

# Assessment of the Pt nanoparticle distribution on oxidized stainless steel surfaces by electrochemical techniques

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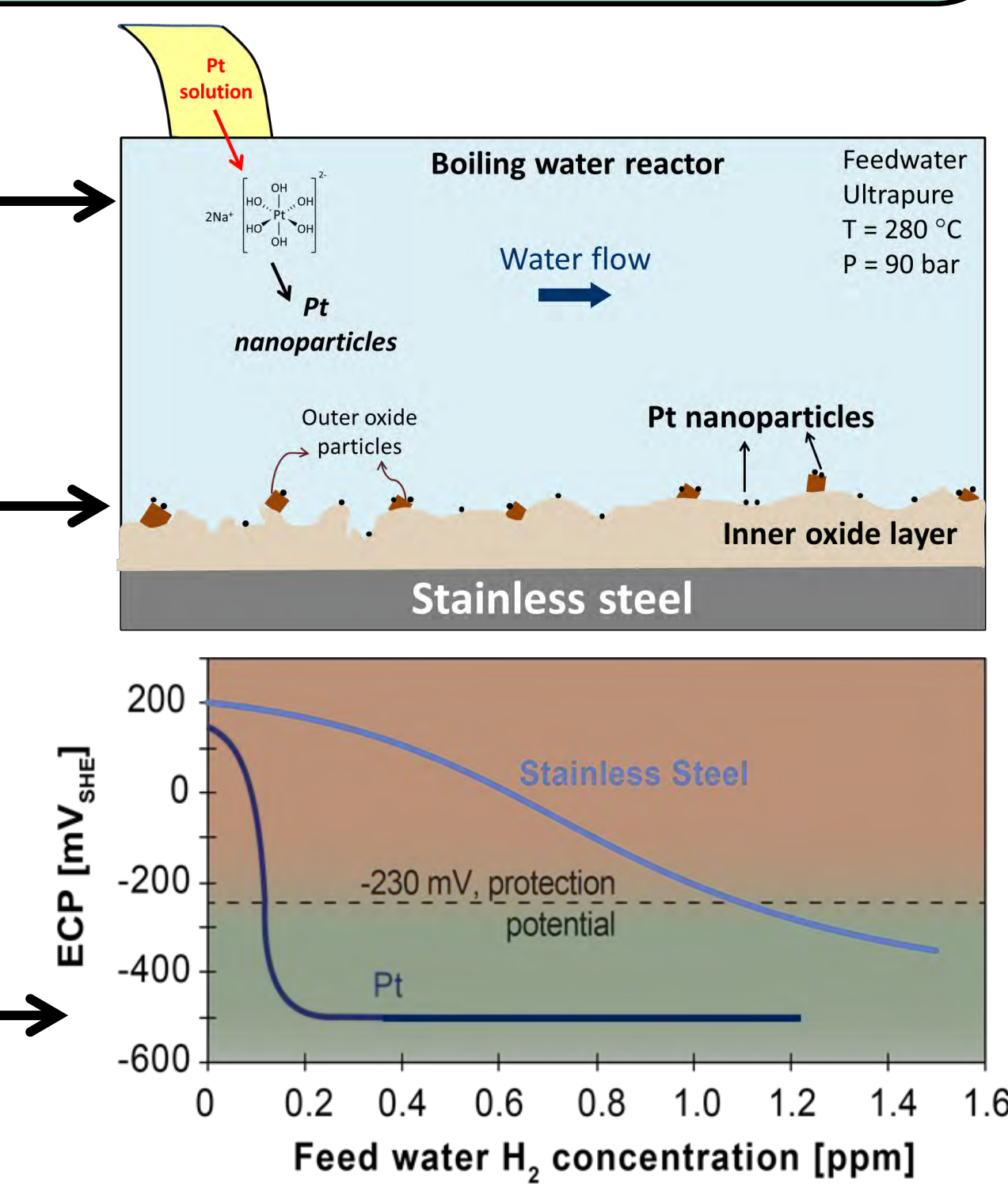
## 1 Background

Online NobleChem (OLNC) technology in boiling water reactors

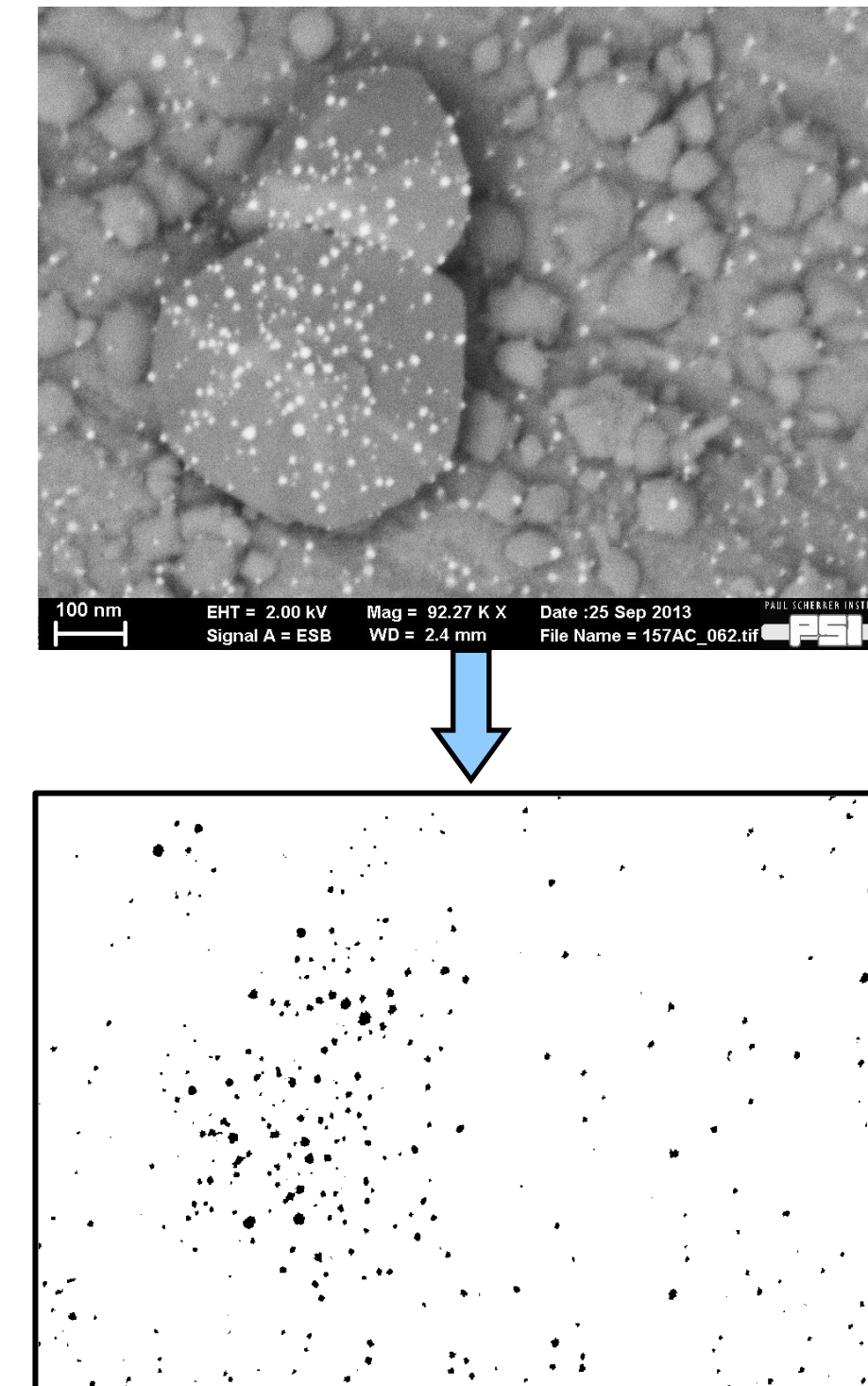
Addition of noble metal (Pt)



