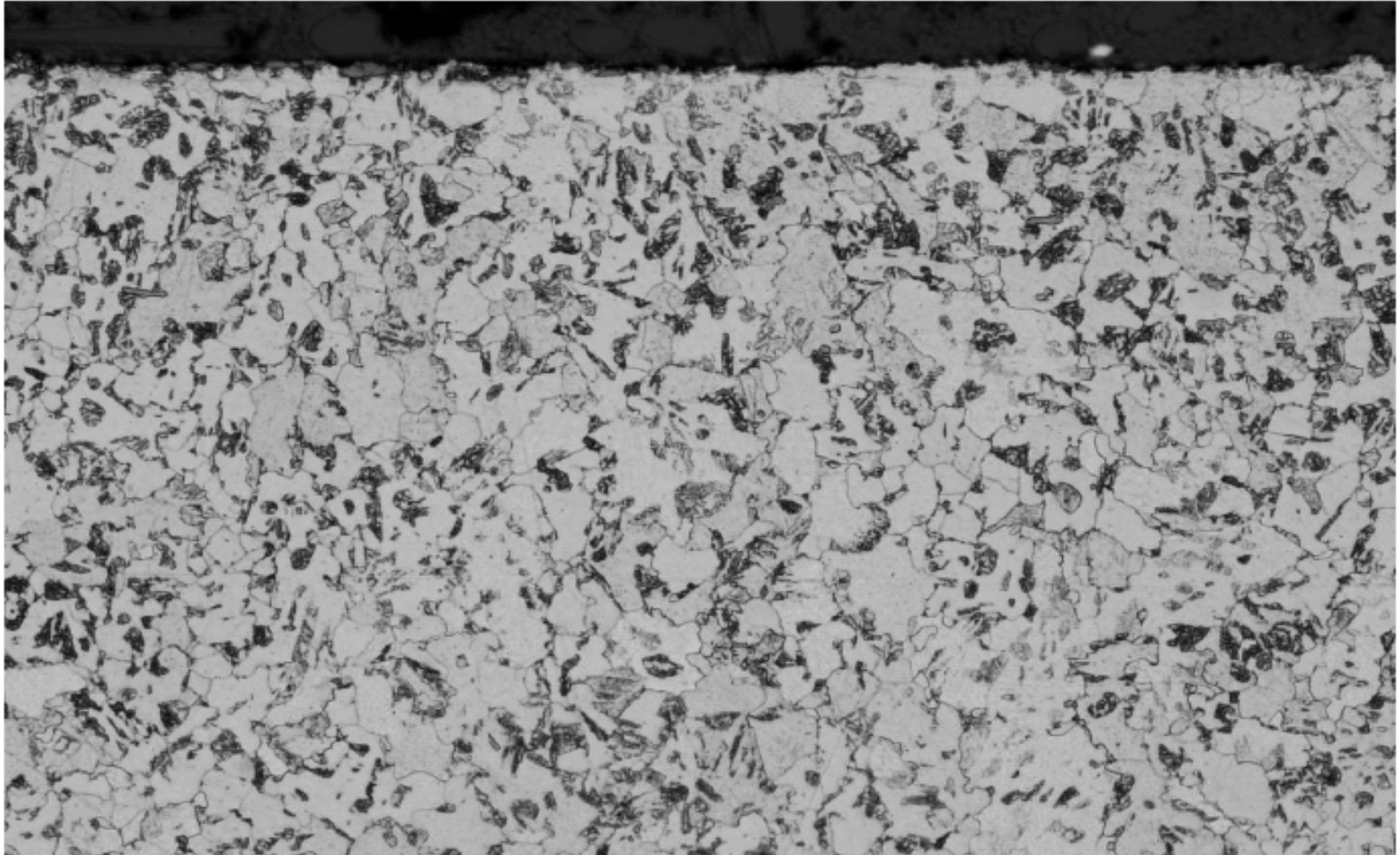


Figure 6: Scoop Sample 1, representative base metal microstructure Test 1. 3% Nital Etch, 400X.



**Results cont.**

**All of the C-0.5 Mo steel materials displayed good resistance to HTHA type of damage.**

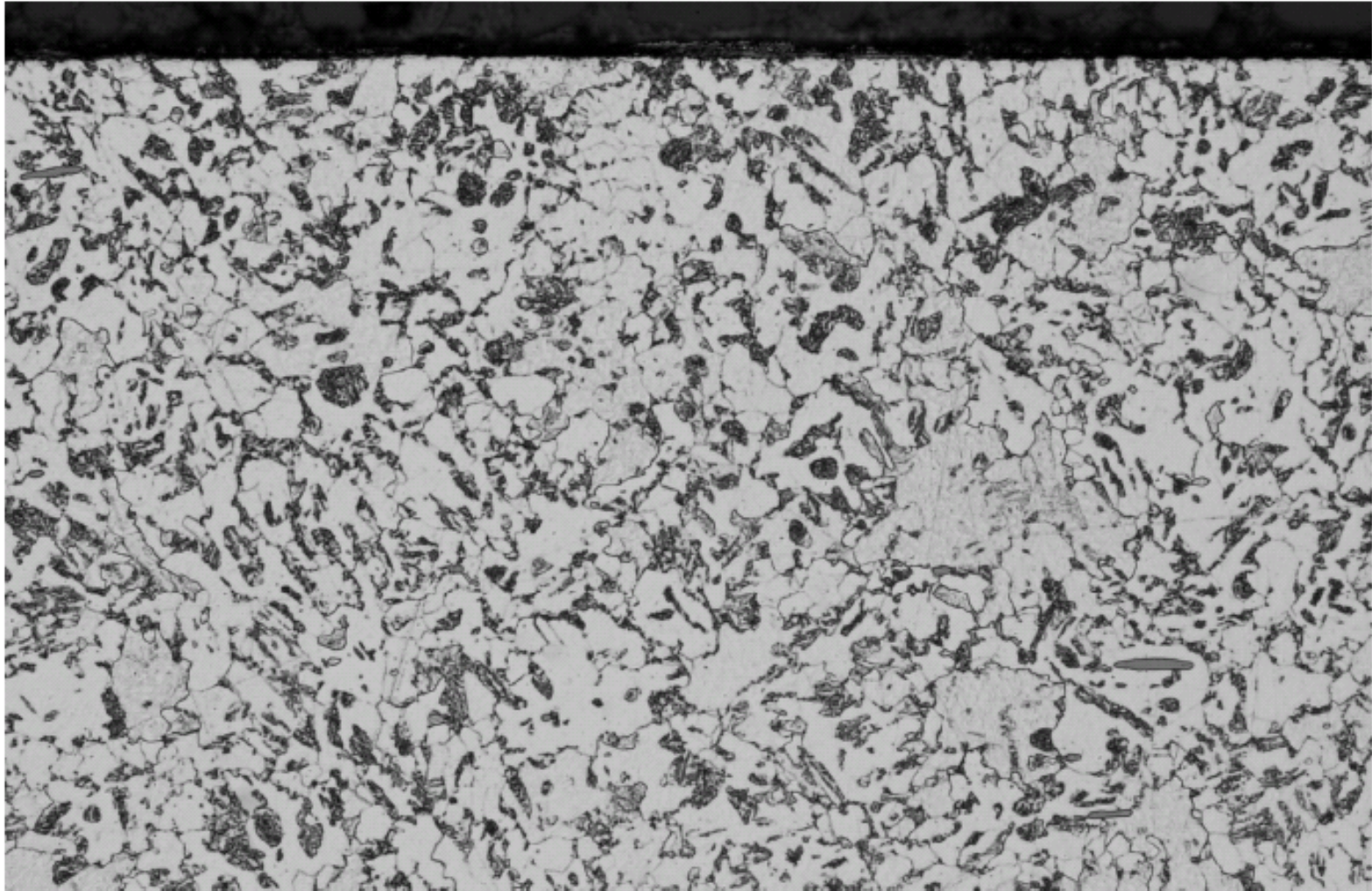


Figure 10: Scoop Sample 1, representative base metal microstructure Test 2. 3% Nital Etch, 400X

## DISCUSSION

The test conditions were set approximately 65°F above the 100 hour C - ½ Mo steel incipient damage curve. It also corresponds to the 100 hour incipient damage as predicted using both  $P_v$  and  $P_w$  calculations. The equations used to calculate  $P_v$  is given below and are taken from the API 580 document. Critical values for  $P_v$  "worst case" annealed material is low susceptibility at 4.78, medium susceptible at  $P_v = 4.87$ , and a high susceptibility at  $P_v = 4.95$ . For "good" normalized material, low susceptibility starts at a  $P_v$  of 5.43, a medium susceptibility at  $P_v$  of 5.51, and a high susceptibility at  $P_v$  of 5.60. Based on published literature and some work by Japanese researchers, the critical value for "good" C - ½ Mo steel base metal is  $P_v$  of 5.80.

$$P_v = \log (P_{H_2}) + 3.09 \times 10^{-4} T (\log t + 14)$$

$P_{H_2}$  : Hydrogen partial pressure (kgf/cm<sup>2</sup>)

t : Total operating time (hr)

T : Operating Temperature (K)

Critical  $P_v$  Factors

Materials	High Susceptibility	Medium Susceptibility	Low Susceptibility	Not Susceptible
Carbon Steel	$P_v > 4.70$	$4.61 < P_v \leq 4.70$	$4.53 < P_v \leq 4.61$	$P_v \leq 4.53$
C <sup>1/2</sup> Mo <sup>a</sup> (Annealed)	$P_v > 4.95$	$4.87 < P_v \leq 4.95$	$4.78 < P_v \leq 4.87$	$P_v \leq 4.78$
C <sup>1/2</sup> Mo <sup>a</sup> (Normalized)	$P_v > 5.60$	$5.51 < P_v \leq 5.60$	$5.43 < P_v \leq 5.51$	$P_v \leq 5.43$

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Fax: (281) 493 5161



The  $P_w$  factor has a similar logarithmic equation with slight differences and is included here for reference, where:

$$P_w = \log t + 3 \log P - 9918/T$$

JPVRC revealed in their investigation<sup>5</sup> that the critical values of  $P_w$  on HTHA of C-0.5Mo steel were the followings;

For base metal  $P_{wcr} = -4.80$

For HAZ with PWHT  $P_{wcr} = -5.25$

For HAZ without PWHT  $P_{wcr} = -7.90$

The results of the calculated  $P_v$  and  $P_w$  factors are given in Table 2. The test conditions for Test 1 (102 actual hour test) gave a calculated  $P_v$  of 5.81. It can be seen that Test 1 produced conditions that exceeded the critical  $P_v$  value of 5.80 and Test 2 exceeded the thresholds by a wide margin for both the  $P_v$  and  $P_w$  factors yet the test samples did not display any evidence of HTHA damage. Therefore, it is concluded that the 52V-2 Hot Separator Drum has HTHA resistance similar to normalized "good" C - ½ Mo steel.