Pt decreases the susceptibility to stress corrosion cracking (SCC)



## Heterogeneous distribution of Pt particles



## 2 Objectives

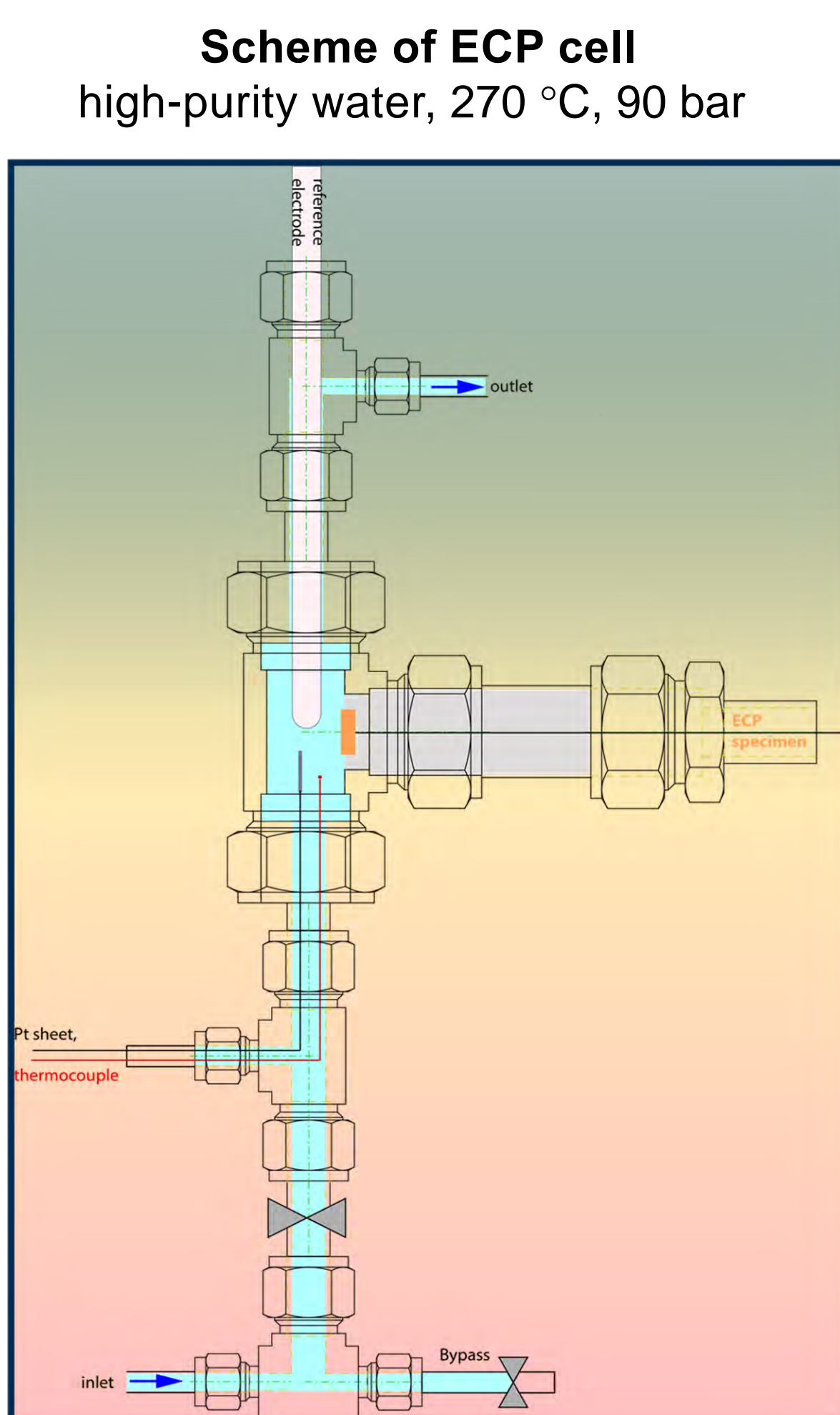
To quantify non-destructively:

- Pt surface loadings →  $\frac{\text{Pt mass deposited}}{\text{total area}}$
- Pt spatial distributions (interparticle distances)
- Establish relations between:
  - (a) Pt surface loadings vs. ECP
  - (b) Pt interparticle distances vs. ECP

Routes → **Electrochemical Corrosion Potential (ECP) & Electrochemical Impedance Spectroscopy (EIS)**