Table 2. HTHA Test Results w/  $P_v$  and  $P_w$  Calculations

	T, °F	P, psia	Hours	$P_v$	$P_w$	Years
<b>Test 1 - No Damage</b>	950	1215	102	5.81	-4.86	---
<b>Test 2 - No Damage</b>	950	1215	223	5.89	-4.52	----
50°F SF	645	587	1,283,039	<b>5.43</b>	-5.20	146
50°F SF	645	587	1,150,000	5.42	<b>-5.25</b>	131
<b>Clad benefit</b>	645	549	1,825,941	<b>5.43</b>	-5.13	208
<b>Clad benefit</b>	645	549	1,400,000	5.41	<b>-5.25</b>	160

Table 2. HTHA Test Results w/  $P_v$  and  $P_w$  Calculations

	T, °F	P, psia	Hours	$P_v$	$P_w$	Years
<b>Test 1 - No Damage</b>	950	1215	102	5.81	-4.86	---
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50°F SF	645	587	1,283,039	<b>5.43</b>	-5.20	146
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<b>Clad benefit</b>	645	549	1,400,000	5.41	<b>-5.25</b>	160

The value of using  $P_v$  and  $P_w$  factors is that they can be used to evaluate different sets of operating conditions and make comparisons based on specified hours of service. Using the proposed upper temperature of 595°F plus a 50°F safety factor (645°F) and a hydrogen partial pressure of 587 psia, the drum could operate for 146 years before reaching a low susceptibility for HTHA damage based on a  $P_v$  of 5.43.

Based on a medium value susceptibility for  $P_w = -5.25$ , the drum could operate for 131 years. While the benefit of the 410 stainless steel cladding on the inside of the drum is not as much as would be given for a 304 stainless steel material, however, it can be seen that at a minimum, the 410 clad provides a fully resistant layer that reduces the hydrogen partial pressure in the C - ½ Mo steel based on the ratio of clad versus base metal thickness. Assuming a 0.109" thick cladding layer and a 1.6" shell, the cladding reduces the hydrogen partial pressure from 587 psia to 549 psia. This is not as significant of a factor at high pressures, but at the operating pressures given, there is an increase in remaining life of approximately 20 - 40%. A less conservative estimate of the pressure reduction associated with the cladding material is prescribed in the API 941 committee Base Resource Document, which indicates a further reduction of hydrogen partial

pressure to approximately 352 psia. The remaining life according to this pressure would be on the order of 2000 years.

While the majority of the vessel is clad, there are six locations where the cladding has been removed. The follow-up field metallography and replication (FMR) inspection in March 2007 repeated the examinations at the previous locations. Upon removal of the internal scab plates, the exposed base metal was found to have a thick layer of black scale, presumed to be FeS. No significant wall loss occurred at these locations, in fact, previous grind marks were still visible in the base metal around the perimeter of the clad removal area. Field metallography and the AUBT inspection found no evidence of HTHA damage.

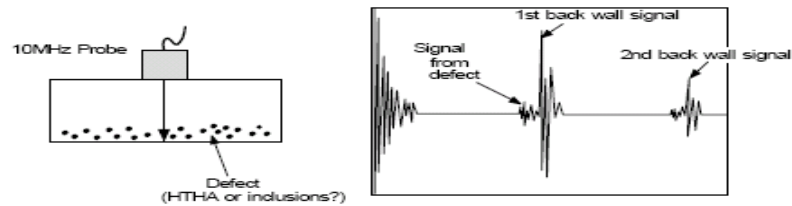
It appears that the scab plates created a stagnant zone where sulfur was able to react with the exposed base metal resulting in a stable FeS scale. It is known that FeS can provide some protection against HTHA damage, as noted in API 941. Therefore, it appears that the scab patches were successful in fostering a stable FeS scale thus avoiding direct hydrogen and base metal contact as would occur if the scab patches did not have a gap in the weld.

# Ultrasonic Inspection

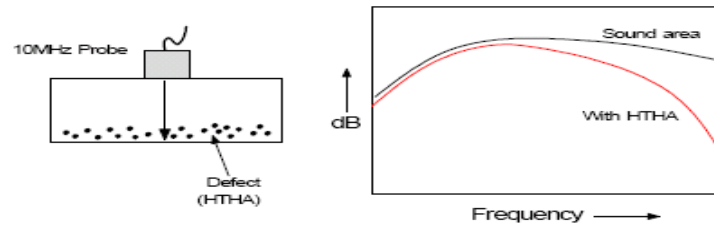
- ▶ **Seven locations were inspected from the outside and inside surfaces using several techniques at approximately 3" X 3" windows in the insulation.**
- ▶ **From the OD, Advanced Ultrasonic Backscatter Techniques (AUBT) were used to examine the base metal.**
- ▶ **Weld joints were examined using Angle Beam Spectrum Analysis (ABSA)**

# Explanation for HTHA detection by AUBT

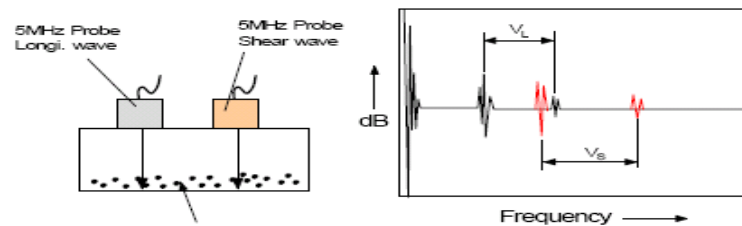
## Back-Scattering Technique



## Spectrum analysis



## Velocity Ratio Technique



Velocity Ratio =  $V_S/V_L$   
 $V_S/V_L \leq 0.55$  ----- No HTHA  
 $V_S/V_L > 0.55$  ----- HTHA



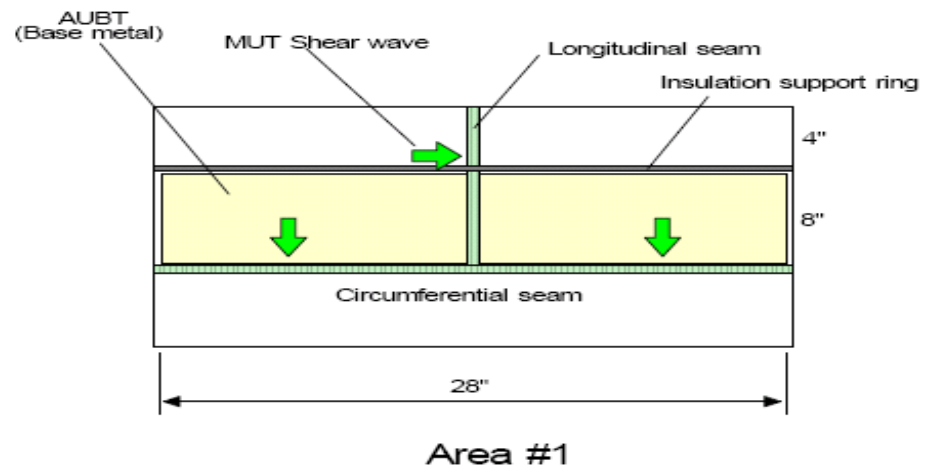
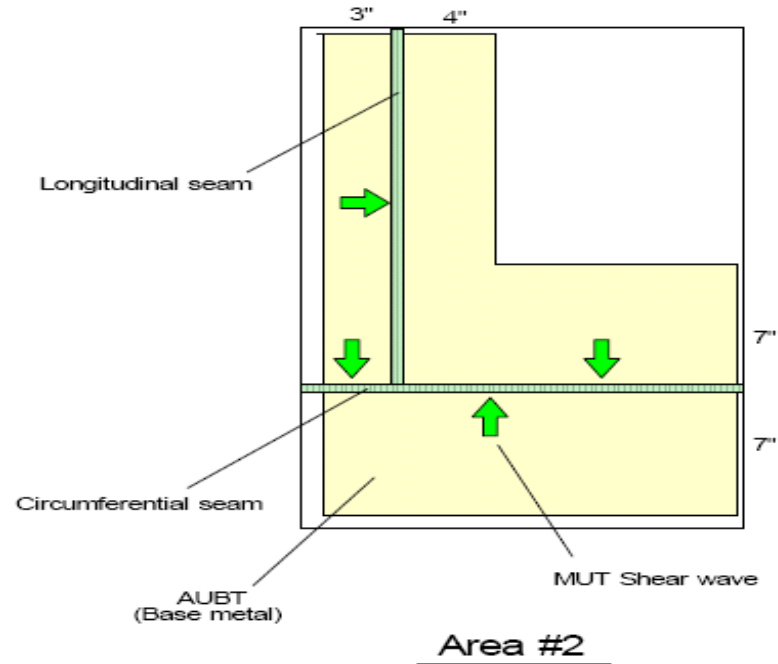
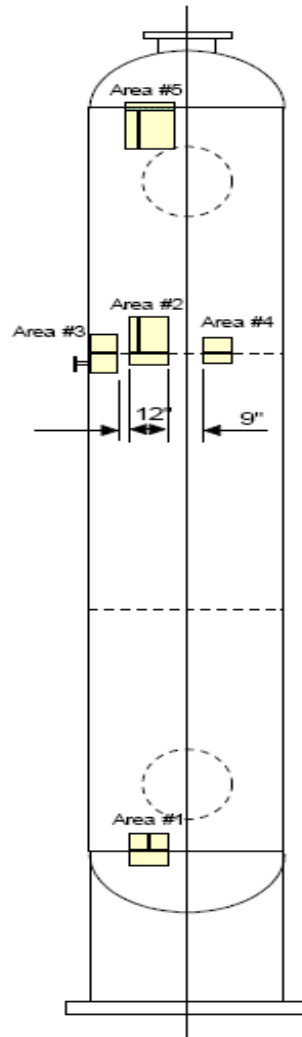
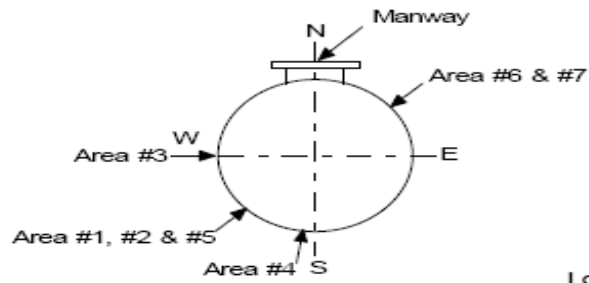
# Advanced Ultrasonic Backscatter Technique (AUBT) Inspection

## Examination Summary

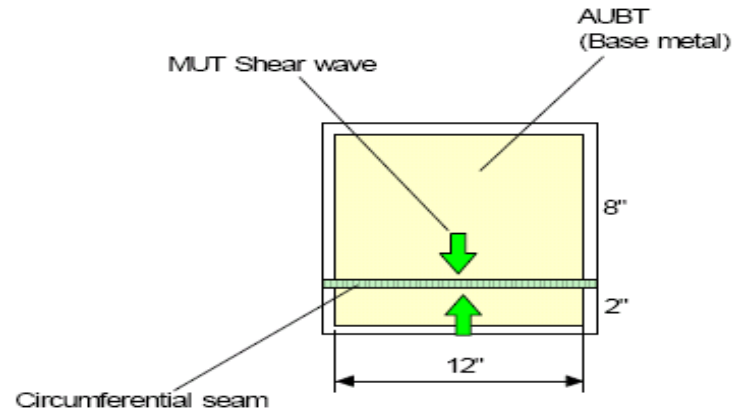
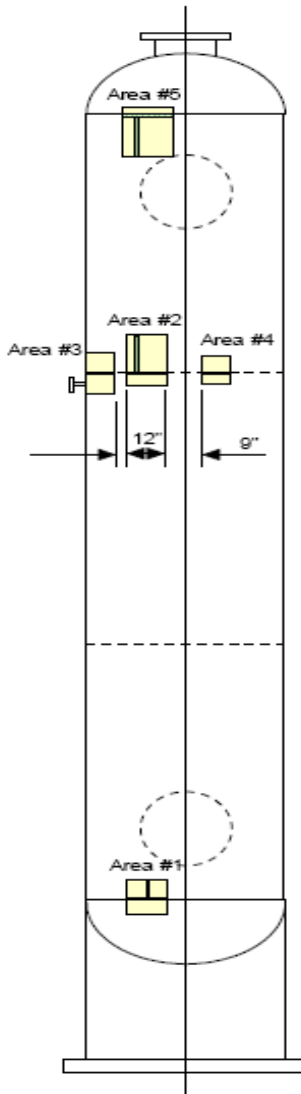
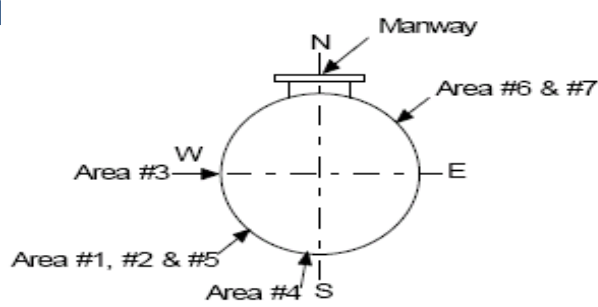
All selected areas of Reactor by the client were performed AUBT for HTHA detection by 0 degree. The inspection was performed to detect backscatter indications in the base metal. Spectrum analysis was applied using the LeCroy system if indication was detected. The probe selected was 0.50" diameter, high frequency (10MHz & 5MHz), and 0 degree to detect small indications such as micro-fisher in the base metal. Also, Angle Beam Spectrum Analysis (ABSA) was applied on the weld seam to detect crack-like indication on the Heat Affected Zone (HAZ) due to HTHA. The probe selected was 0.50" diameter, high frequency (10MHz) 45 & 60 degree shear wave probe. The purpose for this inspection is to locate and evaluate backscatter indications in the base metal, and to determine if these indications are the result of HTHA.



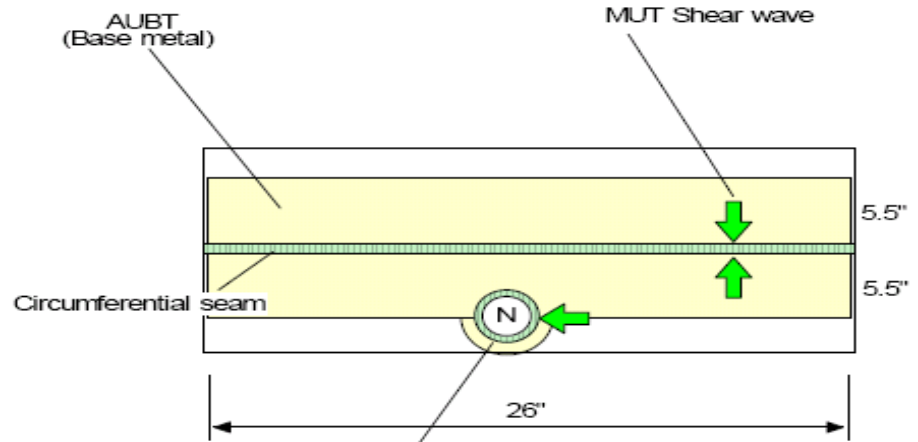
# Scan Map



# Scan Map

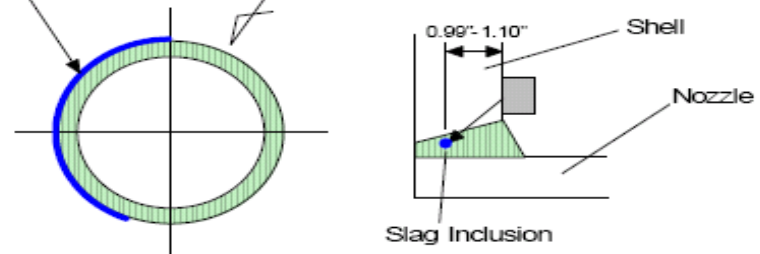


**Area #4**



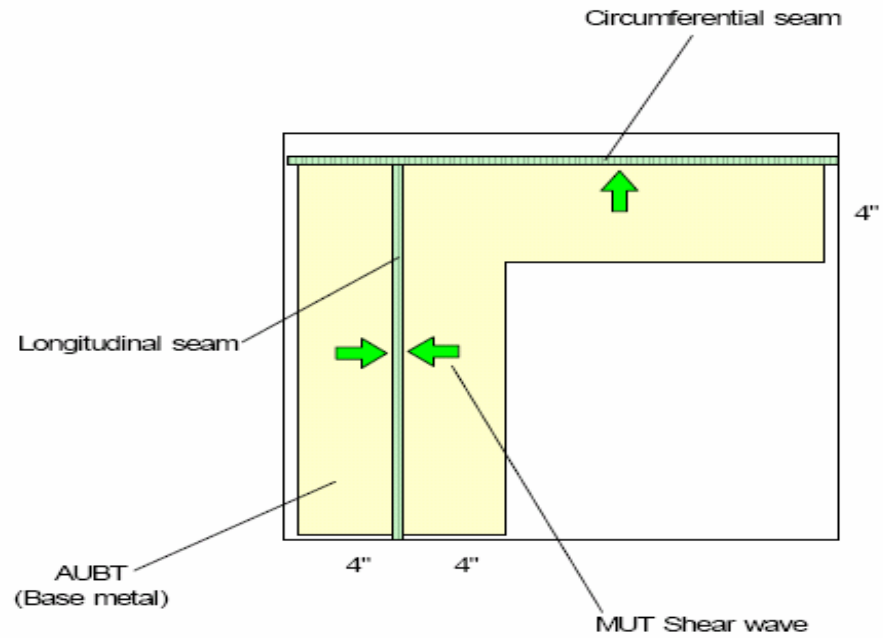
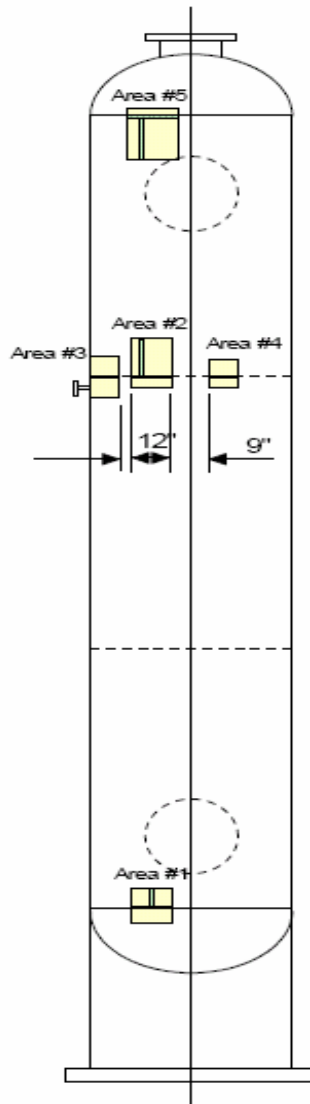
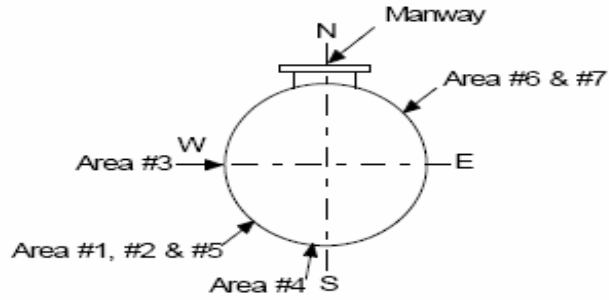
**Area #3**

Slag Inclusion  
Depth : 0.99" - 1.10"