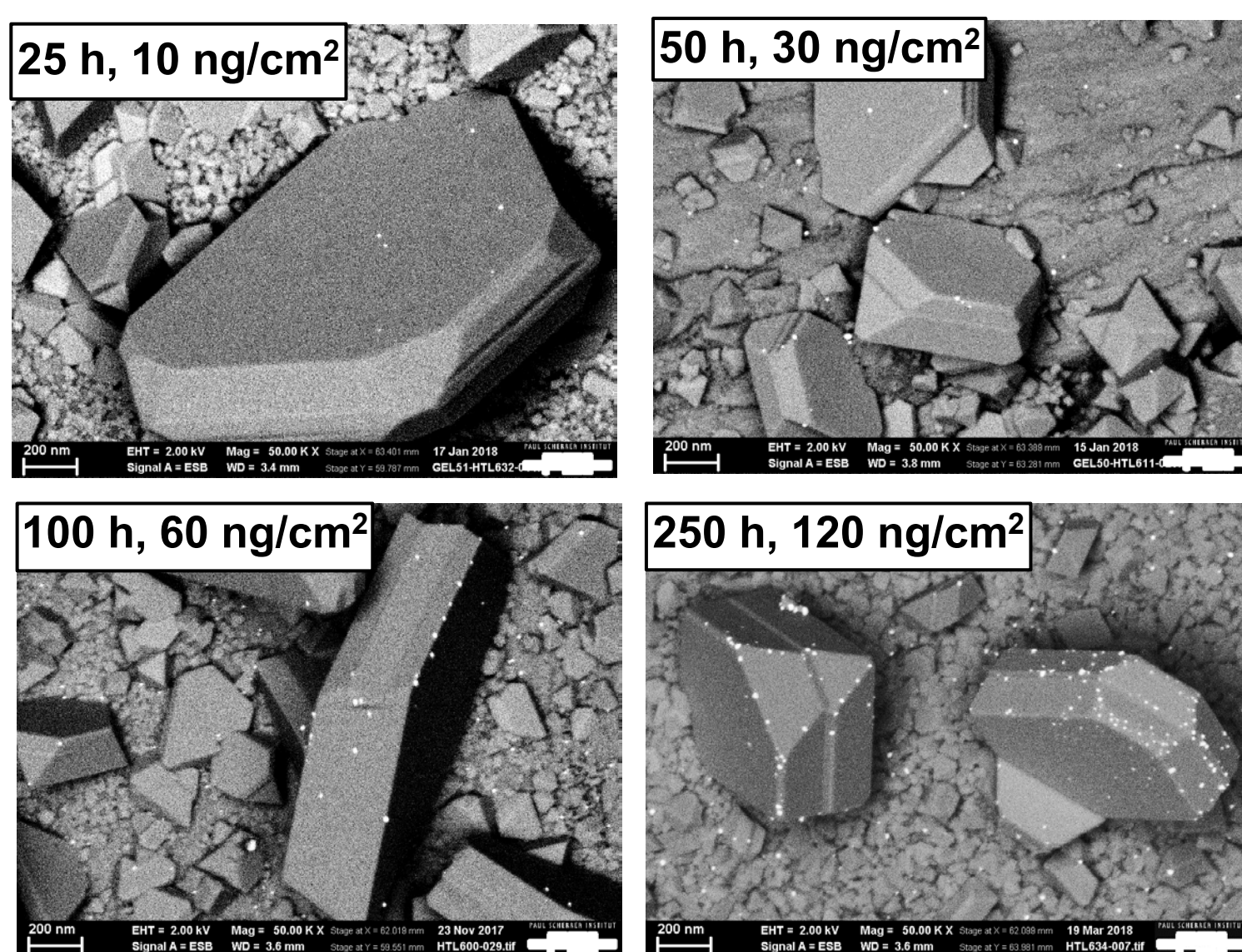
## 3.1 Electrochemical Corrosion Potential (ECP) ↔ Pt Spatial Distribution

### Experimental setup and details



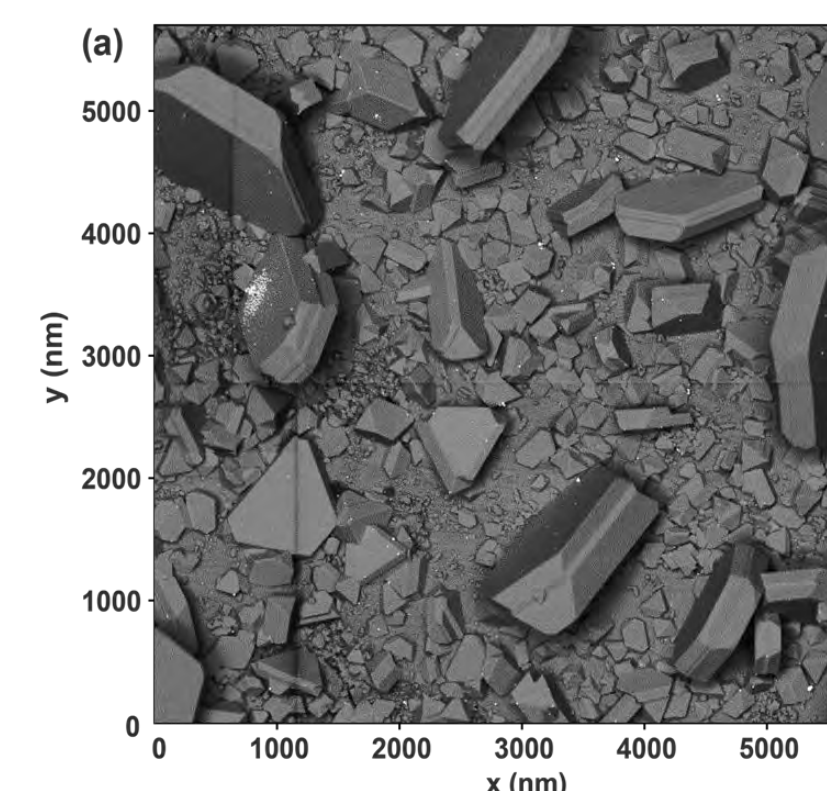
### Specimens with varying Pt loadings and interparticle distances