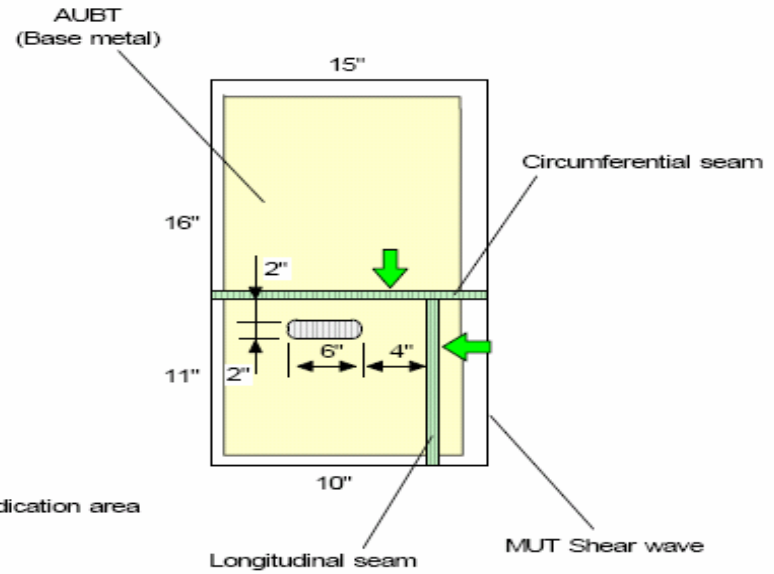
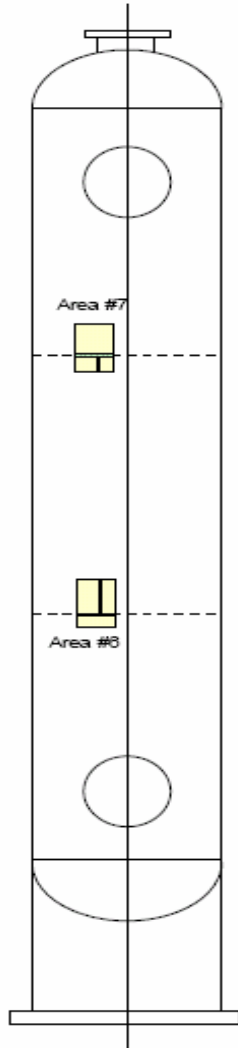
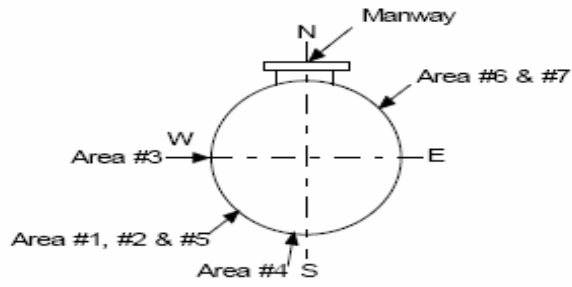
**Detail of Indication on Nozzle at Area #3**

# Scan Map



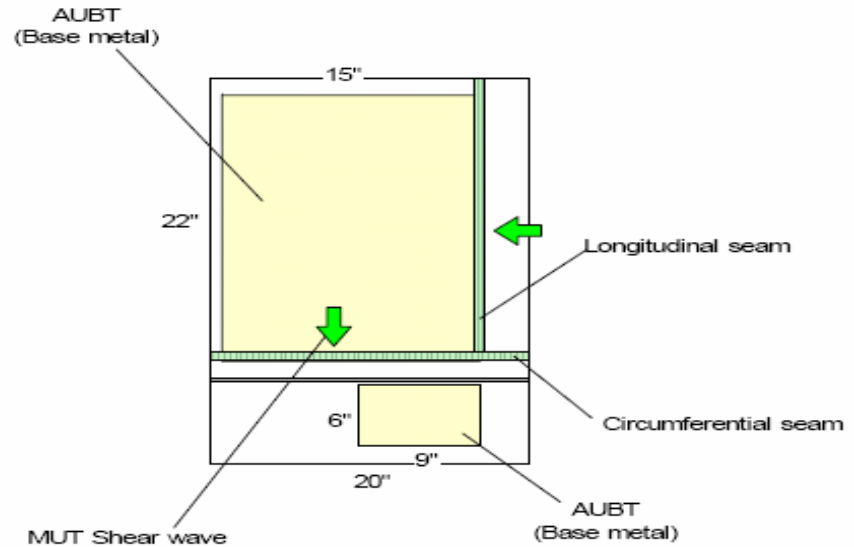
Area #5

# Scan Map



: Indication area

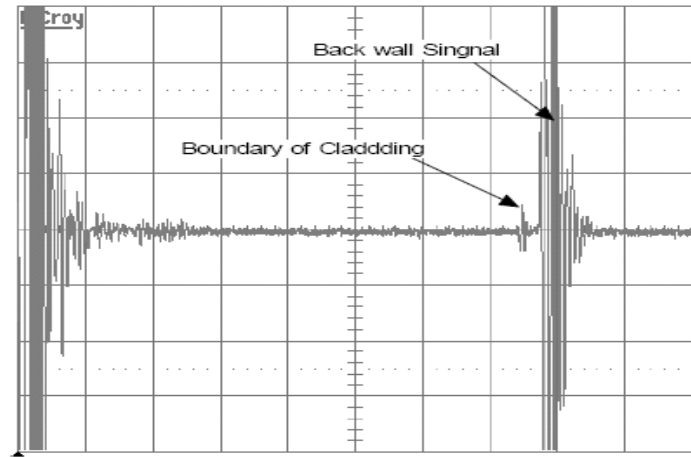
Area #7



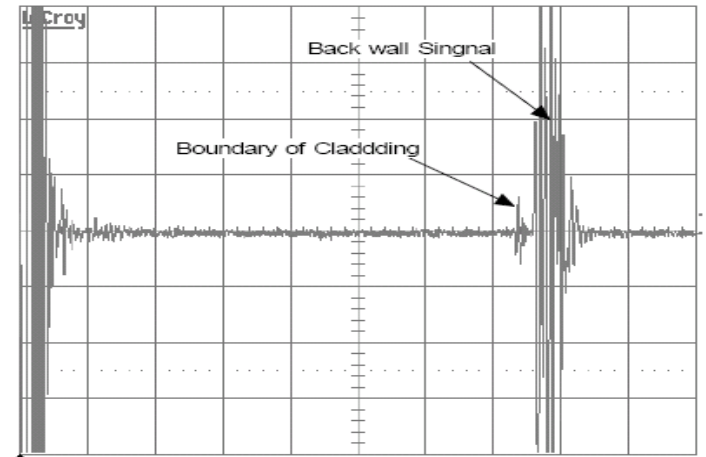
Area #6

# Typical signal form

**Figure 7 : Typical signal form of Backscatter (Area 5 & 6)**

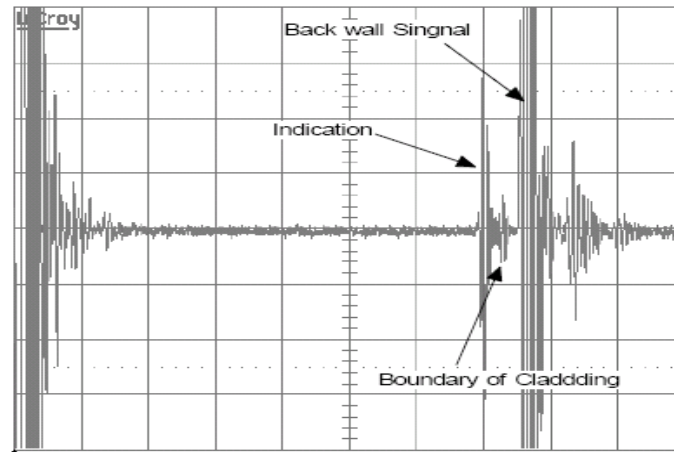


Backscatter Technique

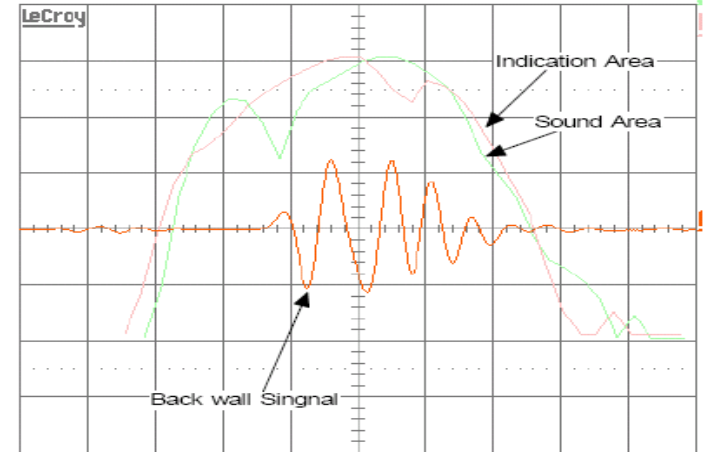


Backscatter Technique

**Figure 8 : Typical signal form of Backscatter / Spectrum Analysis (Area 7)**



Backscatter Technique



Spectrum Analysis

# AUBT Results

The areas inspected were focused on high stress and suspected areas; such as T-cross section on the shell and head. Drawings are included in this report that shows the location of these areas. The AUBT procedure starts by looking for backscatter indications in the base metal by 0 degree. If indications are detected then analysis is performed to determine whether these indications were the result of HTHA.

In according to the results of Backscattering Technique and Spectrum Analysis 0 degree, **no evidence of advanced or micro fissuring due to (HTHA) was detected in the scanned areas of Reactor.** Indications such as inclusions were detected in the base metal on the area #7 of 52V-2 Reactor. These Indications of base metal were not described in this report. Typical signal forms of Backscatter Technique and Spectrum Analysis are shown in Figures 5, 6, 7 & 8.

Also, no crack-like indication was detected on the HAZ using ABSA on the longitudinal weld seam, circumferential weld seam and one nozzle weld. One long slag inclusion was detected on the nozzle weld. The detail of this indication is shown in Figure 2.