#### Pt injection duration, nominal surface loading

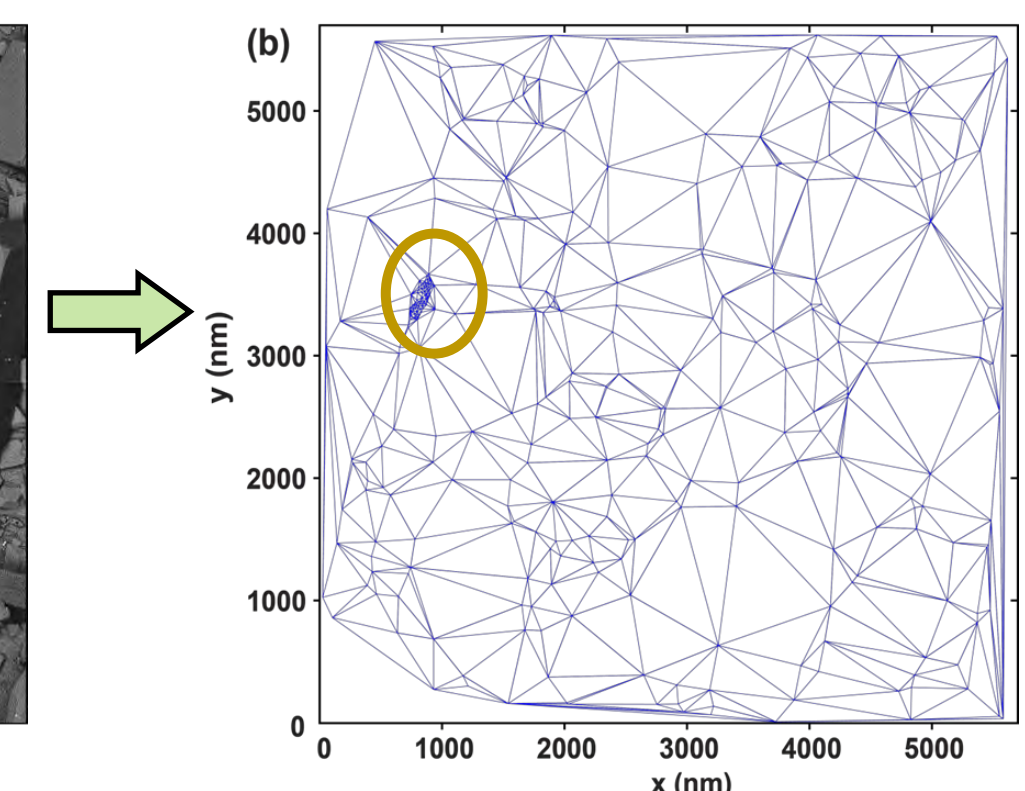


Duration of Pt injection (h)	Pt amount (µg)	Mean particle size (nm)	Avg. Pt surface loadings (ng/cm²)	Mean interparticle distance ± 1SD (nm)
25	57	8.5 ± 3.8	9 ± 5	563 ± 351
50	113	14.4 ± 8.7	36 ± 30	344 ± 272
100	207	10.1 ± 4.8	44 ± 30	290 ± 232
250	553	10.9 ± 5.7	123 ± 36	201 ± 162

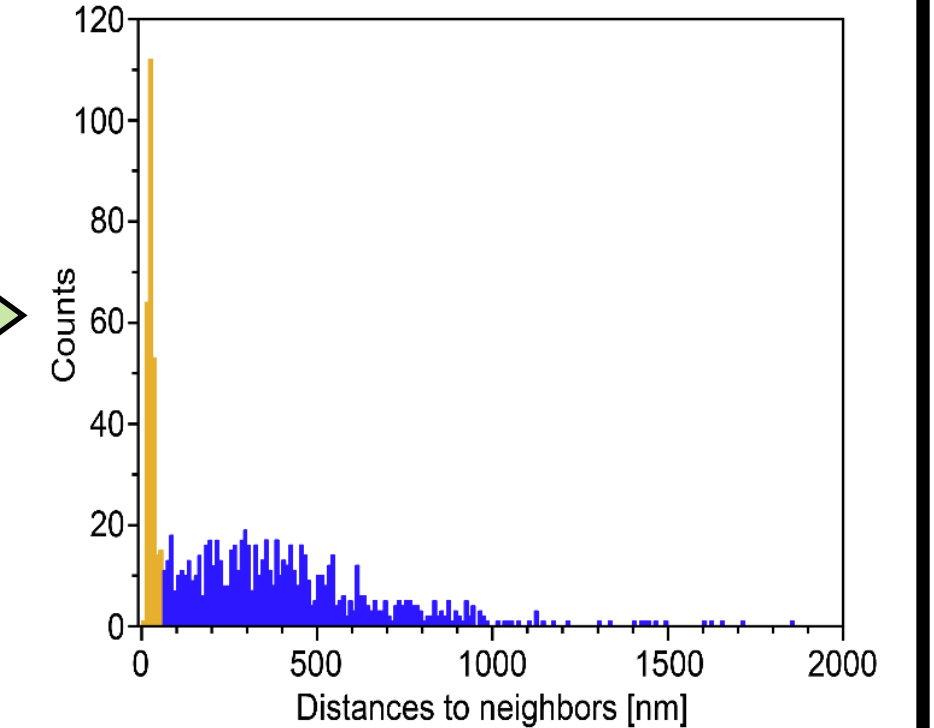
#### Stitched SEM pictures



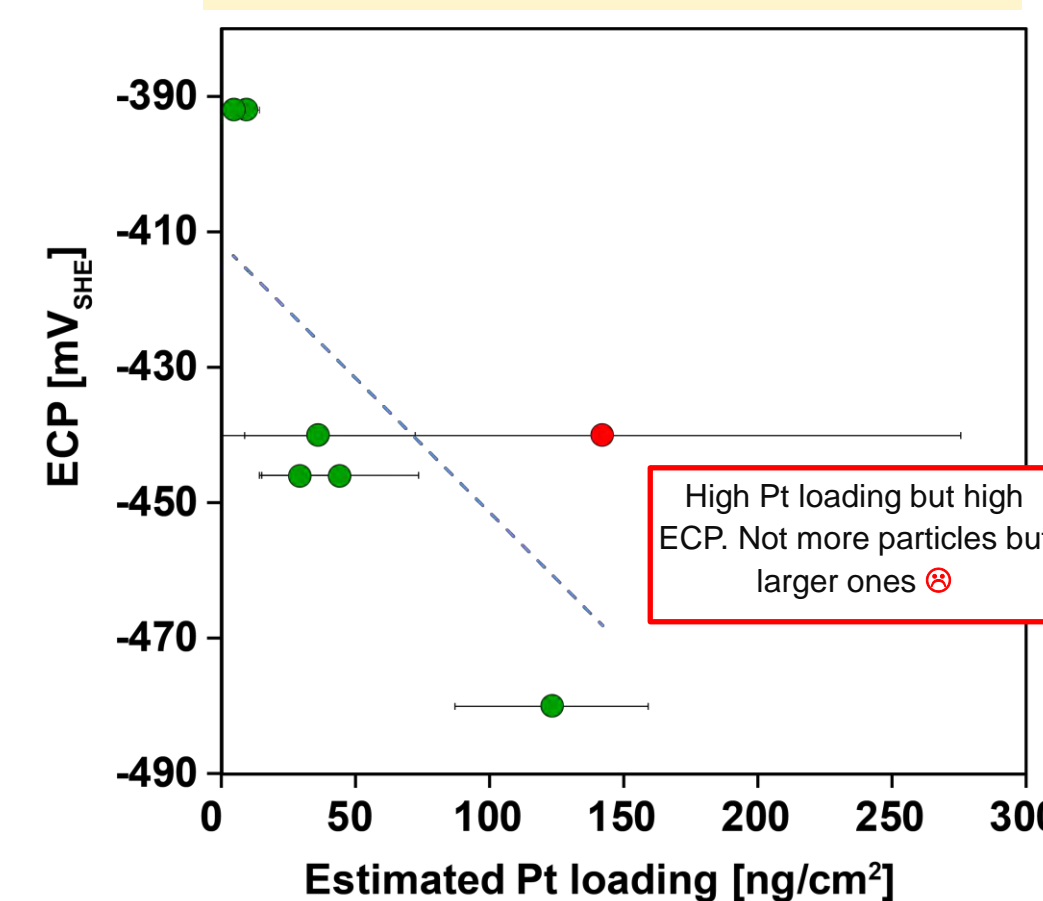
#### Delatunay triangulation



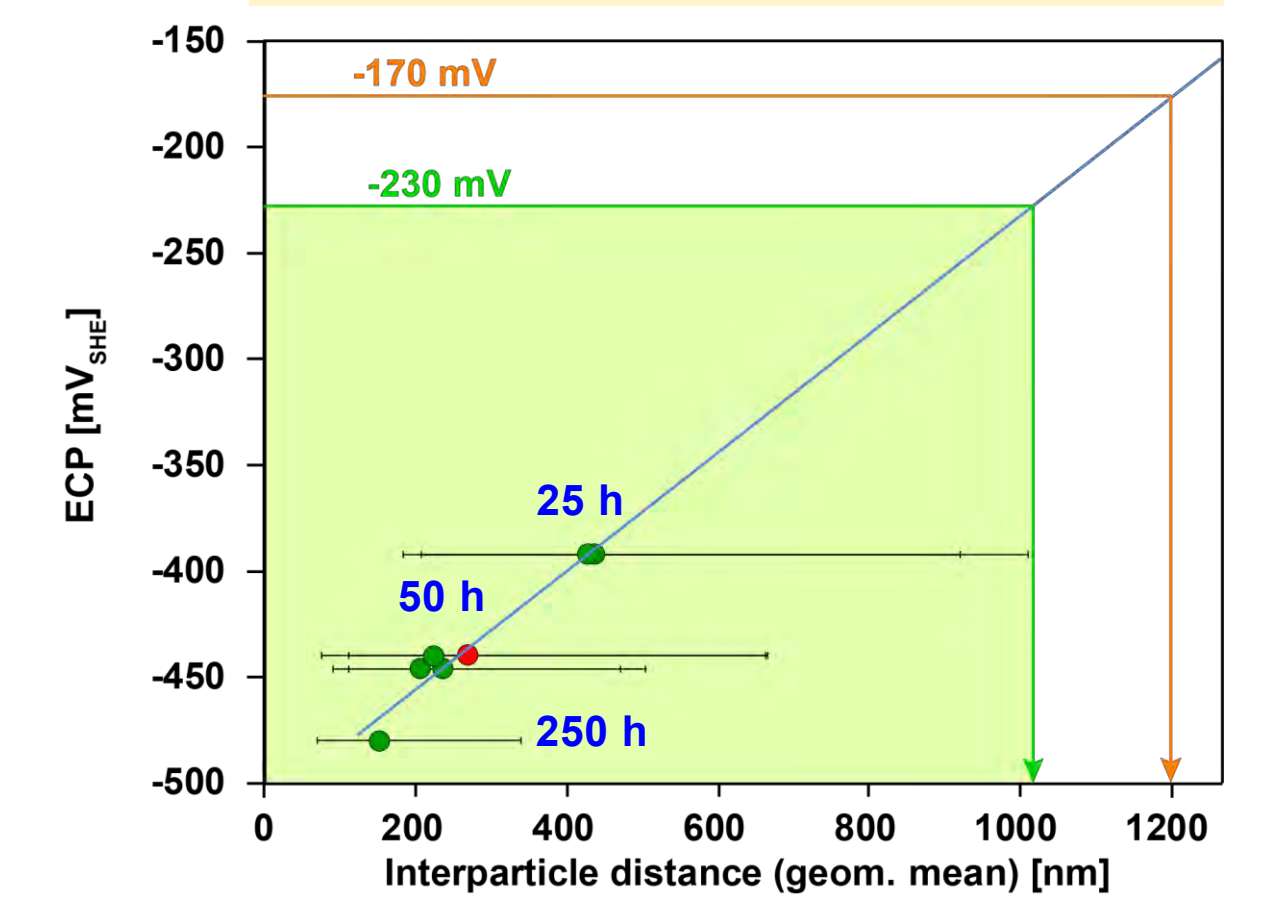
#### Interparticle distances



#### Pt loading vs. ECP



#### Interparticle distance vs. ECP



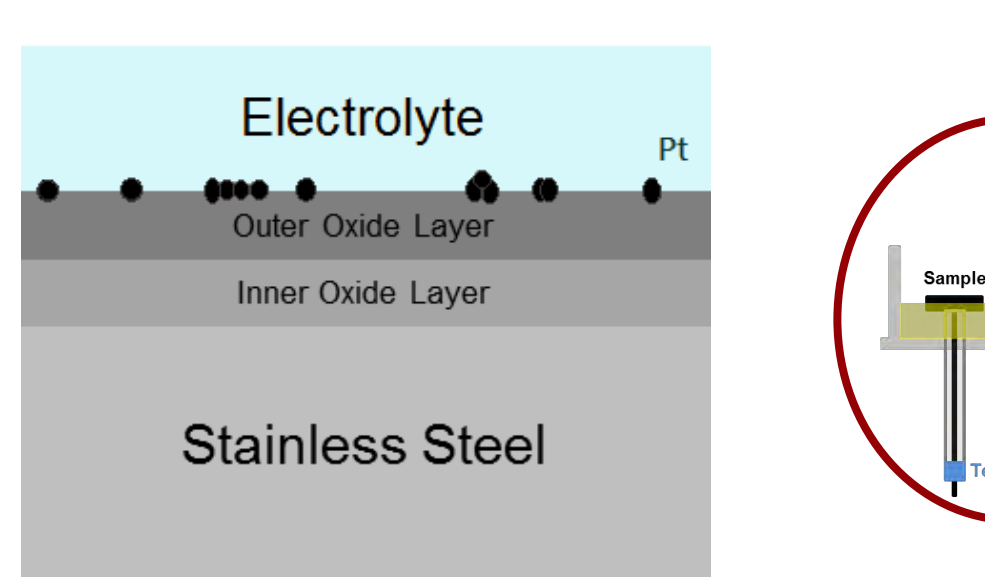
✓ Interparticle distances are better predictors of ECP as compared to loadings

P. V. Grundler et al., Assessment of the SCC Mitigation Capabilities of the Noble Metal Chemical