# Review

- ▶ **Scoop sampling and testing revealed no evidence of decarburization, methane voids, or microfissuring in any of the samples**
- ▶ **Field Metallography (FMR) – No evidence of HTHA**
- ▶ **Angle Beam Spectrum Analysis (ABSA) – No cracks found**
- ▶ **Advanced Ultrasonic Backscatter Technique (AUBT) – No advanced or microfissuring due to HTHA**

## **Appendix 6**

# **How to use the EFC WP15 Refinery Cases Web page**

## Failure cases atlas

**CORROSION IN REFINERY INDUSTRY FAILURE ATLAS**

CASE HISTORY # 2 Date


Process: Visbreaking  
Equipment: Furnace tube

DATE OF INCIDENT AND/OR INFORMATION: 1905 / refinery inspection team

NATURE OF THE INCIDENT:  
Rupture after 23 years at the entry of the fluid in the last box of the furnace

CONSEQUENCES:  
Shutdown of the unit

MATERIAL COMPOSITION and REFERENCES  
ASTM A335 P9 steel (9% Cr-0.5%Ni)

PICTURES AND SCHEMES:  


EFFECT:  
S102 35502

MEDIA AND OPERATIVE CONDITIONS:  
450 - 470°C 47 Bars  
Catalytic process 100-120 TBP 0.2 \* TAN + 2.5 mg KOH/g  
Fuel gas (outside)

TIME TO DETERIORATION : 23 years

**CORROSION IN REFINERY INDUSTRY FAILURE ATLAS**

CASE HISTORY # 2 ANSWER

TYPE OF CORROSION : Naphthenic acid corrosion  
API 571 CLASSIFICATION: 5.1.1.7

CAUSES:  
Naphthenic acid corrosion due to high molecular acids (T = 420°C) of ASTM A335 P9 steel: 1

REMEDY:  
Replacement by ASTM 335 P9 steel (9% Cr-0.5%Ni)

PUBLICATION - TECHNICAL REPORT:

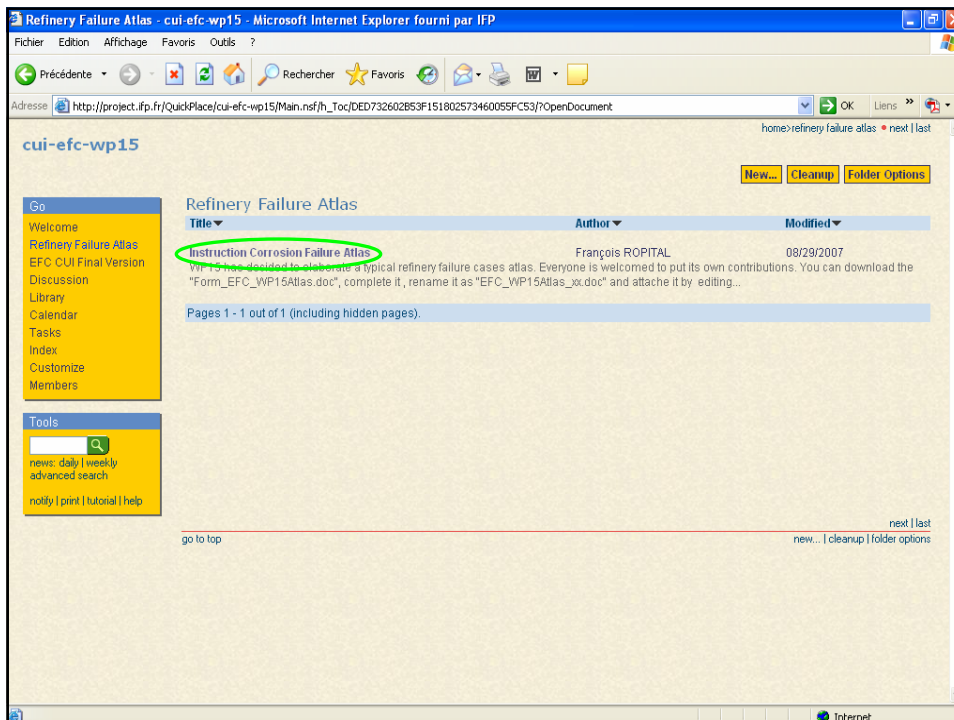
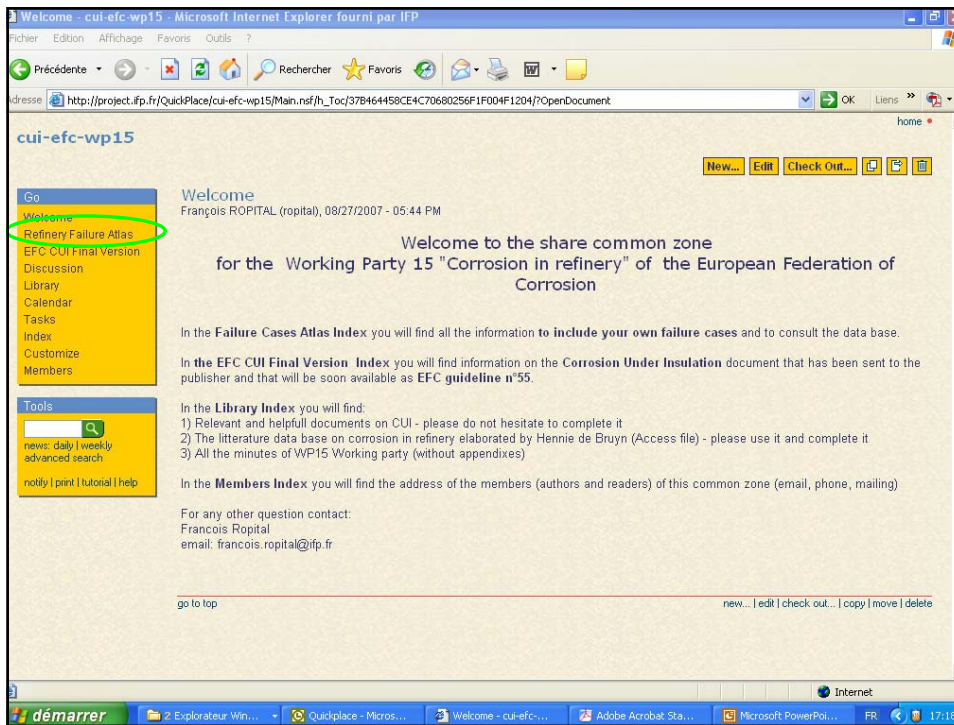
BIBLIOGRAPHIC REFERENCES:

<http://project.ifp.fr/cui-efc-wp15>

Guide line : how to use the failure case web page available

## Failure cases atlas

N° File	Writer	Date	Process	Equipment	Causes	API 571 Classification	Type of material
1	J. Hucinska	2006	Hydrocracking	Reactor	Sulfidation	5.1.1.5	347 SS
2	F. Ropital	29/06/1905	Visbreaking	Furnace	Naphthenic acid corrosion	5.1.1.7	5% Cr steel
3	A. Visgard Nielsen	13/09/2007	Hydrodesulfurizer	Heater	Creep	4.2.8	304 SS
4	F. Ropital	20/12/2007	Continuous Catalytic Reforming	Furnace	Metal dusting	4.4.5	2.25%Cr steel



Instruction Corrosion Failure Atlas - cui-efc-wp15 - Microsoft Internet Explorer fourni par IFF

Fichier Edition Affichage Favoris Outils ?

Précédente Recherche Favoris

Adresse [http://project.iff.fr/QuickPlace/cui-efc-wp15/Main.nsf/h\\_1D69D1CEE4B6647C80257346056232A/4CF98997A29F11E8025734400544FA3/?OpenDocument&F...](http://project.iff.fr/QuickPlace/cui-efc-wp15/Main.nsf/h_1D69D1CEE4B6647C80257346056232A/4CF98997A29F11E8025734400544FA3/?OpenDocument&F...) OK Liens »

home>refinery failure atlas » folder

### cui-efc-wp15

[New...](#) [New Revision](#) [Edit](#) [Check Out...](#) [Print](#) [Delete](#)

**Instruction Corrosion Failure Atlas**  
François ROPITAL (ropital), 08/29/2007 - 04:43 PM

WP15 has decided to elaborate a typical refinery failure cases atlas.

Everyone is welcomed to put its own contributions. You can download the "Form\_EFC\_WP15Atlas.doc", complete it, rename it as "EFC\_WP15Atlas\_xx.doc" and attache it by editing your own page.

To numerate your file, please take the following available number from the file : "List\_EFC\_WP15Atlas\_August\_2007.xls"

If you prefer you can email your failure cases files to :

francois.ropital@iff.fr who will incorporate it on this web page

You can download or launch any of the files below by clicking on them:

[Form\\_EFC\\_WP15Atlas.doc](#)  
[List\\_EFC\\_WP15Atlas\\_August\\_2007.xls](#)  
[Classification API 571.pdf](#)  
[EFC\\_WP15Atlas\\_1.pdf](#)  
[EFC\\_WP15Atlas\\_2.pdf](#)

folder  
go to top [new...](#) [new revision](#) [edit](#) [check out...](#) [copy](#) [move](#) [delete](#)

Internet

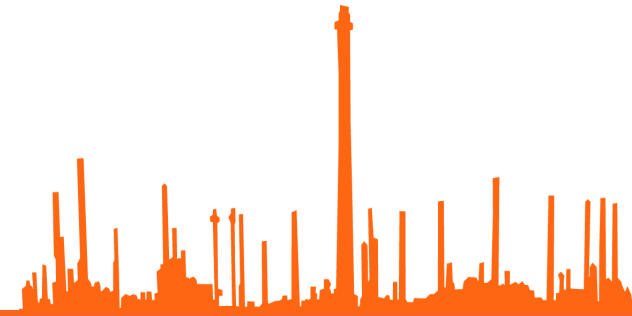
## **Appendix 7**

### **PASCC case in a FCCU**

**Wim Verstijnen (Shell)**



# FCCU Liftpot Failure & Repair

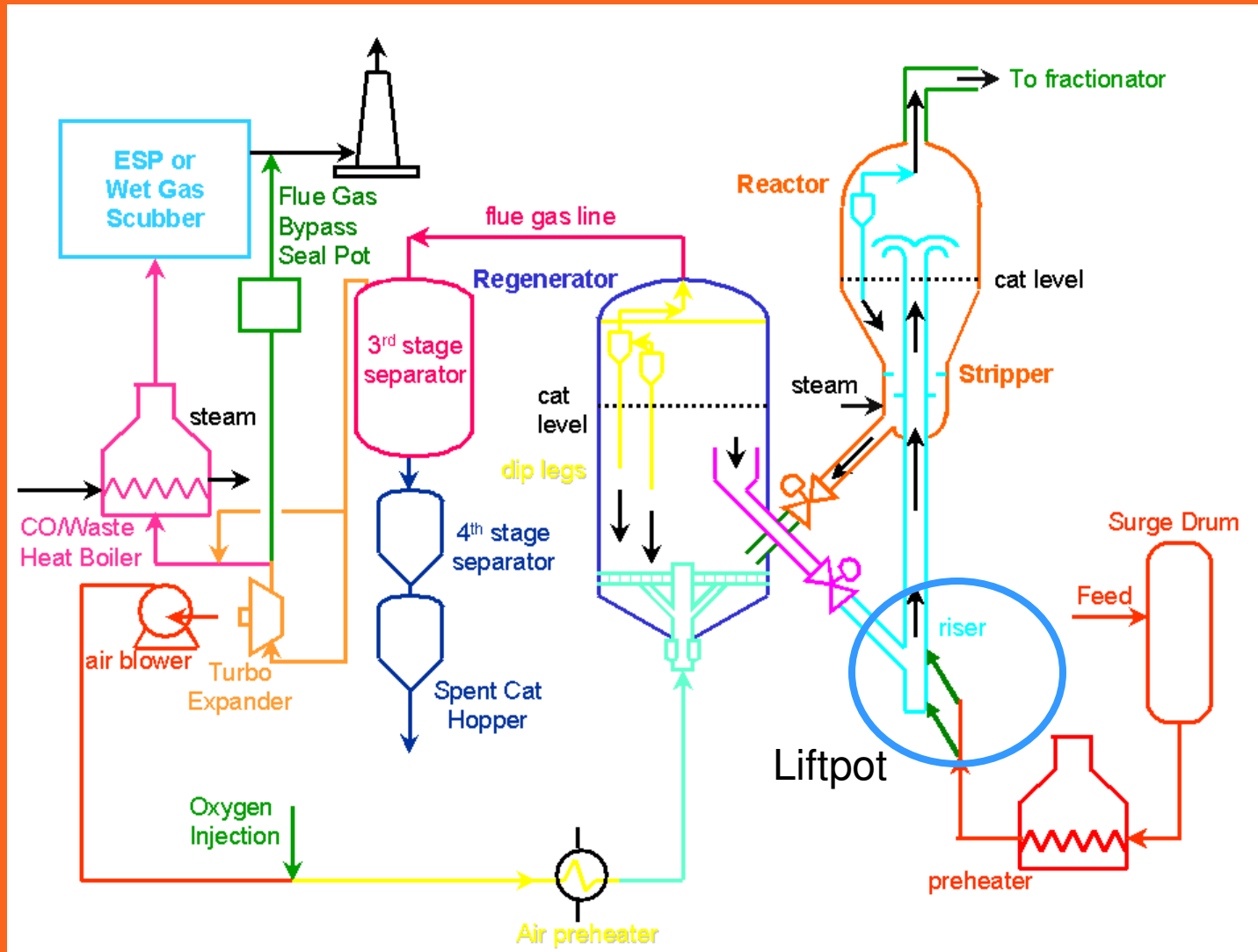


Wim Verstijnen

Head Engineering Support

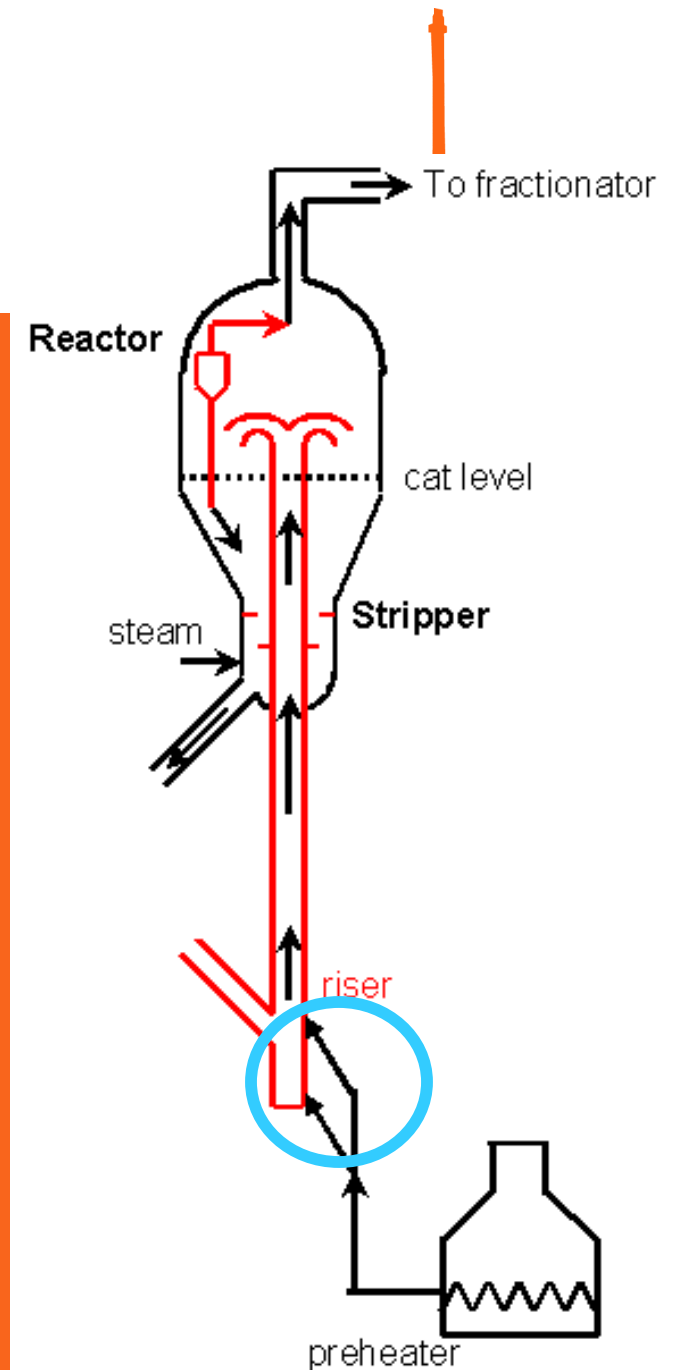
Shell Nederland Raffinaderij  
Pernis - Rotterdam

# The Process: Catalytic Cracker Unit



# Process “Liftpot”

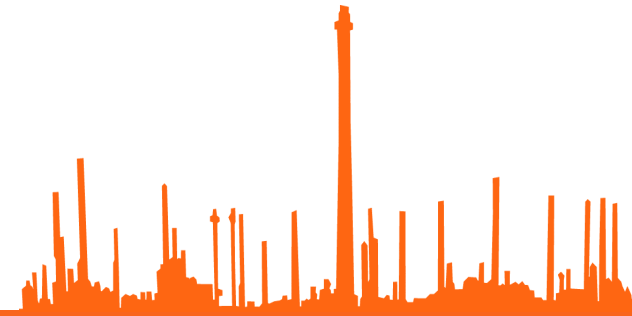
- Feed mixes with Catalyst of 680 degC
- Mix of Feed and Cat “cools” to 520 degC @ 2.5 barg Riser outlet
- Steam ring at bottom of riser liftpot to enhance/assist mixing and transport of catalyst.
- Medium: Oxygen, Air, H<sub>2</sub>S, Steam/water, Salts(?), Hydrocarbons



# Materials

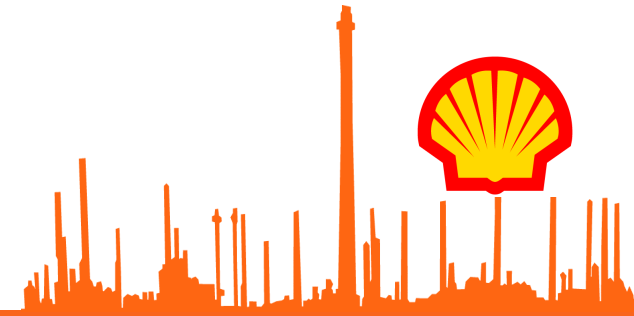
- Reactor liftpot Avesta 253MA (austenitic SS ~ SS 304H)
- Wall thickness head/wall 40mm / 25mm
- Erosion resistant liner of 35 mm.
- Dished end filled with insulating concrete
- Other riser parts SS 304H

# History Liftpot:



- Installed in 1986
- No problems till crack was found

# Crack in Liftpot Body

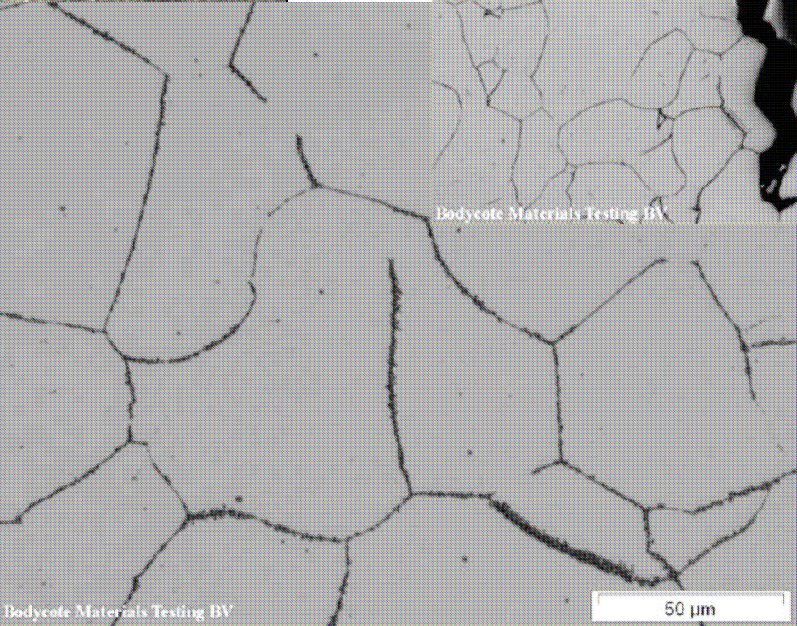
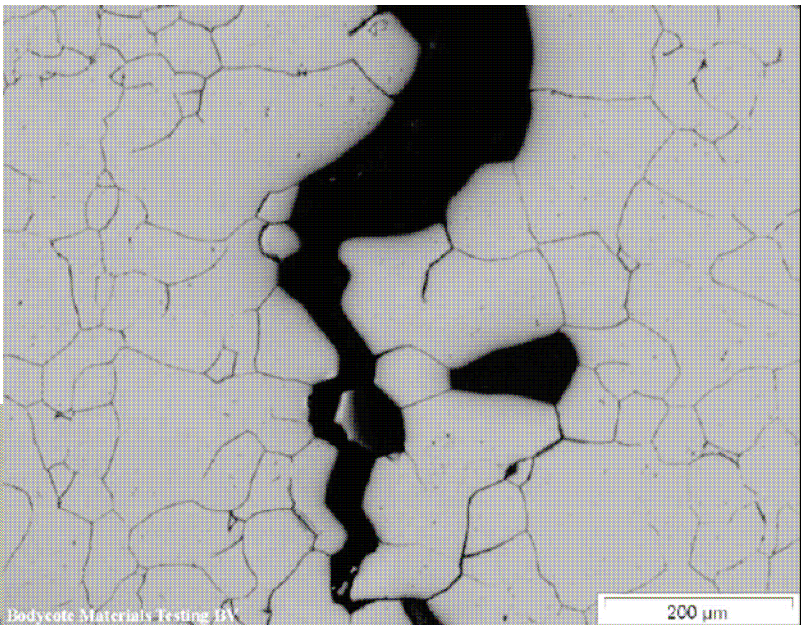
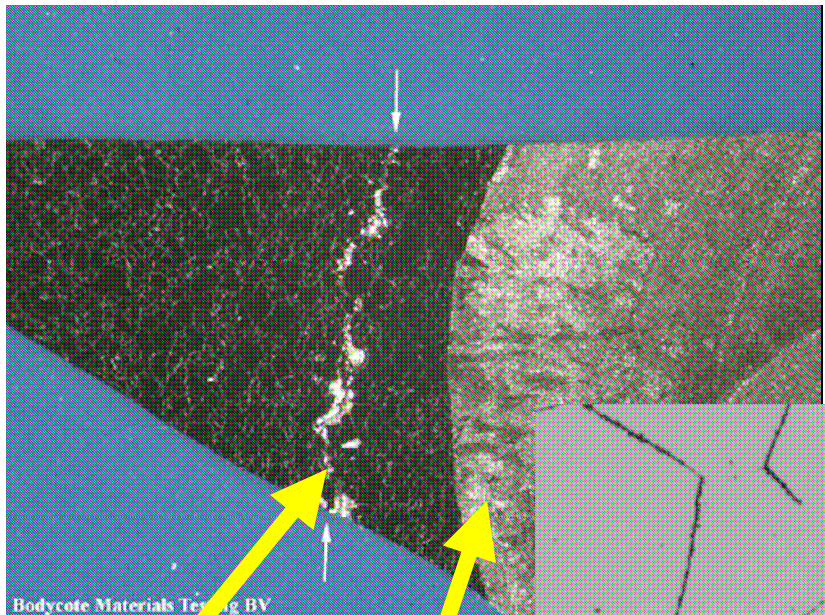
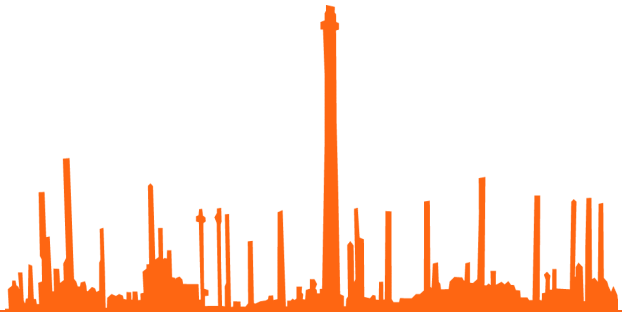