Application Technology in a Simulated BWR Environment, 19<sup>th</sup> International conference on Environmental Degradation of Materials in Nuclear Power Systems, Aug 18-22, 2019, Boston, USA

## 3.2 Electrochemical Impedance Spectroscopy (EIS) ↔ Pt Surface Loading

### Experimental setup and details

25 °C, 0.5M Na<sub>2</sub>SO<sub>4</sub>, atm. pressure, Ar-purged



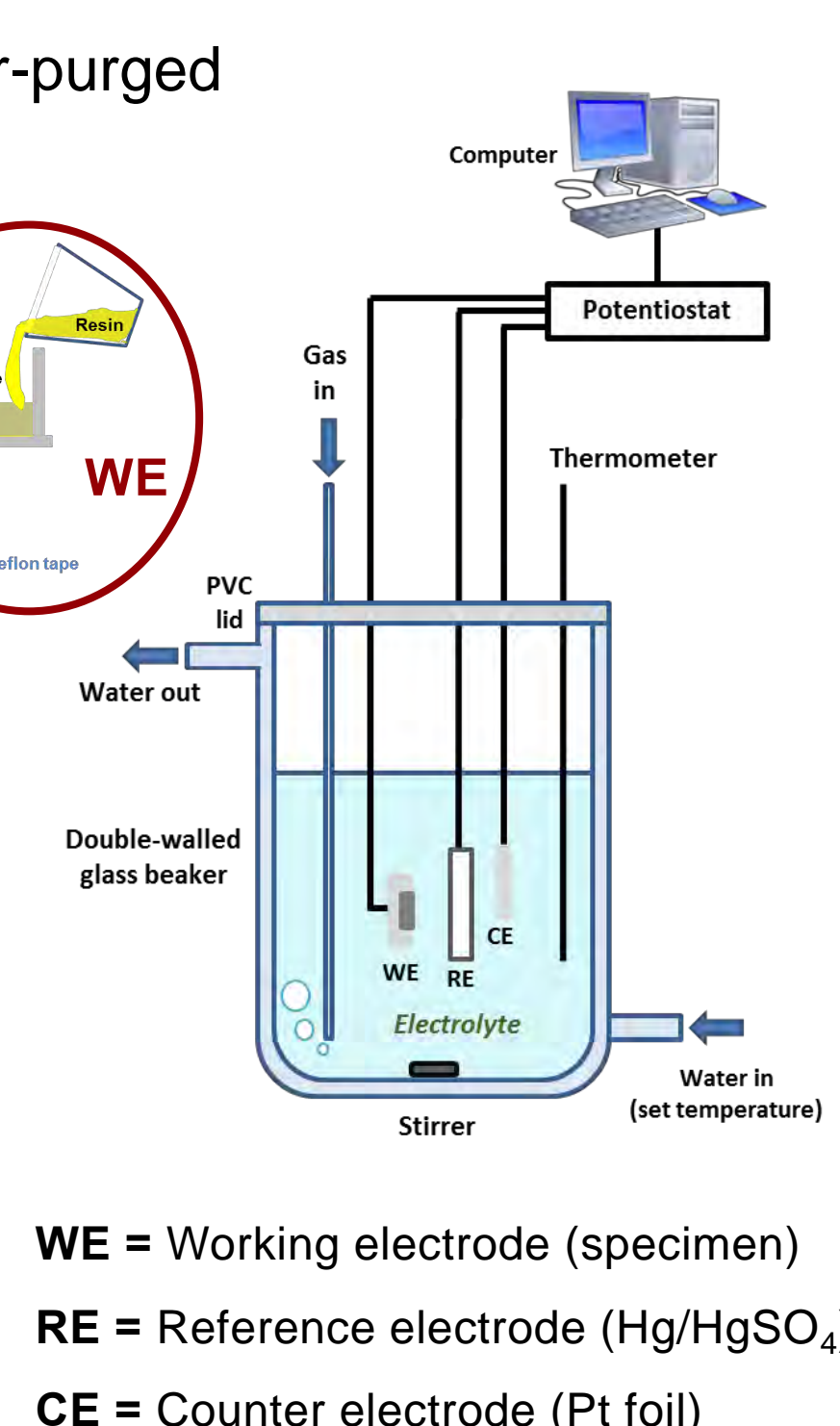
$$\text{Impedance } Z(f, t) = \frac{V_o \sin(2\pi ft)}{I_o \sin(2\pi ft - \theta)}$$