**Weld**





# Metallography of Crack (Boat Sample)



Crack in HAZ

Weld

# Findings:

- Leakage (Inspection, operations)
- Visual/NDT: Crack in Circumferential weld
- Metallurgical investigation:
  - Sensitized HAZ weld,
  - Inter-crystalline attack in HAZ,
- NDT: Other welds and (plate) material free from cracks

## Conclusion:

- Polythionic Stress Corrosion Cracking of sensitized weld area.

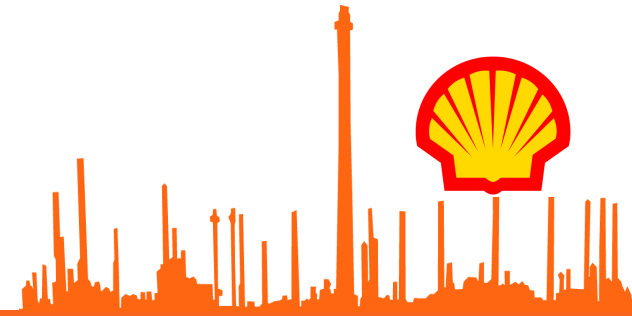


# Solution Short/Long-term

- Short-term actions:
  - Repair with external reinforcing ring.
- Long-term actions:
  - SCC resistant material
  - Different design



Thank you



Questions?

## **Appendix 8**

### **Use of fitness for service assessments**

**Hennie de Bruyn (Borealis Group)**

# Use of Fitness-for-Service Assessments

Hennie de Bruyn  
Chief Engineer Material Technology



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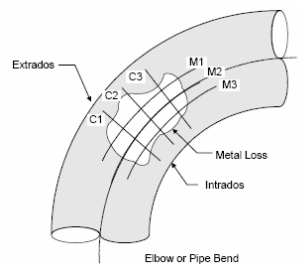
## Introduction

- Inspection results

- Detect damage / degradation
- Deviations from original design
- What now?
  - Do nothing
  - Repair
  - Replace

- Fitness-for-Service (FFS)

- FFS assessment facilitates the decision-making process



- FFS Definition (API 579-1)

- FFS assessments are quantitative engineering evaluations that are performed to demonstrate the structural integrity of an in-service component containing a flaw or damage

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2

15 April 2008

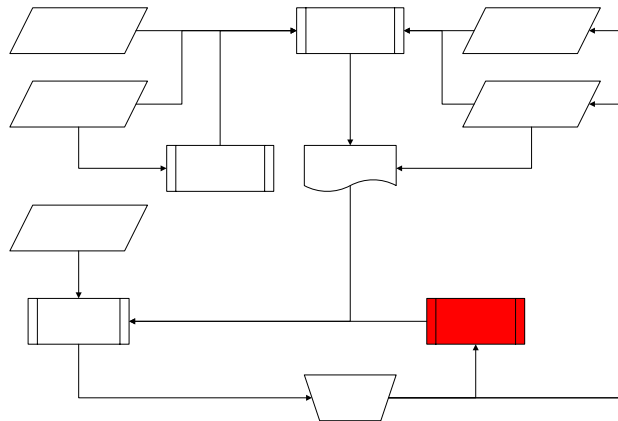
Use of FFS Assessments - Hennie de Bruyn





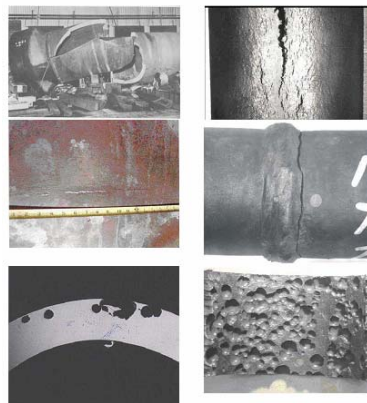
# Plant Integrity Management

## The Role of FFS



# Damage & Degradation

- Brittle fracture
- Metal loss
  - General
  - Localised
  - Pitting
- Hydrogen damage
- Weld misalignment / shell distortions
- Cracking / Plastic collapse
  - Stress corrosion cracking
  - Fatigue
- Creep
- Fire damage / other undesirable events



Design  
build

Pro

Main  
In  
Plan

# Cracking / Plastic Collapse

## FFS methods, procedures & standards

Document	Title	Published by
R6	Assessment of the integrity of structures containing defects	British Energy
BS 7910	Guide to methods for assessing the acceptability of flaws in metallic structures	British Standards Institution
API 579-1 / ASME FFS-1	Fitness For Service	American Petroleum Institute / American Society of Mechanical Engineers
ASME SEC XI	Rules for Inservice Inspection of Nuclear Power Plant Components	American Society of Mechanical Engineers
SINTAP	Structural Integrity Assessment Procedures for European Industry	EU open source
JSME S NA1	Codes for Nuclear Power Generation Facilities - Rules on Fitness-for-Service for Nuclear Power Plants	Japanese Standards Association (JSA)
RSE-M	Rules for in service inspection on nuclear power plant components	AFCEN

# Fatigue

## FFS methods, procedures & standards

Document	Title	Published by
R6	Assessment of the integrity of structures containing defects	British Energy
BS 7910	Guide to methods for assessing the acceptability of flaws in metallic structures	British Standards Institution
API 579-1 / ASME FFS-1	Fitness For Service	American Petroleum Institute / American Society of Mechanical Engineers
ASME SEC XI	Rules for Inservice Inspection of Nuclear Power Plant Components	American Society of Mechanical Engineers
JSME S NA1	Codes for Nuclear Power Generation Facilities - Rules on Fitness-for-Service for Nuclear Power Plants	Japanese Standards Association (JSA)

# Creep

## FFS methods, procedures & standards

Document	Title	Published by
R5	Assessment Procedure for the High Temperature Response of Structures	British Energy
BS 7910	Guide to methods for assessing the acceptability of flaws in metallic structures	British Standards Institution
A16	Design and Construction Rules for Mechanical Components of FBR Nuclear Islands	AFCEN
API 579-1 / ASME FFS-1	Fitness For Service: includes now the Materials Properties Council (MPC) Omega method	American Petroleum Institute / American Society of Mechanical Engineers

# Thinning / Corrosion / Metal Loss

## FFS methods, procedures & standards

Document	Title	Published by
ASME B31G	Manual for Determining the Remaining Strength of Corroded Pipelines	American Society of Mechanical Engineers
DNV RP F-101	Corroded Pipelines	Det Norske Veritas
PDAM	The Pipeline Defect Assessment Manual	Penspen Ltd (UK)
API 579-1 / ASME FFS-1	Fitness For Service	American Petroleum Institute / American Society of Mechanical Engineers

## Open questions to EFC WP15

- Is your company using FFS assessments when damage/degradation is detected by inspection?
- What assessment methods/procedures/standards are used?
- Who is responsible to perform assessments in your company?
- Does your company allow inspectors to perform basic assessments?
- Are FFS assessments accepted by your regulating authorities?

**Hennie de Bruyn**  
**15 April 2008**

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## **Appendix 9**

**How, when and where to monitor**

**Dimphy Wilms (Applus<sup>+</sup> RTD)**

# How, when and where to monitor ?

Dimphy Wilms

EFC Working Party 15

15 april 2008



NDT & Inspection



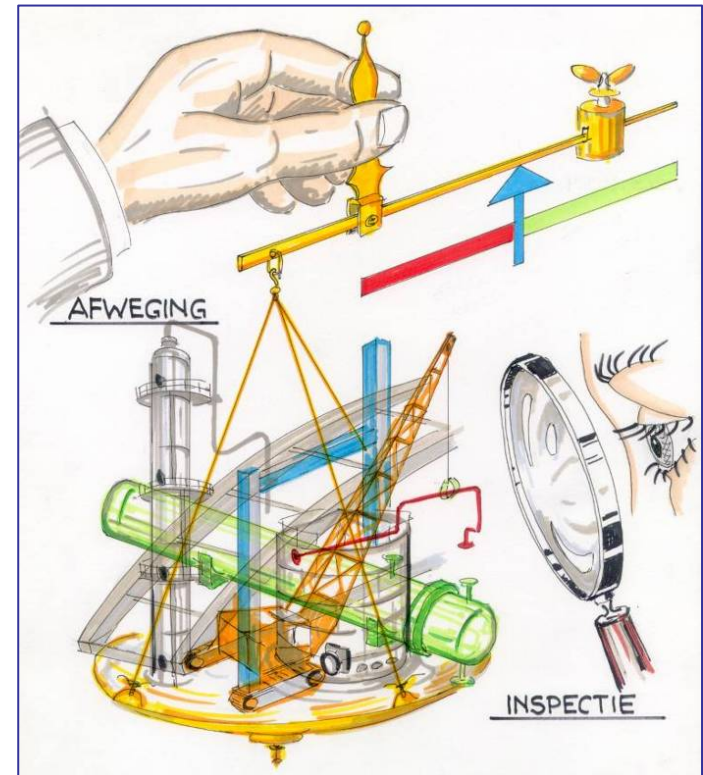


# What are you looking for ?

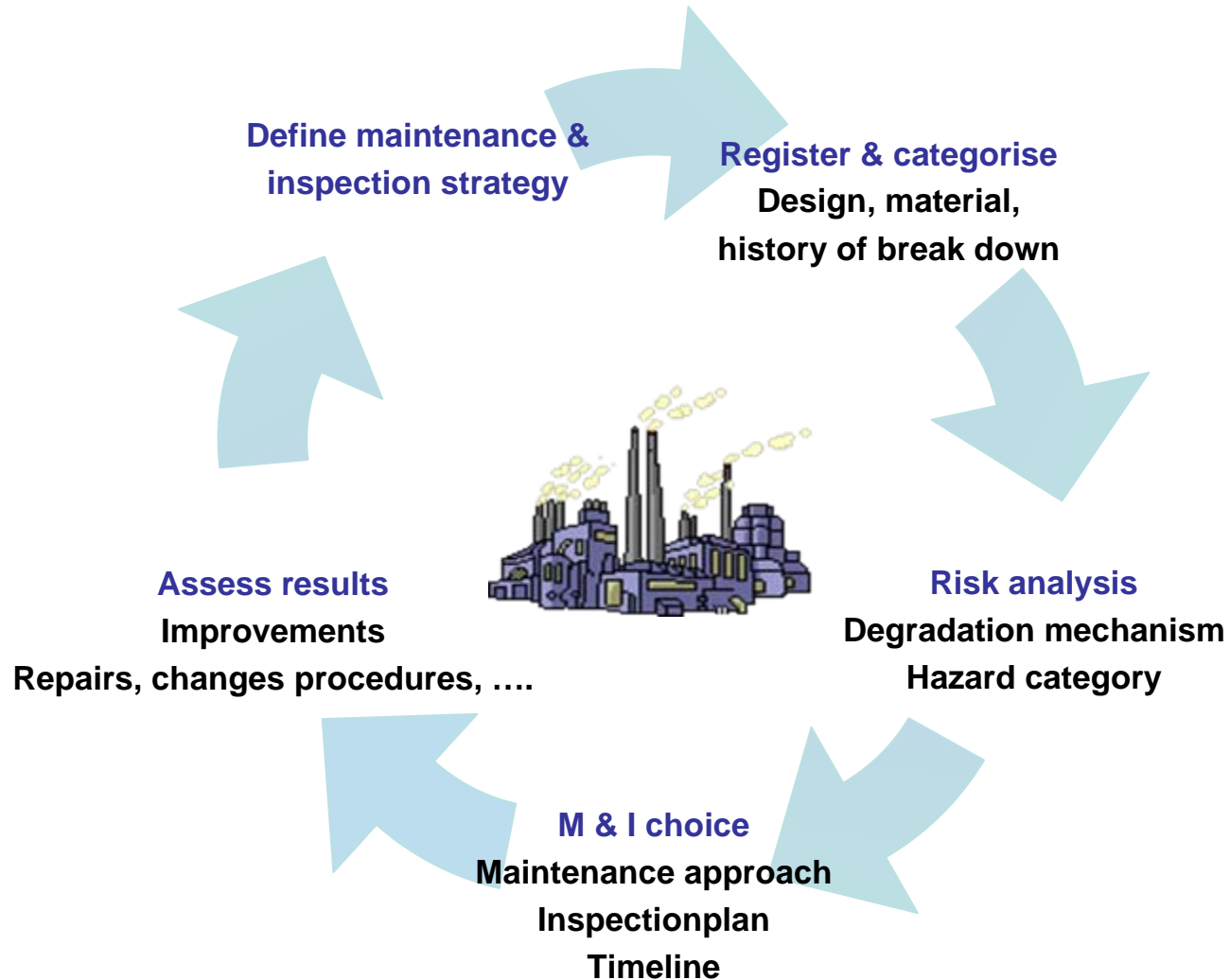


# Purpose of inspection

- ⊕ Maintain the **integrity** of an asset
- ⊕ Increase / maintain **reliability**
- ⊕ Maintain a **safe workplace**
- ⊕ Ensure **Fitness for Service**
- ⊕ Do it at the **lowest possible cost!**



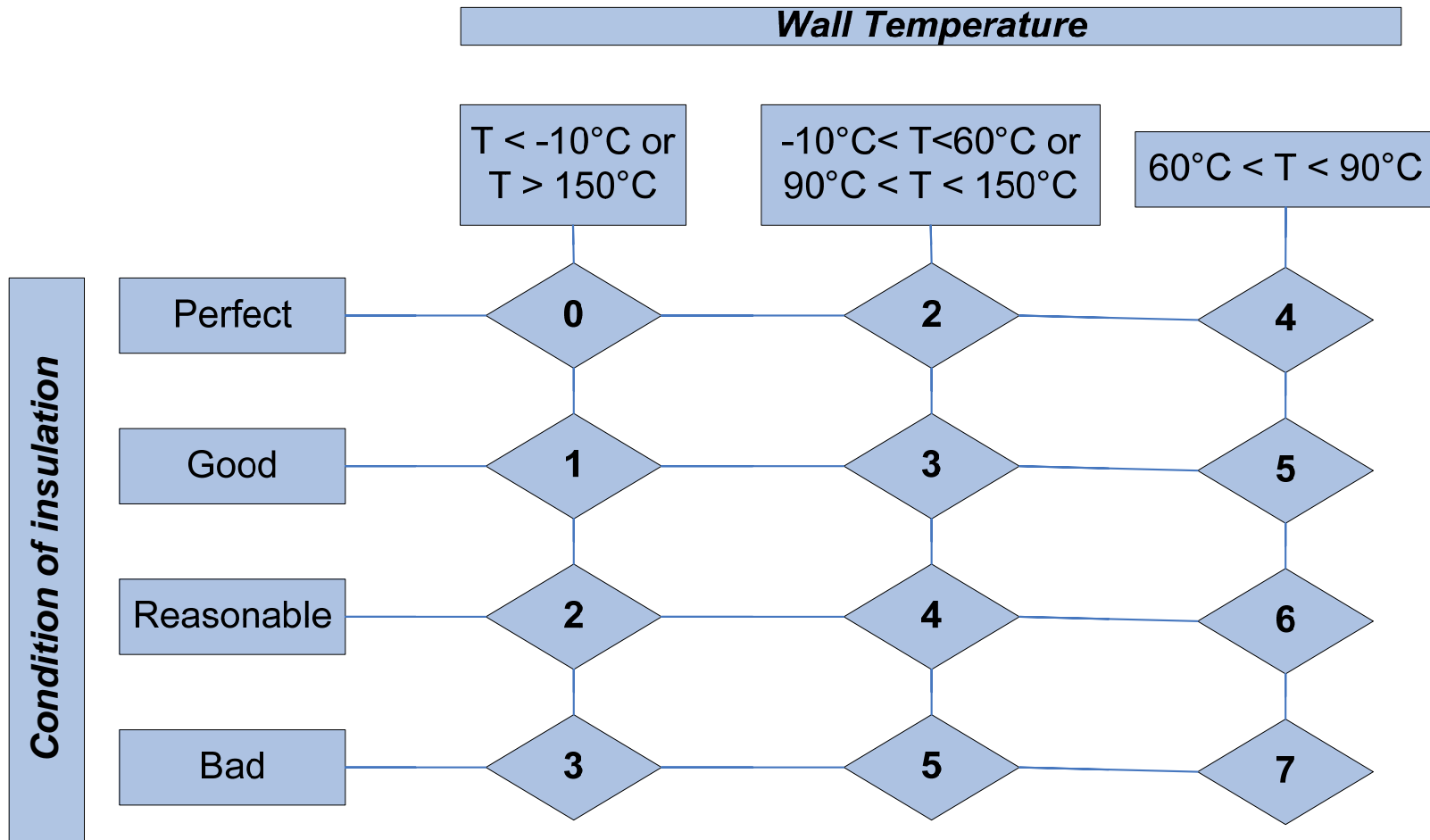
# Condensed version of Seveso II



- ⊕ Specifically for objects and degradation mechanism
- ⊕ Minimum impact for operation
- ⊕ Acceptable according to rules and regulations
- ⊕ Reliability focused
- ⊕ Time optimised



# How to design a CUI programme ?



- ⊕ Combination of methods
- ⊕ In line with rules and regulations
- ⊕ Sampling to be decided
- ⊕ Inspection effectiveness to be agreed upon
- ⊕ Non-Intrusive ?
- ⊕ Screening or Measuring
  
- ⊕ *Update database in order to improve*



# What technique is the 'right' one ?

- ⊕ A vessel was inspected by using four different NDT techniques
  - Automated C-scan
  - INCOTEST
  - UT raster
  - Slofec

# 4 technologieën



## Automated C-scan

- ⊕ 1 Day
- ⊕ Insulation removal & surface prep
- ⊕ Highest resolution
- ⊕ 100% coverage
- ⊕ Accurate WT reading

## INCOTEST

- ⊕ 4 hours
- ⊕ No insulation removal
- ⊕ Lowest resolution
- ⊕ 100% coverage
- ⊕ No absolute WT reading

## Manual UT Raster

- ⊕ 2 Days
- ⊕ Insulation removal & surface prep
- ⊕ Medium resolution
- ⊕ Bad coverage
- ⊕ Accurate WT reading
- ⊕ Standard UT set

## Slofec

- ⊕ 2 hours
- ⊕ Insulation removal only
- ⊕ High resolution
- ⊕ 100% coverage
- ⊕ No absolute WT reading

# Never forget the human factor





Never forget the human factor



- ⊕ Using the appropriate method(s)
- ⊕ At a time that is suitable
- ⊕ At locations that need to be watched
- ⊕ No system replaces common sense, however it can help you making the right decision