$$Z = Z_{\text{real}} + iZ_{\text{imag}}$$

$$|Z| = \sqrt{Z_{\text{real}}^2 + Z_{\text{imag}}^2}$$

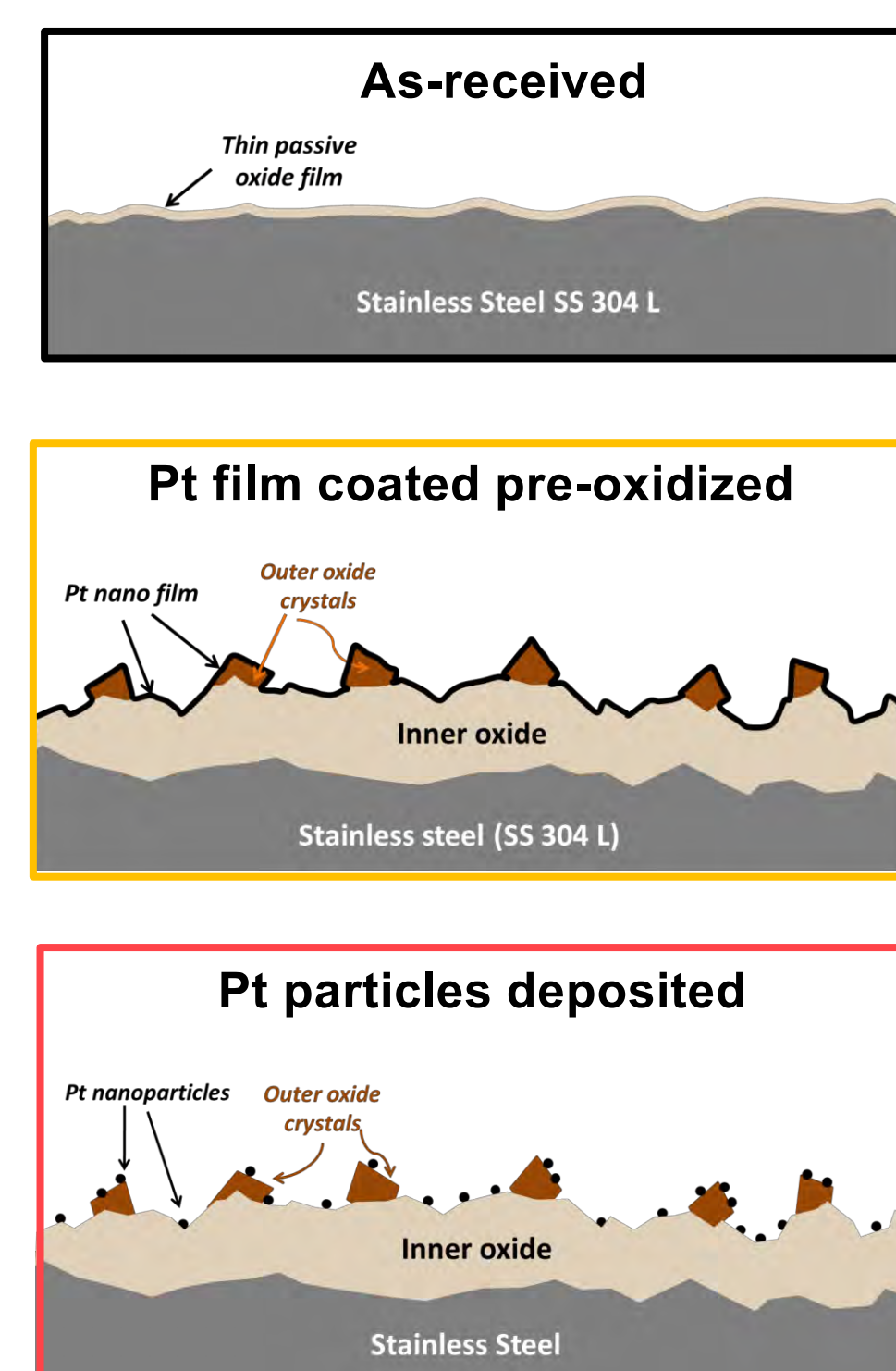
Frequency (f) = 20,000-0.005 Hz

V<sub>o</sub> = 10 mV, 10 points/decade

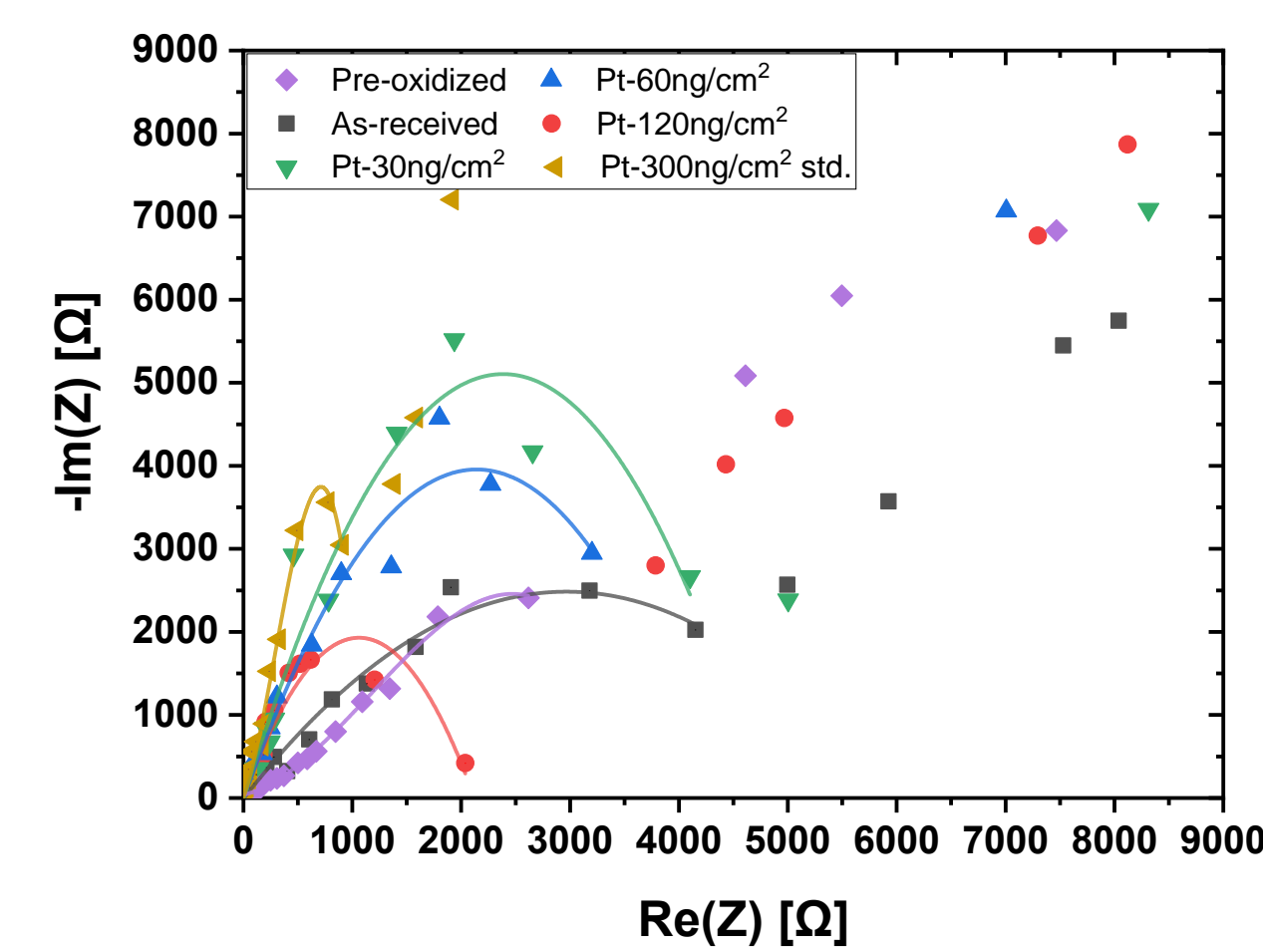


WE = Working electrode (specimen)  
RE = Reference electrode (Hg/HgSO<sub>4</sub>)  
CE = Counter electrode (Pt foil)

### Specimen-oxide-Pt configurations

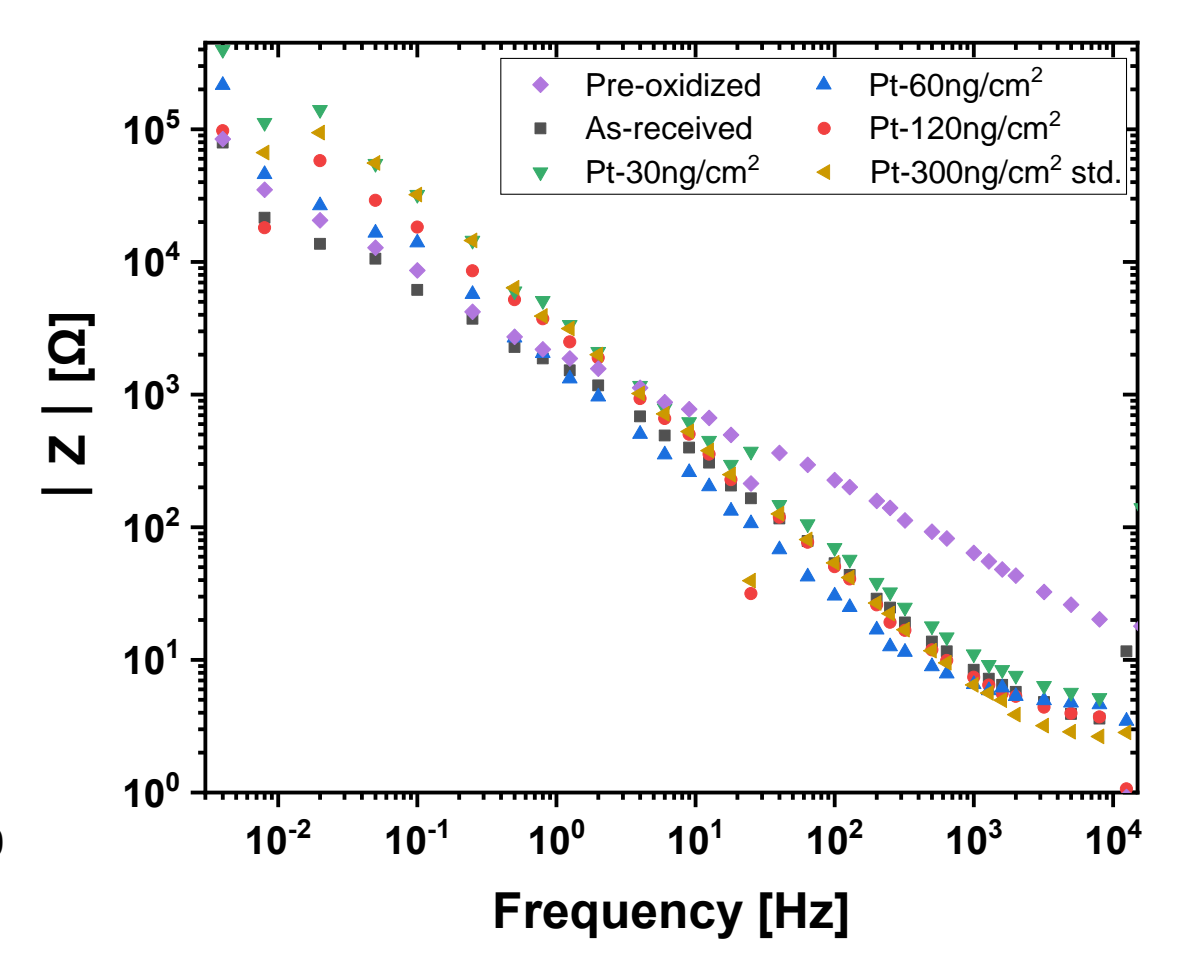


### Nyquist Plots



Re(Z): Pre-oxidized ≈ As-received > Pt-30 > Pt-60 > Pt-120 > Pt-300 (std.)

### Bode plots



✓ Qualitative differences are showcased for varying Pt amount and type (particles/coating) in these preliminary measurements

## 4 Takeaways and Outlook

- ✓ Mean interparticle distances are better predictors of ECP values compared to Pt loadings → Predicted a max. 1000 nm Pt interparticle distance for "protective" ECP of < -230 mV<sub>SHE</sub>
- ✓ EIS method seems promising, revealing qualitative differences for varying amounts of Pt quantities and distributions → Further investigations ongoing to confirm results